Dynamics of methane ebullition from a peat monolith revealed

from a dynamic flux chamber system

By

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ABSTRACT OF THE DISSERTATION

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Methane (CH₄) ebullition in northern peatlands is poorly quantified in part due to its high spatiotemporal variability. In this study, we employed a continuous measurement approach to better understand the coupling between CH₄ ebullition fluxes and subsurface gas bubble dynamics and to examine potential triggering effects of atmospheric pressure and temperature on ebullitive CH₄ releases. A dynamic flux chamber system (DFC), consisting of a flow-through gas chamber and a fast methane analyzer (LI-COR 7700), was used to continuously measure CH₄ fluxes from a monolith of near-surface *Sphagnum* peat over eight weeks. By using a graphical separation method, episodic ebullition fluxes were extracted from total flux recorded, and the timing and CH₄ content of individual ebullition events identified. Coincident transmission ground penetrating radar (GPR) measurements of gas content and dissolved CH₄ concentrations in pore water were also acquired at three depths (upper, middle, and lower) within the monolith. Estimated episodic ebullition fluxes were not sensitive to the uncertainties in steady flux

quantification associated with the graphical model and the application of the DFC had minimal disturbance on air-peat CH₄ exchange. Episodic and steady ebullition fluxes, constrained by modeled diffusion fluxes using Fick's law and the bulk CH₄ concentrations in peat, were estimated as on average 38% and 36% of the total fluxes over the entire study period, respectively. The observations of gas content variations within the three layers along with the timing of episodic ebullition fluxes support the existence of an ebullition threshold regulating CH₄ ebullition. However, a larger threshold (gas content of $0.14 \text{ m}^3 \text{ m}^{-3}$) was found for the middle and lower layers, suggesting that multiple mechanisms related to the depth variation of peat structure were responsible for the complex behavior of episodic CH₄ ebullition. Temperature variation (23 °C to 27 °C) was likely only responsible for small episodic ebullition events from the upper peat layer, while large ebullition events from the deeper layers were most likely driven by drops in atmospheric pressure.

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1. Introduction

Northern peatlands are a global methane (CH₄) source, accounting for 12.2% of global CH₄ emissions to the atmosphere, and a net sink for atmospheric carbon dioxide [Wuebbles and Hayhoe, 2002]. Carbon cycling in these ecosystems is controlled by the interactions among microbiological [Turetsky et al., 2008], hydrogeological [Slater and Reeve, 2002], plant ecological [Riutta et al., 2007], and climatological processes [Bragazza et al., 2012], which all have the potential to affect CH₄ production, oxidation and transport dynamics [Lai et al., 2009]. During the last two decades there has been a growing interest in biogenically produced gas bubbles in northern peatlands. The presence of these bubbles in peat below the water table impacts the buoyancy of floating peat [Fechner-Levy and Hemond, 1996], results in the development of over-pressured stratum [Rosenberry et al., 2003], decreases hydraulic conductivity and substrate delivery [Kellner et al., 2005], and contributes to CH₄ release from peatlands via rapid bubbling (ebullition) [Glaser et al., 2004]. Compared to other CH₄ transport pathways in peatlands, i.e., diffusion of dissolved CH₄ and venting of CH₄ through aerenchyma of plants, ebullition of CH₄-containing bubbles from the subsurface might release larger amounts of CH₄ and exhibit much greater spatiotemporal variations [Christensen et al., 2003]. This variability means that ebullition as a component of the total CH₄ emissions from northern peatlands remains poorly quantified.

Theoretically, a bubble is formed in a fully saturated peat when the total pressure of the water-dissolved gases exceeds the sum of hydrostatic pressure imposed by the water table and atmospheric pressure [Strack et al., 2005]. Once formed in peat, biogenic bubbles can

be trapped under confining layers that consist of peat with pore diameters small enough to block the passage of bubbles [Rosenberry et al. 2003; Glaser et al., 2004]. Bubbles migrate upward to the water table once buoyancy forces surpass the trapping forces at the bubble-peat interface [Kellner et al., 2005]. Table 1 is a compilation of recent findings on CH₄ ebullition in peatlands obtained from various methods, showing that ebullition fluxes can vary over three or four orders of magnitude even under laboratory conditions, suggesting a variety of controlling factors (Table 1).

gas content	dissolved CH ₄ (mg L ⁻¹)	ebullition flux (mg m ⁻² d ⁻¹)	controls on ebullition flux ^e	reference
0.01 - 0.15	0.1 - 11.7	2 - 83	bubble threshold	Baird et al. [2004]
		0 -196	AT	Beckmann et al. [2004]
		45 - 225	surface vegetation	Christensen et al. [2004]
		3 - 68	AP	Comas and Wright [2012]
	0.1 - 1.3 ^b	0.2 - 784	surface vegetation	Green et al. [2012]
	1.9 - 6.0	<10	AP	Green et al. [2013]
		49 - 152ª	CH ₄ production, wind speed	Goodrich [2012]
0.02 - 0.16	11.7 - 13.9	270	Bubble threshold, AP, AT	Kellner et al. [2006]
		7 - 12	none	Stamp et al. [2013]
0.05 - 0.15 ^c	0.1 - 5.8	65	AP, water table	Strack et al. [2005]
	~ 10.5 ^d	76 - 1233	AP	Tokida et al. [2005]
		48 - 1440	AP	Tokida et al. [2007]
0.03 - 0.15	1.0 -11.6	1471 - 33093	AP, AT	Waddington et al. [2009]
0.04 - 0.15	0.1 - 11.2	0 - 237	AP, AT	this study

Table 1 Gas content, dissolved CH₄, ebullition flux and its controls from peatland studies

a: flux was obtained by multiplying seasonally-averaged ebullition event flux to mean

CH₄ concentration for each event.

b: estimated from Figure 4 in the original paper.

c: change in volumetric gas content relative to the beginning point.

d: depth-averaged value, estimated by assuming partial pressure representing 45% by

volume of CH₄ in bubbles.

e: AP and AT denote atmospheric pressure and atmospheric temperature, respectively.

One potential reason for the episodic nature of CH_4 ebullition is that a large bubblestorage capacity of peat may result in a partial decoupling between biological CH₄ production and bubble dynamics in subsurface peat [Strack and Waddington, 2008], such that physical factors can determine and/or alter the mechanical balance of the bubble-peat interaction, triggering ebullition [Tokida et al., 2009]. On the one hand, the physical properties of peat, such as degree of decomposition [Kettridge and Binley, 2011] and compressibility [Price et al., 2005], may set boundaries on the bubble trapping capacity and, subsequently, ebullition potential for peat. Results from laboratory incubations of near-surface Sphagnum peat suggest that there is a threshold gas content of $10 \sim 16\%$ by volume, depending on the effective compressibility of peat, that must be reached in the peat matrix before the main period of ebullition [Baird et al., 2004; Kellner et al., 2006]. Changes in environmental factors, particularly atmospheric pressure and temperature (Table 1), may trigger ebullition through their direct impact on bubble volume as described by the Ideal Gas Law and Henry's Law [see Tokida et al. [2009] for a review]. Indeed, when the biological controls are steady state, CH₄ ebullition in northern peatlands can often be thought as a net consequence of mechanical disequilibrium associated with coupled below- and aboveground hydro-climatological changes.

Closed static chambers (CSC) and gas traps have been widely used to quantify ebullition fluxes in peatlands [e.g., Tokida et al., 2005; Stamp et al., 2013]. Stamp et al. [2013] reported a skewed spatial pattern of CH₄ ebullition in a raised bog based on 28 gas traps

with only 9 accounting for \sim 76% of the total seasonal flux. The large uncertainties associated with CH₄ ebullition estimates may in part result from a lack of suitable monitoring methods [Mastepanov and Christensen, 2009]. The CSC approach suffers from low temporal resolution, while gas traps only provide cumulative bubble volume trapped at the water table and thus indirect ebullition flux. In addition, neither of these two methods are able to clarify how individual ebullition events result from triggering factors over a long time scale. More recently, Goodrich et al. [2012], using automatic flux chambers with a high sampling frequency, observed a prominent diel variation of CH₄ ebullition in a temperate peatland over three seasons. This study highlighted the need for continuous flux measurements in order to obtain accurate ebullition estimates. Furthermore, although an ebullition threshold can be used to reasonably predict bubble release by volume in the laboratory [Kellner et al., 2006], it provides little insight into how physical characteristics of peat control the relationship between bubble accumulation and ebullition fluxes [Coulthard et al., 2009]. Continuous flux measurements are thus of required to gain better process-based understanding of the complex behavior of CH₄ ebullition.

As a step towards better understanding of the importance of CH₄ ebullition fluxes, we developed a dynamic flux chamber (DFC) for the continuous monitoring of CH₄ ebullition from a near-surface peat monolith under laboratory conditions. A fast methane analyzer (FMA) providing high-precision and fast CH₄ concentration measurements was incorporated into the dynamic system for the quantification of both episodic CH₄ releases and integrated daily CH₄ ebullition estimates. Ground penetrating radar (GPR), a non-

invasive geophysical method for estimating gas content in peat [Comas and Slater, 2007] was also used, dissolved CH₄ concentrations in pore-water were measured. We first demonstrate the capacity of the set up and then use it to examine the dependence of CH₄ ebullition fluxes on subsurface bubble dynamics leading to insight into the underlying mechanisms triggering CH₄ ebullition in northern peatlands.

2. Methods

2.1. Sample collection and preparation

A peat monolith (c. $394 \times 241 \times 330$ mm, surface area 0.095 m²) was extracted on November, 2012 from the surface of a lawn dominated by *Sphagnum* and low shrub in Caribou Bog, a large, multiunit freshwater bog complex in Maine. Vegetation on the surface of the monolith was left intact and consisted mainly of *Sphagnum* and ericaceous shrubs. The monolith was first removed from the field by inserting a mold of the same size ($394 \times 241 \times 350$ mm) as the container used to hold the monolith (see below) into the ground and subsequently cutting and pulling back the surrounding peat, and cutting the base of the mold with a machete. The monolith was then placed into a temporary holder and transported back to the laboratory.

The monolith was pushed out from the mold into container made of 10 mm-thick transparent acrylic plastic. Three pore-water sampling ports were drilled vertically into one side of the container at 57 mm, 127 mm, 197 mm below the peat surface, representing an upper, middle, and lower layer of the peat monolith (Fig. 1). The exterior walls of the container were entirely covered by black tapes to prevent penetration of light and heat. This approach resulted in a tight interface between the peat monolith and the edge of the container, reducing the possibility of preferential transfer of gas bubbles up the container walls. Three pore-water samplers constructed from unglazed pipe (1.58 mm inner diameter) and fitted with a three-way stopcock were horizontally inserted into the ports. Distilled water was sprayed onto the peat surface to maintain saturated anaerobic conditions in the peat. The peat monolith was maintained at room temperature (c. $22 \sim 24$ °C).

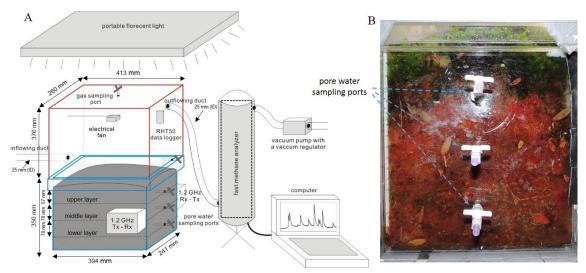


Figure 1 (A) dynamic flux chamber setup for continuous monitoring of CH₄ ebullition in the laboratory; (B) side view showing stratum characteristics for the three layers.

2.2. DFC setup and flux calculation

The DFC is a technique that has been developed to continuously measure earthatmosphere fluxes of various compounds including CO₂ and CH₄ [Lai et al., 2012]. In contrast to the CSC technique, the DFC is designed to maintain a constant flow of outside air through the chamber enclosing the areas of interest. When the system operates under a dynamic equilibrium, the CH₄ flux (F, mg m⁻² min⁻¹) from the monolith is associated with the concentration difference between the air entering (C_{in} , mg m⁻³) and leaving (C_{out} , mg m⁻³) the chamber headspace [Pape et al., 2009],

$$F = \frac{Q}{A} \times (C_{out} - C_{in}), \qquad \text{Eq. 1}$$

where Q and A are the flow rate of purging air (m³ min⁻¹) and surface area (m²) of the peat monolith, respectively.

A DFC requires careful system design to achieve minimal modification of the soil-air CH₄ exchange regime. In this study, we modified the DFC system described in Mastepanov and Christensen [2009] by employing a LI-7700 open path fast methane analyzer (FMA, LI-COR Biosciences, Lincoln, NE, USA) to determine CH4 concentration. This DFC system consisted of a gas chamber ($413 \times 260 \times 325$ mm) that can be fitted to the water-filled collar on the top of the peat container for gas flux measurement (Fig. 1A). The chamber was made of 10 mm-thick transparent acrylic plastic with a headspace volume of 0.035 m³. Inflow and outflow air ducts (25 mm inner diameter and 0.5 m length), a gas sampling port and a standpipe for regulating the water table of peat monolith was installed (Fig. 1A). The inflow duct was located 0.09 m above the peat surface, with the outflow duct 0.32 m above the peat surface on the opposite side of chamber (Fig. 1A). A steady purging flow of 0.0047 m³ min⁻¹ was maintained by a vacuum pump with a vacuum regulator (Fig. 1A). This flow rate resulted in a mean residence time of air in the chamber of 7.5 minutes, being in the middle of the range of residence times reported in the literature [Table 4 in Pape et al., 2009] and considered a good compromise between flux detection limit and time response of the system. Preliminary tests confirmed that this setup caused no significant pressure difference between the chamber and the laboratory (data not shown for brevity). Turbulent mixing inside the gas chamber was encouraged using an electric fan blowing downwards (Fig. 1A). The approach promoted mixing of the air in the chamber headspace such that the

CH₄ could be assumed uniform throughout the chamber headspace. The continuous measurement of C_{out} then provides information on CH₄ dynamics in the chamber [Pape et al., 2009].

The LI-7700 open path FMA is a high-speed, high-precision, wavelength modulation spectroscopy-based CH₄ analyzer [McDermitt et al., 2011]. Its root mean square noise is 5 ppb for a 10 Hz measurement at typical ambient CH₄ concentrations (~2 ppm). In this study, we modified the data acquisition mode of the LI-7700 FMA by enclosing it within the calibration shroud provided by the manufacturer, resulting in a closed inner cell (V_{FMA} , 0.009 m³) as the terminal space for CH₄ analysis. The FMA was positioned in series with the gas chamber and the vacuum pump (Fig. 1a). After leaving the gas chamber, the purging air flows into the inner cell via the bottom fitting and then leaves through the top fitting at the same flow rate as applied to the gas chamber. Since the volume of the inner cell is much smaller than the gas chamber, complete mixing is insured in the inner cell space. By assuming steady state condition in the inner cell, a CH₄ mass balance can relate C_{out} to CH₄ concentrations recorded by the FMA in the inner cell, a CH₄

$$(C_{out} - C_i) \times V_{out} = (C_{i+1} - C_i) \times V_{FMA}$$
Eq. 2

where C_i and C_{i+1} are CH₄ concentrations in the inner cell (mg m⁻³) recorded at time moment *i* and *i*+1, respectively, and V_{out} is the volume of outflowing air entering the inner cell over the time period from *i* to *i*+1 (m³). Consequently, C_{out} can be inversely calculated from the FMA time series of inner cell concentration collected at 5 Hz. Uncertainties in calculated *C*_{out} and CH₄ fluxes were estimated using Gaussian error propagation [Oswald et al., 2013],

 ΔC_{out}

$$= \sqrt{\left[\left(\frac{\partial C_{out}}{\partial V_{FMA}}\right)_{V_{out},C_{i+1},C_{i}} \cdot \Delta V_{FMA}\right]^{2} + \left[\left(\frac{\partial C_{out}}{\partial V_{out}}\right)_{V_{FMA},C_{i+1},C_{i}} \cdot \Delta V_{AF}\right]^{2} + 2 \cdot \left[\left(\frac{\partial C_{out}}{\partial C_{i+1}}\right)_{V_{out},V_{AF},C_{i}} \cdot \Delta C_{i+1}\right]^{2}}$$
Eq. 3

$$\Delta F = \sqrt{\left[\left(\frac{\partial F}{\partial Q}\right)_{A,C_{out},C_{in}} \cdot \Delta Q\right]^{2} + \left[\left(\frac{\partial F}{\partial A}\right)_{Q,C_{out},C_{in}} \cdot \Delta A\right]^{2} + 2 \cdot \left[\left(\frac{\partial F}{\partial C_{out}}\right)_{Q,A,C_{i}} \cdot \Delta C_{out}\right]^{2}}$$
Eq. 4

The error in bulk CH₄ concentrations (ΔC_{i+I}) was set to the noise of the instrument under ambient CH₄ concentrations [Oswald et al., 2013] measured in the laboratory. The errors for the purging air flow rate (ΔQ) and outflowing air volume (ΔV_{out}) were estimated from the standard deviation of the measured flow rate using a gas flow meter. The errors for the soil surface area (ΔA) and volume of the inner cell (ΔV_{FMA}) were determined from a length measurement.

2.3. CH₄ concentration corrections

The determination of C_{out} by the FMA is affected by variations of temperature and water vapor via thermal expansion and water vapor dilution; in addition, temperature, pressure, and water vapor density in the inner cell impact concentration measurement by shifting the spectroscopic light absorption of the FMA [McDermitt et al., 2011]. In order to take these effects into account, one RHT50 data logger (Extech Instruments, USA) was installed right beside the outflowing duct in the gas chamber to measure temperature, atmospheric pressure, and relative humidity of the chamber headspace every 10 minutes. Another identical RHT50 data logger was installed in the laboratory to measure ambient

conditions for comparison. As the FMA was connected to the gas chamber for more than five days per experiment (see below), it is reasonable to assume that the inner cell quickly reached a dynamic equilibrium with the gas chamber in terms of water vapor density. The raw bulk CH₄ concentration data were corrected by using 10-minute averaged water vapor densities converted from the relative humidity of the chamber headspace, and FMA measured temperature and pressure according to the Webb-Pearman-Leuning (WPL) corrections [Webb et al., 1980] and spectroscopic correction [McDermitt et al., 2011] prior to C_{out} inversion. Figure 2A summarizes the sequence of the data correction and calculation steps described above, and Fig. 2B shows a typical flux dataset obtained from the DFC system.

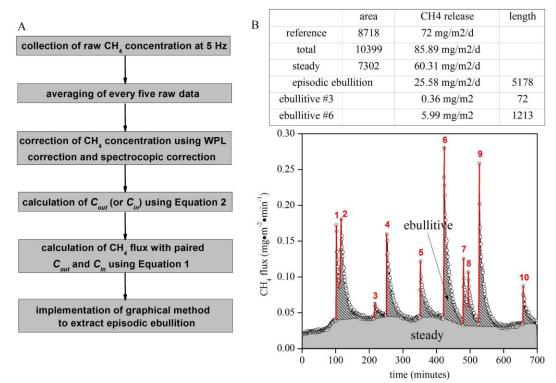


Figure 2 (A) flowchart of procedure for data correction, calculation, and extraction. (B) an example of the dynamics of CH₄ flux observed using DFC. Steady flux (gray areas)

and episodic ebullition flux (shaded area) were separated and identified using a graphical model. Ten ebullitive CH₄ releases were identified (straight red lines). Calculations of ebullition flux and ebullitive CH₄ release are shown in the table.

2.4. Flux partitioning and ebullition extraction

Based on distinctions in temporal continuity, diffusion and plant-mediated transport are usually classified as steady emission pathways, while ebullition is conventionally considered as random in time. However, recent studies suggest that in peatlands CH₄ ebullition can also occur in a steady manner over time frames of minutes to hours [Coulthard et al., 2009]. In some studies, diffusion, plant-mediated transport, and steady ebullition are collectively defined as steady emission, while ebullition occurring in shortlived bursts or episodes where fluxes are generally much higher and more variable than background steady fluxes are termed episodic ebullition [Green and Baird, 2012]. In this study, we follow such definitions to partition measured CH₄ fluxes.

Using the DFC system, the frequency of episodic ebullition during a specific time period can be determined by visual counting of flux peaks in flux time series (Fig. 2B). Ideally, episodic ebullition fluxes can also be accurately separated out from steady fluxes as outliers in the flux dynamics. However, such situation requires a gas residence time of DFC as short as that necessary for a released gas bubble to disseminate its content throughout the chamber headspace, which is not feasible in practice with the consideration of achieving a minimal perturbation of pressure regime in the headspace. Since some accumulation of ebullition-derived CH_4 in the gas chamber is unavoidable,

the calculated fluxes from the DFC system are apparent fluxes rather than instantaneous CH₄ emission rates from the peat monolith. In this sense, episodic ebullition fluxes are best isolated from the baseline steady fluxes by integrating areas under ebullition flux peaks [Panikov et al., 2007] (Fig. 2B).

Methods exist to statistically identify a baseline in a time series and quantitatively integrate area under peaks, e.g. chromatographic algorithms. However, these methods may only be applicable in cases where the baseline is stable [Panikov et al., 2007]. By using a similar flow-through system, Christensen et al. [2003] defined steady baselines visually with a fixed flux rate for individual datasets. In this study, we adopted a simple graphical model that is analogous to the hydrological approach for extracting peak flows in stream hydrographs. We defined the baselines by tracking the smooth variation in flux until first arrival of a peak and identifying the point at which the peak returned to a steady state, such that the baseline was set as a straight line under the peak (Fig. 2B). Calculated CH₄ fluxes were first plotted in OriginPro 8.1 (OriginLab Corporation, USA) and normalized to a standard axis format. The flux graphs were then imported into ImageJ for calculation of areas under peaks. Polygons were drawn to define areas under the baseline along with a reference rectangle of known flux in the standardized graphs, respectively, and their areas were automatically calculated (see inserted table in Fig. 2B). Steady flux can then be quantified by comparing its integrated area to the reference. The difference between total flux and steady flux gives the integrative episodic ebullition flux over the time period of interest. The accuracy of the estimated episodic ebullition flux was assessed by analyzing the potential error in the steady flux quantification.

Similar to other flow-through systems (e,g,, inline membrane probes for *in situ* measurement of pore water chemistry [Mastepanov and Christensen, 2009]), the difference between peak height and baseline level is a good approximation of CH₄ content in individual episodic events. Once integrated episodic ebullition fluxes were extracted from the time series, we graphically measured the peak height from the starting point of a peak, which may be on the baseline or on the declining slope of the immediately prior peak when two peaks occur closely in time (e.g., later phase in Fig. 2B), to its highest point in the standardized graphs (Fig. 2B). Individual peak heights were then compared to total lengths over a certain time period to calculate ebullitive CH₄ release for each episodic ebullition events. All original flux data, showing the defined baseline and identified peak heights are provided in the appendix.

2.5. CSC measurements

The gas chamber of the DFC system was modified for CSC measurements of steady fluxes by closing the inflow and outflow holes with fittings and extending the water resupply tube to be a vent tube for pressure balance. For each CSC measurement, a 30 minutes chamber closure time was adopted; seven 30 mL gas samples were taken from the top sampling port at a 5 minute time interval after chamber closure and injected into pre-vacuumed 20 mL vials for gas chromatograph (Shimadzu, GC-14A, Japan) analysis. Linear regressions were performed on each set of CH₄ concentration series to calculate flux. The CSC flux measurement was accepted if the determination coefficient of linear regression (R^2) is larger than 0.90.

2.6. GPR measurements

The theory and methods employed to use GPR to characterize gas bubble dynamics in peat soils have been documented elsewhere [e.g., Comas et al., 2007; Comas et al., 2008]. In this work, we modified a laboratory GPR system and data acquisition protocol initially used by Comas and Slater [2007], improving the accuracy of gas content estimates by replacing reflection-based measurements with transmission measurements.

A Mala-RAMAC GPR system equipped with a pair of 1.2-GHz shielded antenna was used to compute the velocity of EM waves in the upper, middle, and lower layers of the peat monolith (three lines as shown in Fig. 1a). Three horizontal positions spaced 0.1 m apart were used for a GPR measurement in each layer. EM waves were transmitted across the monolith and direct wave travel times, complete with reciprocal data for each location, recorded. The velocity (v) of the EM wave in peat is calculated using the known distance that the EM wave traveled (i.e., across the length of the peat container). Assuming a low-loss medium [Comas and Slater, 2007; Parsekian et al., 2012], v can then be simply converted to the bulk relative permittivity of the peat ($\varepsilon_{r(b)}$),

$$\varepsilon_{r(b)} = \left(\frac{c}{v}\right)^2,$$
 Eq. 5

where *c* is the EM wave velocity in a vacuum $(3 \times 10^8 \text{ m s}^{-1})$.

The strong contrast in relative permittivity between air ($\varepsilon_{r(a)} = 1$) and water ($\varepsilon_{r(w)} = 79$ at laboratory temperature of 23 °C) forms the basis for using measurements of $\varepsilon_{r(b)}$ to determine gas content in peat soils. A multiphase dielectric mixing model variant of the

Complex Refractive Index Model (CRIM) was applied to estimate gas content [Parsekian et al., 2012],

$$\varepsilon_{r(b)}{}^{\alpha} = \theta \varepsilon_{r(w)}{}^{\alpha} + (1 - \emptyset) \varepsilon_{r(s)}{}^{\alpha} + (\emptyset - \theta) \varepsilon_{r(g)}{}^{\alpha}, \qquad \text{Eq. 5}$$

where $\varepsilon_{r(s)}$ is relative permittivity of the solid phase assumed here as 2 [Comas and Slater, 2007], θ is the water content, ϕ is porosity, and α is an empirical coefficient related to the orientation of the electromagnetic waves to the peat particles. The term $\phi - \theta$ is the volumetric gas content of the peat. In this study, we set the value of $\alpha = 0.33$ as previously determined empirically for Caribou bog peat samples [Parsekian et al., 2012]. We assume that porosity did not change throughout the experiment due to container wall preventing horizontal peat deformation [Comas and Slater, 2007]. Thus, the porosity measured at the end of the incubation (see below) was used in Eq. 5 to calculate gas content for the entire study.

2.7. Dissolved CH₄ concentration measurements

To measure the dissolved CH₄ concentrations in pore-water, 2 mL of pore-water was collected by syringe from the samplers in the upper, middle, and lower layers respectively and injected into an 8 mL vial filled with ambient air. The vials were then shaken vigorously for 15 minutes to equilibrate CH₄ between the headspace and water phase. The CH₄ concentration in the headspace of the vial and two replicates of ambient air were determined by GC, and the original dissolved concentration was reconstructed using the headspace concentrations, the ambient CH₄ concentration, the volumes of the headspace and water phase, and temperature-corrected Bunsen solubility coefficient [Wiesenburg and Guinasso, 1979].

2.8. Modeling CH₄ diffusion fluxes from the peat

CH₄ diffusion fluxes across the soil surface can be theoretically calculated from the CH₄ concentration gradient within the soil using Fick's first law of diffusion [Reid et al., 2013]. In flooded peat, CH₄ diffusion occurs in both air-filled and water-filled pore space such that the presence of gas bubbles causes CH₄ diffusion fluxes much higher than modeled results that only consider dissolved CH₄ transport in pore water [Stephen et al., 1998a]. In this study, CH₄ diffusion fluxes from the peat monolith were modeled using Fick's first law with model coefficients specifically derived for flooded soil containing gas bubbles (Eq. 6),

$$F = D_e \frac{\partial C_b}{\partial z} \approx D_e \frac{C_b}{\partial z}, \qquad \text{Eq. 6}$$

$$D_e \approx \left[\frac{(\epsilon+\theta)\sqrt{D_a D_w}}{\epsilon\sqrt{D_w}+\theta\sqrt{D_a}}\right]^2$$
, Eq. 7

$$C_b = \epsilon C_a + \theta C_w = \left(\frac{\epsilon}{\alpha_b} + \theta\right) C_w$$
, Eq. 8

where D_e is the effective bulk diffusion coefficient and can be expressed as a function of the CH₄ diffusion coefficients in free air (D_a) and water (D_w), gas content (ϵ), and water content (θ) of the peat medium (Eq. 7) [Nielson et al., 1984; Stephen et al., 1998a]. The term $\partial C_b / \partial z$ is the gradient of bulk CH₄ concentration over depth *z* to the water table. By assuming that equilibrium in the gas-pore water system can be reached rapidly and obeys Henry's law, C_b can be calculated using measured ϵ , θ , dissolved CH₄ concentration (C_w), and Bunsen solubility coefficient for CH₄ (α_b) according to Eq. 8 [Stephen et al., 1998a]. In the calculations using Eq. 6 a simplification was made because C_b in peat is much higher than the equilibrated CH₄ concentration at water table [Reid et al., 2013]. All diffusion and solubility coefficients used in this study were corrected for temperature. An important boundary condition on Fick's first law is that the concentration gradient in soil has to be taken in the direction where it is steepest [Rothfuss and Conard, 1998]. Gas bubbles affect CH₄ concentration gradients in flooded soils and thus CH₄ diffusion modeling by (1) resulting in a three-dimensional uneven structure of CH₄ gradients in which the concentration isopeths follow the surface of trapped gas bubbles [Rothfuss and Conard, 1998] and (2) increasing dissolved CH₄ concentrations in shallower peat via redissolution during their migration toward the surface [Tang et al., 2010]. As a result, the distribution of CH₄ in flooded soil profile is highly variable even in layers right below the water table [Panikov et al., 2007], and any choice of depth in the calculation of concentration gradient will be arbitrary. For simplicity, we used average values of the three layers and set the lower boundary of the concentration gradient at the middle layer to model CH₄ diffusion flux for each experimental day.

2.9. Experimental procedure

As our goal was to understand how CH₄ ebullition depends on the evolution of biogenic bubbles in peat and how atmospheric pressure and temperature influence ebullition, we conducted the experiment, such that the whole experiment system was subject to natural fluctuations of atmospheric pressure and temperature of the laboratory. A fluorescent light was set up above the gas chamber for simulating a 12 hour day/night cycle with a daytime light intensity of 4350 Lux at the peat surface and providing additional heating that enhanced the amplitude of temperature variation in the gas chamber (Fig. 1A). Two days before the incubation, the peat monolith was drained by unscrewing three porewater samplers and gently tilting the peat container to pour the water out. We then slowly rewetted the peat monolith from the bottom pore-water sampling port using distilled water until the water table was 2 cm below the peat surface. The rewetting encouraged a low initial CH₄ level and gas content in peat. One set of GPR measurements, one round of pore water sampling, and three sets of CSC flux measurements were taken immediately after the rewetting to establish pre-incubated conditions. During the entire study period, the water table was maintained at its initial level by replenishing with distilled water through the water filling tube.

Eight consecutive weeks of CH₄ ebullition monitoring began on March 26, 2013. For every experimental week, the DFC system was started 12 hours before the continuous flux monitoring in order to establish dynamic equilibrium. Details of the data acquisition procedure for every experimental week are summarized in Table 2. Importantly, C_{out} was continuously measured for 120 hours every experimental week under continuous chamber closure, and measured C_{in} values were averaged to one 24-hour dataset and then used to calculate flux for every experimental day. After eight weeks of flux monitoring, three replicate samples were extracted from the upper, middle, and lower layers of the monolith respectively by cutting from the surface. Then, the humification degree, stratum characteristics, bulk density, and porosity were measured using a weight loss on dying procedure previously used by Comas and Slater [2007].

Day 1 – Day 5	Day 6	Day 7
 <i>C</i>_{out} from the gas chamber was continuously measured under 5 day/night cycles (120 h) GPR and pore-water samplings at three layers were conducted twice a day during daytime 	FMA was disconnected from the gas chamber to measure <i>C_{in}</i> for 24 h	three sets of CSC flux measurement were taken during daytime

Table 2 Data acquisition procedure for every experimental week

2.10. Data statistics

The corrections and calculations of all CH₄ fluxes were performed in MATLAB (Mathworks, USA), and all data statistical tests were performed using SPSS (SPSS Inc., USA). A one-way ANOVA followed by Fisher's least significant difference test for pairwise comparison was used to determine significant differences between independent variables. Simple linear regression and Pearson's Correlation Coefficient were used to detect significant relationships among independent variables.

3. Results

3.1. Peat properties

The von Post humification test indicated that the upper layer of the monolith was undecomposed (H2), while the middle and lower layers were slightly decomposed (H3-H4). Below the upper layer, *Sphagnum* accumulated with slightly decomposed branches, stems, and variable amounts of woody material (Fig. 1B). Bulk density was found to be 0.052 g cm⁻³, 0.065 g cm⁻³, and 0.078 g cm⁻³ for the upper, middle, and lower layers, respectively. Porosity for the upper, middle, and lower layers were 0.96, 0.95, and 0.94, respectively.

3.2. Incubation conditions

The temperature in the gas chamber exhibited significant diel patterns (Fig. 3). Averaged 10-minute temperatures were 23 °C to 24 °C during nighttime, increasing to 26 °C to 27 °C when the florescent light was on. Regardless of the diel shifting, temperatures in daytime and nighttime were stable. While relative humidity exhibited a reverse diel pattern to temperature, relative humidity was always greater than 95% in the gas chamber. During the entire experiment, the peat monolith experienced a range of atmospheric pressure from 100.18 kPa to 103.47 kPa. The averaged CH₄ concentration in ambient air (*C_{in}*) ranged from 1.31 mg m⁻³ to 1.46 mg m⁻³ (Fig. 3). While daytime *C_{in}* was relatively stable and consistent among the experimental weeks, nighttime *C_{in}* varied more, especially in the later period, with a maximum standard deviation of 0.24 mg m⁻³.

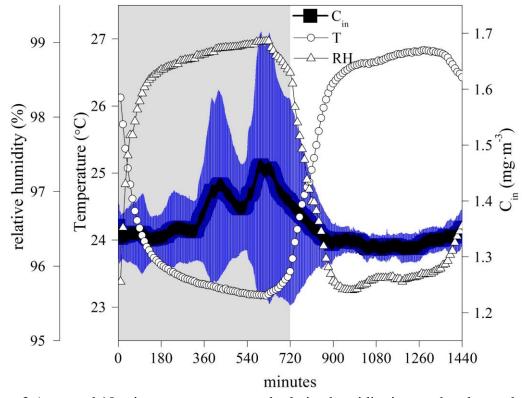


Figure 3 Averaged 10-minute temperature and relative humidity in gas chamber and averaged C_{in} for every minute. Standard deviation of C_{in} is shown by blue line. Shaded area denotes nighttime.

3.3. Gas content and dissolved CH₄ concentrations

Pre-incubation gas content for the upper, middle, and lower layers was 0.04 m³ m⁻³, 0.06 m³ m⁻³, and 0.07 m³ m⁻³, respectively. Gas content of middle and lower layers, ranging from 0.06 to 0.14 m³ m⁻³ and 0.08 to 0.15 m³ m⁻³, respectively, generally increased from the start of the experiment with a higher rate of increase in week 1 to week 5 (Fig. 4A). The gas content of the upper layer, ranging from 0.04 to 0.07 m³ m⁻³, showed a lower rate of increase throughout the experiment (Fig. 4A). The difference test indicated that gas content of the lower layer was significantly higher than the middle layer and the upper layer (*P*<0.01 in all comparisons).

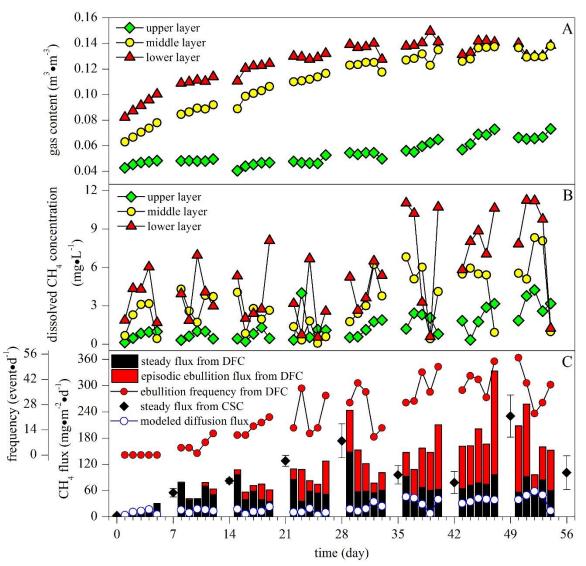


Figure 4 Gas content (A), dissolved CH₄ concentrations (B), steady and ebullition fluxes, ebullition frequency, and modeled diffusion fluxes (C) for each day of the experiment.

Pre-incubation dissolved CH₄ concentrations were 0.075 mg L⁻¹, 0.003 mg L⁻¹, and 0.007 mg L⁻¹, respectively, for the upper, middle, and lower layers. Dissolved CH₄ concentrations of the middle and lower layers increased rapidly after the experiment started but exhibited strong variability throughout the experiment with the largest concentration reduction (8.5 mg L⁻¹) in the lower layer (Fig. 4B). In contrast, dissolved

CH₄ concentrations of the upper layer increased slowly and smoothly (Fig. 4B). Despite the greater fluctuation in concentration, the lower layer of the monolith had significantly higher dissolved CH₄ concentrations (0.54 to 11.25 mg L⁻¹) than the middle layer (0.07 to 8.33 mg L⁻¹) and the upper layer (0.13 to 4.24 mg L⁻¹) (P<0.01 in all comparisons). A positive correlation between dissolved CH₄ concentration and gas content was detected in the middle layer (R²=0.40, P<0.05) and upper layer (R²=0.65, P<0.01).

3.4. CH₄ fluxes measured using CSC

Twenty four out of 27 CSC flux measurements met the linearity criteria with R²>0.90 (Fig. A9). Pre-incubation CSC CH₄ flux was 2.67 \pm 0.39 mg m⁻² d⁻¹ (Fig. 4C). Throughout the experiment, CSC CH₄ fluxes ranged from 50.93 \pm 8.68 mg m⁻² d⁻¹ to 212.16 \pm 44.28 mg m⁻² d⁻¹ with the highest flux observed in week 7 (Fig. 4C).

3.5. Modeled CH₄ diffusion fluxes

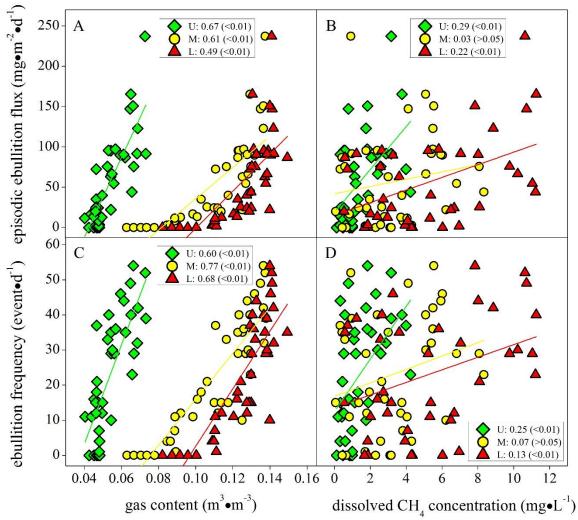
Modeled CH₄ diffusion fluxes from the peat monolith ranged from 3.48 mg m⁻² d⁻¹ to 53.43 mg m⁻² d⁻¹ (Fig. 4C). Modeled CH₄ diffusion fluxes had significant positive correlations with all measured gas content and dissolved CH₄ concentrations of the three layers (R^2 ranging from 0.50 to 0.91; *P*<0.01 for all cases).

3.6. CH₄ fluxes measured using DFC

CH₄ fluxes measured by DFC and quantifications of steady fluxes and episodic ebullition fluxes are shown in Fig. A1-A8. The average uncertainty in DFC flux for the entire experiment was $\pm 2.85\%$. Steady fluxes (Fig. 4C), ranging from 3.84 mg m⁻² d⁻¹ to 148.05

mg m⁻² d⁻¹, gradually increased in week 1 and then did not differ significantly after week 1 when compared on a weekly basis (P>0.05 for all comparisons). A difference test indicates that the steady fluxes from DFC were significantly higher than modeled CH₄ diffusion fluxes (P < 0.01). Episodic ebullition did not occur until the first day of week 2. Episodic ebullition fluxes, ranging from 1.84 mg m⁻² d⁻¹ to 237.31 mg m⁻² d⁻¹, were then recorded, with fluxes generally increasing towards the end of the experiment. The highest daily ebullition flux occurred in the last day of week 7. Episodic ebullition fluxes for weeks 6, 7, and 8 were significantly higher than those for weeks 2, 3, and 4 (P < 0.01 for all comparisons). After averaging steady fluxes and episodic ebullition fluxes measured within the same week and averaging CSC fluxes obtained before and after every experimental week, we found that steady fluxes from DFC were significantly lower than CSC fluxes (P < 0.05) but CSC fluxes were not significantly different from total fluxes (steady plus episodic ebullition) from DFC (P>0.05). Ebullition frequency followed the same temporal pattern as ebullition fluxes ($R^2=0.91$, P<0.01) with the first day of week 8 having the highest frequency of 54 event d⁻¹.

Figures 5A and 5C show the relationships between episodic ebullition fluxes, ebullition frequency and gas content of the three layers. Episodic ebullition fluxes increased dramatically when gas content of the middle and lower layers leveled off at 0.14 m³ m⁻³. Relationships between ebullition frequency and gas content for either of these layers tended to be more linear. In contrast, positive but weak relationships exist between dissolved CH₄ concentrations and episodic ebullition flux and frequency for the upper and lower layers (Fig. 5B and D).



Fgure 5 Relationships between gas content, dissolved CH₄ concentrations, ebullition fluxes, and ebullition frequency for three layers. Letter 'U', 'M', and 'L' in the figure represent upper, middle, and lower layers, respectively. Linear regression lines are shown. R^2 and *P* values of the linear regressions are shown in the legend and in bracket, respectively.

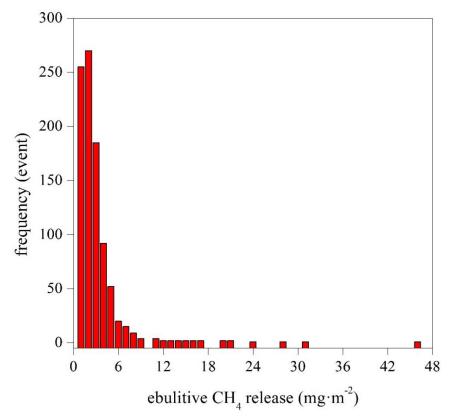


Figure 6 Frequency distribution of ebullition CH₄ releases observed over the entire study period (n=926).

There were in total 926 events of ebullitive CH₄ release, ranging from 0.12 mg m⁻² to 45.86 mg m⁻², observed in this study. The frequency distribution of ebullitive CH₄ releases was non-normal with a strong positive skew (Fig. 6), with 95% of ebullitive CH₄ releases being below 6.12 mg m⁻². A negative coupling between atmospheric pressure and ebullitive CH₄ releases is evident from plotting the time series of both parameters together (Fig. 7). Ebullition occurred more frequently in daytime than in nighttime over the entire experiment (P<0.01) (Fig. 8A and B). While the ebullitive CH₄ flux was not significantly different between daytime and nighttime for week 2 to week 5 (P>0.05), larger ebullitive CH₄ releases occurred during nighttime when temperature was lower for

week 6 to week 8 (P<0.01) (Fig. 8A and B). The cumulative ebullitive CH₄ releases and the frequency of ebullition, both calculated for every 12 hours in each experimental day, were significantly and negatively related to changes in atmospheric pressure, defined as the difference between averaged atmospheric pressure for any 12-hours and that of the previous 12-hours, especially in week 6 to week 8 (P<0.01 for all cases) (Fig. 8C and D).

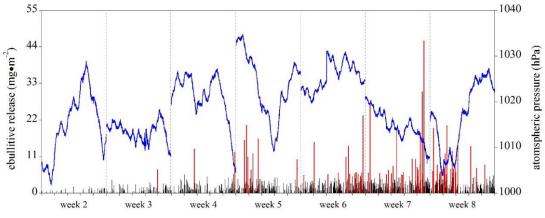


Figure 7 Ebullitive CH₄ releases and atmospheric pressure as a function of time. The largest 5% of ebullition CH₄ releases are shown in red.

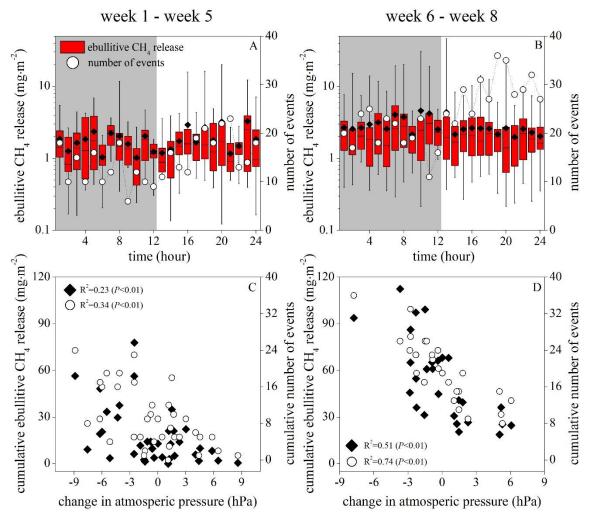


Figure 8 (A) and (B) ebullitive CH₄ releases observed every experimental hour for weeks 2-5 and weeks 6-8, respectively. The box shows the distance between the 25^{th} and 75^{th} percentiles, with the 50th percentile shown as a line. The error bars denote the minimum and maximum, and the filled diamond denotes the average. Note log scale for y axis. (C) and (D) relationships between cumulative ebullitive CH₄ release (filled diamond), cumulative number of ebullition (open cycle) and change in atmospheric pressure for weeks 2-5 and weeks 6-8, respectively. Linear regression R² and *P* values are shown.

4. Discussion

4.1. Reliability of DFC for continuously monitoring CH₄ ebullition

Using the graphical model adopted here, the quantification of episodic ebullition fluxes depends critically on the choice of the baseline. Unfortunately, the accumulation of ebullition-derived CH₄ in the gas chamber is a shortcoming of the DFC system as it is superimposed on the steady fluxes and thus complicates the baseline identification in the flux graphs. The steady baseline can be accurately defined when episodic ebullition is infrequent, but we have to assume a trend for the steady fluxes when the baseline is obscured by the ebullition peaks (Fig. 2B). We assume that higher ebullition frequency results in higher uncertainty in defining the baseline. We assigned an arbitrary error level, ranging from $\pm 1\%$ to $\pm 10\%$, proportional to the ebullition frequency, to the steady fluxes and propagated this error to the resulting episodic ebullition fluxes. As shown in Fig. 9, when ebullition frequency was higher than 27 event d⁻¹ and calculated episodic ebullition fluxes constituted over half of the total flux, the distribution of potential error in calculated episodic ebullition fluxes is constrained to a very narrow range, highlighting the insensitivity of episodic ebullition fluxes to the quantification of the steady fluxes during the later phase of the experiment.

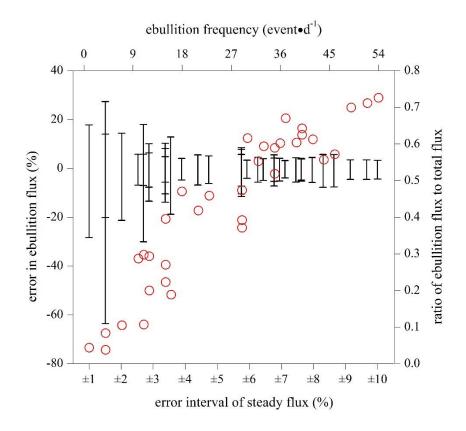


Fig. 9 Propagated error in calculated episodic ebullition fluxes using the graphical model (error bars) and relationship between ebullition frequency and the ratio of ebullition flux to total flux (red dots).

It may be argued that the natural CH₄ gradient at the air-peat interface was reduced by long-term closure of the gas chamber and excess accumulation of ebullition-derived CH₄ such that the natural regime of steady emission was disturbed [Hutchinson et al., 2000]. Averaged bulk CH₄ concentration of the peat monolith calculated by Eq. 8 (2.26 to 33.76 mg L⁻¹) was at least three orders higher than dissolved CH₄ concentration (0.04 to 1.13 μ g·L⁻¹) equilibrating with observed *C*_{out} (1.32 to 35.43 μ g·L⁻¹). In contrast to other types of soil, the combination of high porosity and essentially unlimited biogenic CH₄ production potential in peat results in a large CH₄ reservoir and high ebullition potential for peat. As a result, given the relatively high purging flow rate applied in this study, the disturbance of the natural gas exchange regime was minimal during the DFC measurements, and the smooth change of steady fluxes under ebullition peaks can be justified in the graphical model.

When compared on a weekly basis, steady fluxes measured by DFC were significantly lower than fluxes from CFC (P<0.05) but significantly higher than modeled diffusion fluxes (P < 0.01) (Fig. 4C). Due to the dominance of *Sphagnum* species, we can safely assume that plant-mediated CH_4 transport was negligible in this peat monolith [Stephen] et al., 1998b]. Subsequently, the differences between fluxes estimated from these three methods is most likely attributed to steady ebullition from the peat. Coulthard et al. [2009] proposed that shallow peat layers are sources of steady CH₄ ebullition that is released to the atmosphere as a continuous stream of relatively small bubbles, causing a linear increase of the CH₄ concentration in CSC headspace over time. However, the large differences in temporal resolution between CSC and DFC make it difficult to extract valuable information from the comparison of their results. While CSC measurements span 90 minutes for every experimental week, continuous DFC monitoring, which accounts for considerable temporal variability in the dissolved CH₄ reservoir in peat (Fig. 4B), provides a better characterization of temporal variability in CH₄ diffusion fluxes. In the later phase of our experiment, we observed much larger deviation in the replicate CSC measurements (Fig. 4C and A9). It is possible that the observed high CSC fluxes (e.g., those measured after week 4, week 7, and week 8, Fig. 4C and A9) might include time periods of active steady ebullition. After using the lowest values of replicate

measurements only to re-calculate CSC fluxes after week 4, week 7, and week 8, no significant difference was found between CSC fluxes and steady fluxes from DFC (P>0.05).

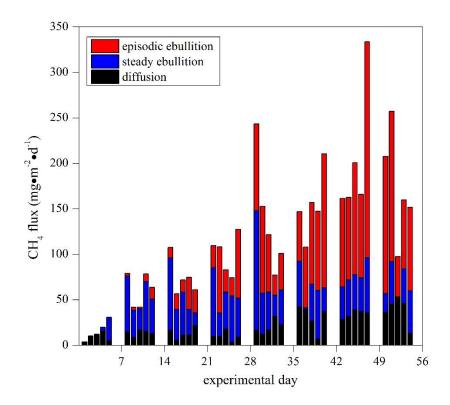


Fig. 10 Partitioned diffusion flux, steady ebullition flux, and episodic ebullition flux for each experimental day.

Therefore, we calculated the steady ebullition fluxes as the difference between steady fluxes from DFC and modeled diffusion fluxes (Fig. 10). Episodic ebullition constituted 0 to 72% of the total fluxes with an average of 38% throughout the experiment, being consistent with previous findings of 17% ~ 52% using similar flow-through systems at the same scale [Christensen et al., 2003]. Steady ebullition constituted 0 to 83% of the total fluxes with an average of 36%. To our knowledge, this is the first attempt to

quantify steady ebullition fluxes from peat. However, the definition of steady ebullition may depend on the detection sensitivity of measurement systems. Assuming that the mixing ratio of CH₄ to the total gases in a bubble is 0.2 [Tokida et al., 2007], the lowest observed ebullitive CH₄ release, 0.12 mg m⁻², suggests that a released gas bubble with volume smaller than ~0.08 cm³ may be classified as steady ebullition under the typical incubation conditions of this experiment. Moreover, with a single FMA we were unable to separate the potential contribution of fluctuations in *C_{in}* to the variations of measured steady fluxes (Fig. 3). However, this limitation could be easily fixed in a future study with better constrains on *C_{in}* dynamics.

We argue that the DFC implemented in a long-term chamber closure is a potentially superior technique to other conventional methods for continuous CH₄ ebullition monitoring in peat. In many cases, the underlying mechanical processes in the peat matrix may be independent of the respective ambient or chamber CH₄ concentrations, plant physiology, and turbulence conditions of the chamber headspace [Mastepanov and Christensen, 2009]. Furthermore, all CH₄ exchange from the isolated monolith can be directly measured by DFC for months without any missing ebullition events. Given the high temporal resolution of DFC, individual ebullition events can be explicitly identified even on a short time scale. Within combination with additional constrains on diffusion and plant-dependent emissions, the DFC system has the potential to precisely partition pathways of CH₄ emissions from peat soils.

4.2. Couplings between CH₄ ebullition and subsurface CH₄ dynamics

Previous laboratory studies have found that even under careful wetting procedures initial under-saturation occurred in undisturbed peat samples, such that the initial gas content was ranged from 0.01 to 0.13 m³ m⁻³ [Beckwith and Baird, 2001; Baird et al., 2004; Strack et al., 2005]. In our study initial gas contents ranged from 0.04 to 0.07 m³ m⁻³ for the three layers, and fast rates of gas buildup were observed before the first ebullition in week two, especially for the middle and lower layers (Fig. 4A). Although the time to the initiation of ebullition (10 days) was shorter than measured in other studies [e.g., Baird et al., 2004] due possibly to the higher incubation temperature of this study, the gas contents at the first ebullition, 0.05 to 0.11 m³ m⁻³, and maximum gas contents, 0.07 to 0.15 m³ m⁻ ³, are consistent with the range reported in previous studies [Baird et al., 2004; Strack et al., 2005; Kellner et al., 2006]. We also found that the dissolved CH₄ concentrations in pore water during gas buildup and initial ebullition were below the theoretical equilibrium concentration (~8 mg L^{-1}) previously assumed necessary for bubble formation (Fig. 4B) [Baird et al., 2004]. Recently, Laing et al. [2008], using membrane inlet quadrupole mass spectrometry, found an average dissolved CH₄ concentration of 1.3 mg L^{-1} to be equilibrated with gas bubbles comprising 11% by volume in shallow peat samples. Together with others, our findings support the argument that peatland CH₄ models only using nucleation to predict gas buildup and ebullition are likely to need revision because minute gas bubbles are always present in the peat pore space [Baird et al., 2004; Strack et al., 2005], causing a lowered equilibrium concentration for the dissolved phase and serving as initial nuclei for biogenic bubble growth Waddington et al., 2009]. Without taking gas bubble dynamics into account, peatland models tend to

underestimate ebullition fluxes by assuming a high dissolved CH₄ concentration for ebullition initiation [Baird et al., 2004].

In a field study in a near-surface fen peat, Strack et al. [2005] reported positive relationships between dissolved CH₄ concentrations and bubble fluxes at different peat depth, and it has also been shown that large reductions in dissolved CH₄ concentrations can occur during short time periods of active ebullition [Strack and Waddington, 2008]. In this study dissolved CH₄ concentrations correlate to gas contents positively and significantly in the upper layer ($R^2=0.65$, P<0.01), but no significant correlation was detected in the middle and lower layers where gas contents were consistently larger (P>0.05). Moreover, dissolved CH₄ concentrations resolved limited variations in episodic ebullition fluxes and frequency (Fig. 5C and D). Several factors may explain the generally decoupled relationship between dissolved CH₄ concentrations and gas contents and episodic ebullition. One possibility is that dissolved CH₄ concentrations are highly variable spatially within the peat and at scales much smaller than the pore-water samplers [Baird et al., 2004; Reid et al., 2013]. Consequently, the measured dissolved CH₄ concentrations might be subject to the micro-environment in the immediate vicinity of the pore-water samplers [Waddington et al., 2009]. Such scaling problems may also exist in field studies using spatially discrete sampling methods. More importantly, due to its low solubility, CH₄ is predominantly stored in gaseous form even when a small bubble volume is present in the peat [Strack and Waddington, 2008]. The disproportion in subsurface CH₄ stocks implies that: Dissolved CH₄ concentrations may be regulated by the storage and release of gas bubbles, resulting in dramatic changes in concentration

during the main period of ebullition (Fig. 4C). Another possibility is that CH₄ bubbles can be produced directly by granular microbial consortia containing methanogens such that dissolved CH₄ concentration is controlled by bubbles going into solution [Green and Baird, 2012]. Based on our observations, we suggest that gas bubble dynamics in nearsurface peat may be decoupled from the dissolved CH₄ pool and the physical characteristics of peat alone can then be important in determining ebullition fluxes [Laing et al., 2008].

As first proposed by Baird et al. [2004], peat soils may need to exceed a threshold bubble volume before ebullition can occur. In an effort to model ebullition based on this threshold theory, Kellner et al. [2006] further suggested a fuzzy threshold, rather than a single value, related to intrinsic properties of peat-bubble interactions, is responsible for the complex ebullition behavior of peat. Despite its conceptual simplicity [Coulthard et al., 2009], the threshold theory has proven to be effective in several laboratory [Comas and Slater, 2007; Slater et al., 2007] and field [Strack et al., 2005] studies with resolved threshold gas content ranging from 0.07 to 0.16 m³ m⁻³ for poorly-decomposed nearsurface Sphagnum peat. In our study, distinct differences in gas content and rate of gas bubble buildup among three layers were resolved (Fig. 4A). While all three layers showed a significant linear relationship between gas content and ebullition fluxes, ebullition fluxes increased dramatically after the gas content of middle and lower layers leveled off at 0.14 (Fig. 5A), an observation consistent with threshold-based ebullition. Thus, our work supports threshold theory using directly measured episodic CH₄ ebullition fluxes. More importantly, different relationships between gas content and episodic

ebullition fluxes found in the three layers highlight the role that vertical heterogeneity of peat structure may have on gas bubble dynamics even in this very shallow peat monolith.

The physical properties of peat, including the porosity, bulk density, decomposition degree, pore size distribution, and compressibility, exert a strong control on peatland ecohydrology and biogeochemistry [Price et al., 2005]. Recently, geophysical imaging methods have been employed to better understand how the physical structure of peat entraps gas bubbles. By using X-ray computed tomography, Kettridge and Binley [2011] concluded that the compressibility and the ability of Sphagnum peat to trap biogenic gas bubbles depends not only on the bulk volume of the peat constituents, which is usually inferred from the bulk density and porosity, but also on their spatial arrangement within the peat. The presence of longer structural components, like stems and branches, strongly increases the tortuosity of pore networks and the ability of peat to trap bubbles. In a laboratory study using electrical resistivity, Slater et al. [2007] revealed that biogenic gas bubbles tend to accumulate at certain depths (within 0.04 to 0.10 m below the water table in their case) within the shallow *Sphagnum* peat as thin layers due to the prevalently layered nature of peat. Based on such findings, differences in gas content and the relationship between gas content and episodic ebullition fluxes among three layers in our peat monolith may be attributed to layered structural differences in peat. The compressibility of peat decreases with increasing decomposition degree [Price et al., 2005]. As peat decomposes with depth, plant remnants successively adjust from a disordered to a layered firmer framework [Boelter, 1969]. In this study, more decomposed middle and lower layers with the presence of rigid root fibers and wood

inclusions had higher bulk densities than the upper layer and may have a denser and more tortuous pore networks to entrap gas bubbles [Kettridge and Binley, 2011]. On the other hand, decreased compaction near the peat surface together with the absence of rigid structural components may make the upper layer less resistant to bubble release [Comas and Slater, 2007].

Interestingly, ebullition frequency increased linearly with gas content and did not exhibit a threshold effect (Fig. 5B). Given that dissolved CH₄ concentrations measured from the three layers were always supersaturated against the chamber air (Fig. 4B), it is reasonable to speculate that the enhanced episodic ebullition in the last three weeks (Fig. 7) is attributed to the less frequent releases of large gas bubbles, rather than depth-independent releases of bubbles with more concentrated CH₄. While the upper layer with a lower bubble trapping potential might have a fast "bubble turnover" and be the source of small episodic ebullition events, large gas bubbles might be episodically vented from the middle and lower layers once the high ebullition threshold was reached. Therefore, we argued that the latest peatland CH_4 models applying a single ebullition threshold for the entire depth of peat [e.g., Zhang et al., 2012] may not able to resolve the complex spatiotemporal pattern of CH₄ ebullition. Similarly, models adopting the pressure balance criteria to predict ebullition [e.g., Tang et al., 2010] may fail to capture large ebullition events originating from deep peat layers where the bubble trapping potential is much higher and over-pressurized stratum are formed. We propose that a depth-dependent multi-threshold model determined by peat physical structure should be adopted in CH₄ models as the criteria for ebullition prediction.

4.3. Relative importance of atmospheric pressure and temperature as ebullition trigger

Both atmospheric pressure and temperature have been identified as ebullition triggers (Table 1). In this study, the incubation temperatures $(23 \,^{\circ}\text{C} \sim 27 \,^{\circ}\text{C})$ were much higher than the peat monolith studied would experience naturally on an annual basis. Optimal temperatures for CH₄ production have been found to be between 10 $^{\circ}$ C and 12 $^{\circ}$ C for northern peat samples [Williams and Crawford, 1984]. Bergman et al. [1998] suggested that high-temperature incubations (above 20 $^{\circ}$ C in their case) of *Sphagnum* peat tend to decouple CH₄ production from temperature. In this sense, a temperature higher than the optimal range for CH₄ production is desirable to minimize its biological control on CH₄ ebullition. On the other hand, an approximately 4 $^{\circ}$ C fluctuation in the diel cycle (Fig. 3), theoretically identical to a 40 hPa change in atmospheric pressure in terms of driving bubble volume alteration [Tokida et al., 2009], is sufficiently large to assess the triggering effects of temperature.

In week 2 to week 5, ebullitive CH₄ releases had no significant difference between daytime and nighttime (Fig. 8A), and ebullition fluxes were not coincide with changes in atmospheric pressure (Fig. 8C), indicating that both changes in temperature and atmospheric pressure had limited controls on CH₄ ebullition when gas contents were far below the ebullition thresholds. When the bubble buildup in the middle and lower layers was substantial in week 6 to week 8, the ebullition frequency increased greatly in the daytime (Fig. 8B). If the increased daytime temperature was able to trigger episodic ebullition from the middle and lower layers during this time period, we would expect to see outliers of ebullitive CH₄ releases, illustrated by the length of upper error bars for every experimental hour in Fig. 8B, during daytime. In fact, however, those outliers tended to occur in nighttime (Fig. 8B) and were often found to be coupled with large drops in atmospheric pressure (Fig. 7). Due to the highly skewed frequency distribution of ebullitive CH₄ releases, those large CH₄ ebullition events might dominate the timeintegrative ebullition fluxes (Fig. 6). Therefore, the negative linear relationship between cumulative ebullitive CH₄ releases and changes in atmospheric pressure with largely improved significance in week 6 to week 8 (Fig. 8D) further suggests that atmospheric pressure was responsible for triggering large CH₄ ebullition from deeper layers where big bubbles were trapped.

The effectiveness of temperature changes in triggering ebullition may result from the thermal conductivity of peat [Yoshikawa et al., 2009]. Previous studies have illustrated that the thermal conductivity of peat decreases with water content and decomposition degree [O'Donnell, 2009]. In our study, the unsaturated moss layer in the upper 2 cm might insulate the deeper layers from warm air in daytime. In addition, even below the water table, poorly decomposed surface peat might also dampen the amplitude of downward heat flux to a large degree [Yoshikawa et al., 2009]. Although we do not have enough information on the temperature dynamics of deeper layers, as implied by other studies [e.g., Tokida et al., 2007] the diel temperature cycle in the middle and lower layers might be small. Therefore, we speculated that the triggering effects of temperature were mostly limited to the upper layer where a low ebullition threshold existed, which resulted in small and frequent ebullitive CH4 releases observed in daytime (Fig. 8B).

On the other hand, since the pressure of a gas bubble below the water table is always the sum of hydrostatic and atmospheric pressures, changes in atmospheric pressure influence the volume of the gas bubble trapped throughout peat profiles in a more direct manner. Moreover, the alteration of gas bubble caused by changes in atmospheric pressure occurs almost instantly, giving the CH₄ ebullition an episodic nature. While increasing atmospheric pressure was also found to facilitate small ebullition events from upper layers of the peat by allowing size-reduced gas bubbles to pass through fiber networks [Comas and Wright, 2012], our data suggests that large ebullition events originating from deeper layers are triggered by drops in atmospheric pressure in peat soils.

5. Conclusions

In this study, we successfully demonstrated the capability of DFC for continuously monitoring CH₄ ebullition from a near-surface *Sphagnum* peat and its potential to partition the pathways of CH₄ emission. The application of DFC with coupled subsurface gas content and dissolved CH₄ concentration measurements at laboratory scale offers new opportunities for understanding the complex behavior of CH₄ ebullition and its couplings to subsurface gas bubble dynamics. Our work highlights a need to integrate peat physical structure into ebullition models. The structural heterogeneity of peat gives rise to a depth-dependent multi-threshold model that was responsible for the spatiotemporal variation of CH₄ ebullition and partially determines the relative importance of changes in atmospheric pressure and temperature in triggering ebullitive CH₄ releases. Future efforts should be dedicated to applying the DFC technique at the field scale to better improve our knowledge of CH₄ ebullition in northern peatlands.

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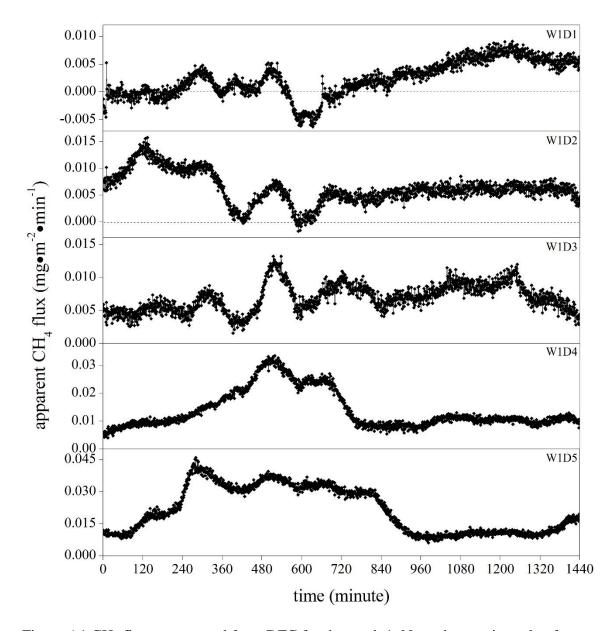
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Appendix: CH₄ flux data from DFC and CSC measurements

Figure A1 CH₄ fluxes measured from DFC for the week 1. Note changes in scale of y axis.

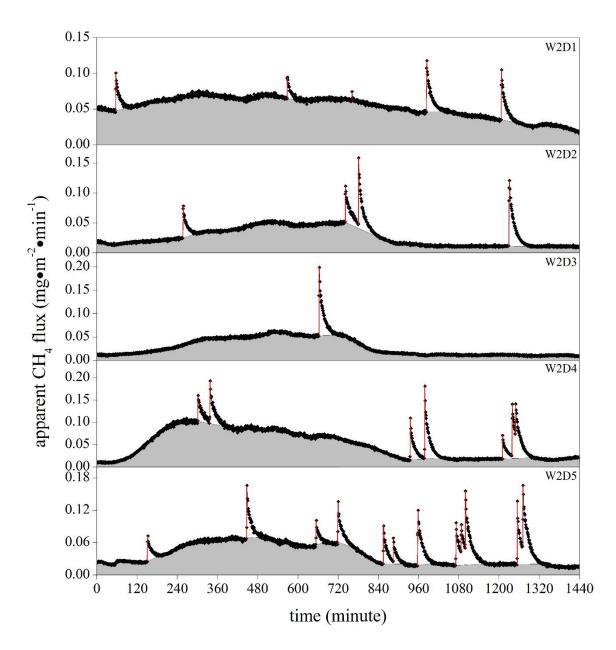


Figure A2 CH₄ fluxes measured from DFC for the week 2. Steady flux is shaded in gray, and ebullitive CH₄ releases are labeled in red lines. Note changes in scale of y axis.

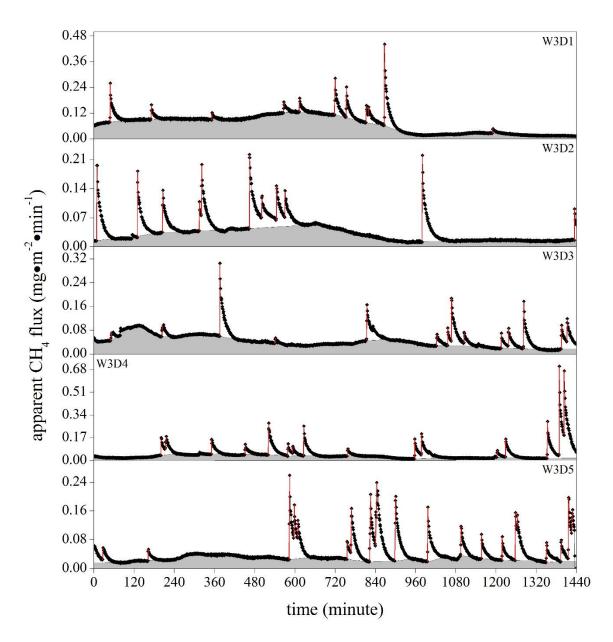


Figure A3 CH₄ fluxes measured from DFC for the week 3. Steady flux is shaded in gray, and ebullitive CH₄ releases are labeled in red lines. Note changes in scale of y axis.

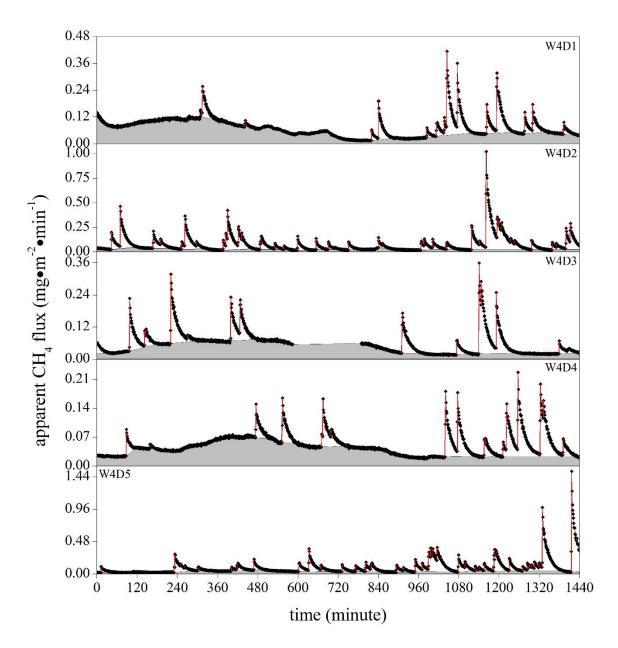


Figure A4 CH₄ fluxes measured from DFC for the week 4. Steady flux is shaded in gray, and ebullitive CH₄ releases are labeled in red lines. The data gap in the fourth day was due to an electrical outage in the laboratory. Note changes in scale of y axis.

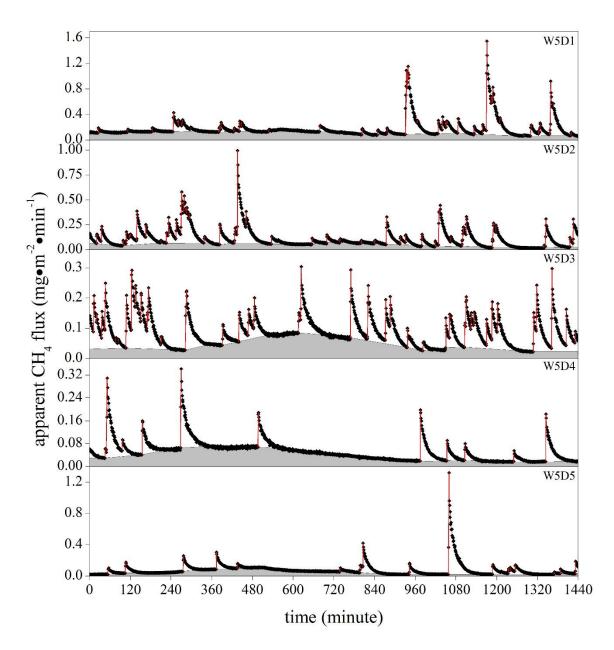


Figure A5 CH₄ fluxes measured from DFC for the week 5. Steady flux is shaded in gray, and ebullitive CH₄ releases are labeled in red lines. Note changes in scale of y axis.

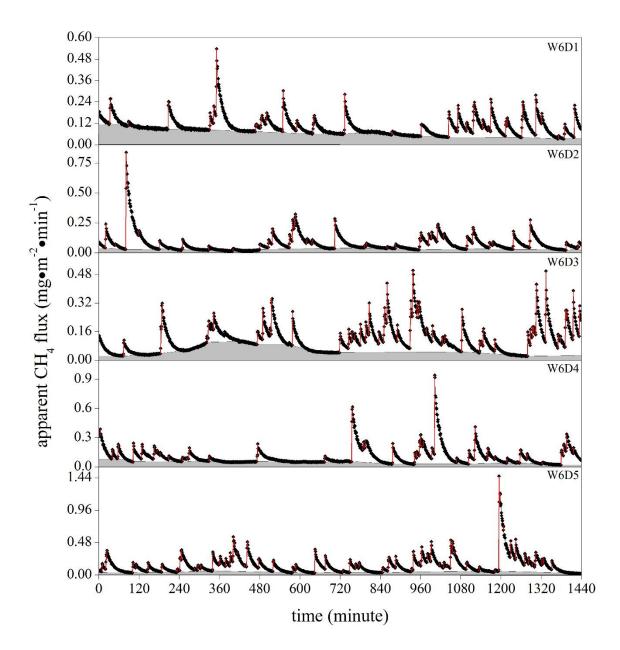


Figure A6 CH₄ fluxes measured from DFC for the week 6. Steady flux is shaded in gray, and ebullitive CH₄ releases are labeled in red lines. Note changes in scale of y axis.

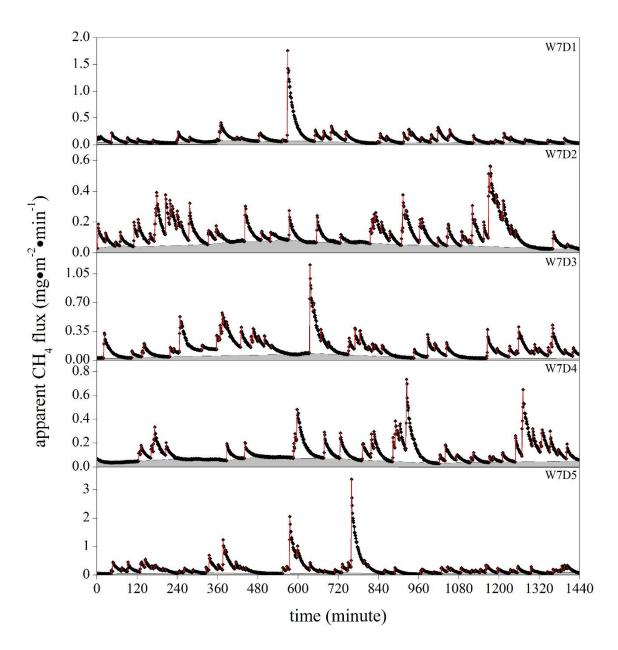


Figure A7 CH₄ fluxes measured from DFC for the week 7. Steady flux is shaded in gray, and ebullitive CH₄ releases are labeled in red lines. Note changes in scale of y axis.

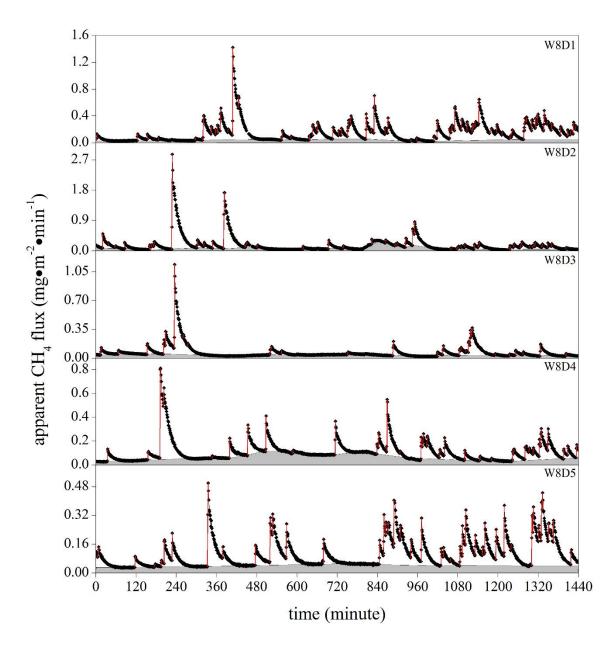


Figure A8 CH₄ fluxes measured from DFC for the week 8. Steady flux is shaded in gray, and ebullitive CH₄ releases are labeled in red lines. Note changes in scale of y axis.

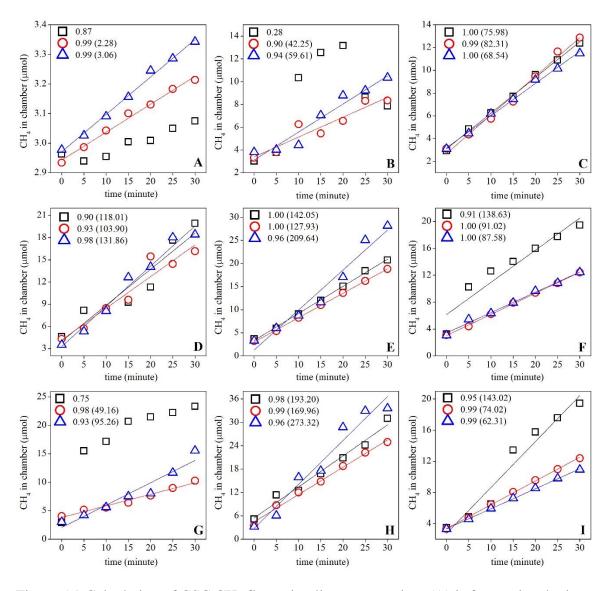


Figure A9 Calculation of CSC CH₄ flux using linear regression. (A) is for pre-incubation condition, and (B) to (I) are for CSC measurements taken after DFC measurement in every experimental week, respectively. Different symbols in each panel are replicate measurements. Coefficient of determination (R^2) is shown in the legend. A criteria is set to be R^2 >0.9 for flux calculation. CSC CH₄ flux (mg m⁻² d⁻¹) from qualified dataset is shown in the bracket. Please note the changes in scale of the y-axis among different panels.

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