#### LINKING PHOTOSYNTHETIC AND CARBON METABOLISM IN MICROALGAE

#### FOR BIOFUEL APPLICATIONS

By

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#### ABSTRACT OF THE DISSERTATION

#### Linking photosynthetic and carbon metabolism in microalgae for biofuel applications

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Microalgae predominantly partition photosynthetically fixed carbon into proteins, starch and lipids. Of these, carbohydrates and lipids are desired as they are the precursors for biofuel production. Photosynthetic electron transport is closely coupled to carbon partitioning, thus frustrating efforts to substantially increase the yield of the desired terminal product without compromising photosynthetic fitness. The objective of this thesis was to investigate the role of starch biosynthesis with respect to photosynthesis and carbon partitioning in a model microalga *Chlamydomonas reinhardtii*. For this, I have used a series of starch deficient mutants together with nitrogen deprivation to modulate the normal carbon partitioning. Nitrogen deprivation is known to redirect biosynthesis genes. Starch-deficient mutants showed a 20-40% lower biomass accumulation under both nutrient replete and deplete conditions. Interrogation of the photosynthetic metabolic flux (water $\rightarrow$ PSII/PSI $\rightarrow$  ATP & NADPH  $\rightarrow$  3PG) revealed that above a threshold light intensity, starch deficiency attenuated NADPH reoxidation by the Calvin-Benson-Bassham (CBB) cycle, which attenuated water oxidation at PSII by product inhibition. Even with starch biosynthesis blocked in the starch-deficient mutant, a high gluconeogenic flux was maintained by redirecting carbon

through the oxidative pentose phosphate (OPP) shunt and ultimately away from starch. Thus, upon loss of the ability to synthesize starch, water oxidation is attenuated to balance energy consumption, while the OPP shunt becomes the dominant pathway for repartitioning of the residual photosynthate that can be produced.

The goal of the second part of the thesis was to use metabolic principles to develop a cyanobacterial mutant capable of attaining high  $H_2$  yield. Cyanobacteria catabolize the photosynthetically assimilated glycogen under anaerobic auto-fermentative conditions and produce hydrogen via the enzyme, hydrogenase. By sequentially enhancing the glycolytic rate together with the elimination of competing pathways in a euryhaline cyanobacterium, *Synechococcus sp.* PCC 7002 mutant with high hydrogenase gene expression, we were able to boost  $H_2$  production by 8-fold over the wild-type strain.

Thus, my dissertation, addresses understanding how to control the metabolism of photosynthetic microbes and using targeted metabolic engineering to transform microalgae into efficient cell factories for biofuel production.

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### Chapter I: Introduction

#### 1. Background: Microalgae for biofuels

Microalgae are a diverse group of fast growing, microorganisms comprising of free living, photosynthetic, prokaryotic or eukaryotic organisms that can convert  $CO_2$  into potential bioproducts. Occupying a diverse range of habitats, microalgae have developed enormous metabolic flexibility (Richmond, 2004), (Hu et al., 2008) and contribute to half of the global photosynthetic activity (Wiessner et al., 1995). The concept of using microalgae as biofuel feedstock gained importance during the oil embargo in the 1970s, with the Department of Energy establishing the 'Aquatic Species Program' (Sheehan, 1998). Originally, only the carbohydrate fraction of microalgae was suggested as a feedstock for biogas production (Meier, 1955). However, later on, it was found that nutrient stressed microalgae could accumulate lipids (triacylglycerols and diacylglycerols) which could be harvested and directly used as fuels. With recent advances in metabolic engineering, microalgae have become excellent platforms to produce designer biofuels in addition to inherent metabolic products such as oils, starch and  $H_2$  (Sheehan, 1998),(Schenk et al., 2008), (Dismukes et al., 2008), (Rittmann, 2008; Gimpel et al., 2013), (Radakovits et al., 2010), (Hannon et al., 2010), (Hu et al., 2008), (Chisti, 2008), (Brennan et al., 2010), (Brune et al., 2009). Especially from a biofuels standpoint, a diverse range of fuel precursors have been produced at least at the pilot scale including but not limited to : 1) biodiesel through transesterification (Chisti, 2007), (Hu et al., 2008), (Miao et al., 2004), 2) photobiological and fermentative bioH<sub>2</sub> production (Melis, 2002), 3) biomethane production through anaerobic digestion (McKendry, 2002), (McKendry, 2002), 4) bioethanol through alcoholic fermentation, 5) bio-oil and syngas through thermochemical liquefaction and pyrolysis (McKendry, 2002), (Miao et al., 2004) and 6) electricity through direct combustion (Brennan et al., 2010). The enormous metabolic flexibility of microalgae allows the use of both aerobic and anaerobic metabolisms for biofuel production.

Besides being the most productive phototrophic organisms, microalgae also confer additional advantages over the existing terrestrial agricultural/forest biomass sources. Unlike terrestrial crops, microalgae are a) devoid of recalcitrant woody matter (cellulose) (Wijffels *et al.*, 2010), (Georgianna *et al.*, 2012), (Dismukes *et al.*, 2008) b) can be cultivated in marginal, saline or wastewater throughout the year without seasonal constraints and c) can thrive in light environments not used by terrestrial plants. Finally, algal biofuels can be produced locally making them a viable alternative for developing countries as well as providing energy security.



Figure 1-1 Representative examples of microalgae

Many microalgal species are known to accumulate about (20-50) % of their cellular dry weight as lipids such as triacylglycerols and diacylglycerols (Griffiths *et al.*, 2009), (Chisti, 2007), (Tornabene *et al.*, 1983). These lipids can yield biodiesel through transesterification or on further distillation and cracking can yield gasoline or jet fuel. Current yields for the existing wild strains are about 20 g m<sup>-2</sup>day<sup>-1</sup> equaling 60-70 dry tons per hectare per year of biomass or 20 tons of oil useful for biodiesel production. These yields are over 10 times higher than any other dry land crops such as Jatropha or Castor, and 3 to 4 times higher than those achieved with oil palm which requires irreplaceable rain forest habitat. Algae also have important implications from the overall land usage. Algae required to replace 1200 billion liters of petroleum used by the United States can be cultivated on only 30 million hectares and can yield upto 40,000 liters per hectare per year (Georgianna *et al.*, 2012). Additionally, algal products have many alternative markets including animal/fish feedstock, nutrichemicals and industrial chemical precursors that provide economic stability for investment.

#### 2. Technological barriers to algal biofuels

High growth rates, reasonable growth densities and high oil contents have all been cited as reasons to invest significant capital to turn algae into biofuels. Numerous technoeconomic analyses of algal oil production costs have already been conducted (Venteris *et al.*, 2013). Cost data for the year 2008, estimated the US cost between \$10.87 gallon<sup>-1</sup> and \$13.32 gallon<sup>-1</sup> (Sun *et al.*, 2011) with recent reports from the National Alliance for Advanced Biofuels and Bioproducts (NAABB) consortium claiming further reduction to \$8 gallon<sup>-1</sup>. Yet, the current costs make algal biofuels less competitive in comparison to fossil fuels and without further improvements, microalgal fuels are economically unviable. It is a widely accepted fact that the current technology is limited by a) developing suitable algal strains: in terms of oil productivity, crop protection, nutrient and resource allocation, and the production of co-products

b) process development to harvest biomass and extract biofuel precursors (Fishman *et al.*, 2010),(Hannon *et al.*, 2010).

The current solar to biomass conversion efficiency of microalgae (3%) fall well short of the theoretical upper limit (8-10%) (Melis, 2009). In addition, the higher lipid accumulators (more suitable for biofuel production) have been empirically found to possess lower growth rates. This led to the prediction that significant improvements in biofuel production are possible only through better understanding of algal physiology and carbon partitioning which will enable the development of robust transgenic strains that produce high levels of lipids without compromising fitness. This topic is a rich area of research that is yielding significant progress as reported in scientific journals. Significant improvements in growth, crop protection against heat stress and parasitic contamination and in lipid production is projected to reduce the production cost further to \$4.5 gallon  $^{-1}$  (Hannon *et al.*, 2010), (Pienkos *et al.*, 2009). For this, a thorough understanding of the molecular mechanisms and the feedbacks underlying microbial conversion pathways is essential.

#### 3. Photoautotrophic growth and carbon allocation for high lipid production

#### 3.1) Nutrient starvation for lipid production and modification of carbon partitioning

In microalgae, biomass composition and therefore, the amount of lipid accumulated is highly dependent on environmental and growth conditions (Becker, 1994). Under favorable growth conditions, even the known oleaginous algae produce very low amounts of triacyl glycerols (TAG) (Liang *et al.*, 2013), (Becker, 1994). Only on the exposure to stress, either chemical or physical, does large amounts of TAG accumulation occurs (Sharma *et al.*, 2012). Nutrient starvation, especially nitrogen starvation is one of the most widely and successfully applied methods to induce and increase the lipid production by up to 53% (Sharma *et al.*, 2012). Upon N starvation, protein and nucleic acid synthesis is inhibited, while the photosynthetic electron transport chain remains active, albeit at lower efficiencies. To prevent the overreduction of the electron transport chain,

various algae accumulate energy-rich storage compounds such as starch and storage lipids (TAGs) (Guschina *et al.*, 2006), (Siaut *et al.*, 2011), (Hu *et al.*, 2008), (Li *et al.*, 2010), (Rodolfi *et al.*, 2009), (Wang *et al.*, 2009), (Moellering *et al.*, 2010), (Work *et al.*, 2010), (Fan *et al.*, 2011). On imposing N starvation, there is a shift in allocation of carbon from proteins to starch and lipids making nitrogen starvation an excellent method to study biosynthetic remodeling. Understanding of the carbon decision tree - the temporal and spatial coordination of biochemical pathways responsible for biosynthesis and catabolism of the major carbon sinks - is in part the focus of this thesis.

#### 3.2) Microalgal energetics and central carbon metabolism under photoautotrophic conditions

Photosynthesis is the fundamental process that converts solar energy into chemical energy where ATP and NADPH produced by the "light reactions" of photosynthesis fuel the "lightindependent reactions" of carbon fixation. Photosynthetically fixed carbon enters central carbon metabolism and a series of enzymatic reactions convert the carbon into metabolic precursors that are destined for biopolymers such as lipids, proteins and carbohydrates.

**Photosynthesis:** The photosynthetic electron transport chain begins with water oxidation at Photosystem II (PSII). Electrons from water are transferred through a series of electron carriers to Photosystem I (PSI). Simultaneously, the excitation of PSI relays electrons via a second series of carriers, ultimately reducing ferredoxin (Fd). Ferredoxin-NADP+ reductase (FNR) use ferredoxin as the substrate for reducing NADP+ to NADPH. Photosynthetic electron transport also establishes a proton (H+) gradient across the thylakoid lumen that is used for generating ATP. As a major photosynthetic electron sink under photoautotrophic conditions, the Calvin-Benson-Basham cycle (CBB cycle) uses two NADPH and three ATP molecules to fix one CO2 molecule into 3-phosphoglycerate (3PG). Triose phosphates (Triose-P) generated in the CBB cycle are directed to three major carbon sinks: carbohydrates, amino acids and lipids. Even though the light-driven redox chemistry of light reactions and the enzymatic reactions of metabolism are spatially separated, they are tightly coupled and necessary for the balancing of ATP and NADPH (Ott *et al.*, 1999). This

form of self-regulation protects the photosystems from irreversible damage. Changing the carbon partitioning will have a direct impact on photosynthetic efficiency and ultimately algal yield.

<u>Terminal sinks and energetics under photoautotrophic conditions</u>: The primary product of CO<sub>2</sub> fixation 3PG enters into central carbon metabolism and is ultimately destined to either a structural or a storage biopolymer or used for energy generation and cell homeostasis. Central carbon metabolism is composed of multiple sub-pathways including the Calvin Benson Bassham cycle (CBB), glycolysis, gluconeogenesis, the oxidative pentose phosphate pathway (OPP), and the tricarboxylic acid (TCA) cycle.

Under standard growth conditions, the major sinks for fixed carbon are proteins, cell wall carbohydrates and membrane lipids (Boyle *et al.*, 2009). While under stress conditions, energy-rich storage compounds such as starch and storage lipids (oil) become the dominant sinks. All the energy stored in the biopolymers is either directly or indirectly originating from light when algae is grown under photoautotrophic conditions. The ATP and NADPH originating from the light reaction are also used for other processes than to produce terminal biopolymers, e.g. to reducing nitrogen and sulfur as well as cell homeostasis. The energetic requirement of each of the terminal sinks is one of the key factors that governs the amount of energy partitioned between the storage sink and the cellular growth leading to variations in growth rates. The energy consumption makes central carbon metabolism a key link in solar to biomass energy conversion in microalgae.

Based on the metabolic pathways used for the sink generation, thermodynamically, lipids are the most expensive terminal product while starch is the least energy requiring product. Glucose derived by the condensation of 3PG is polymerized to starch and requires 4.2 ATP/C and 2 NADPH/C polymerized (Subramanian *et al.*, 2013), (Raven, 1982), while the energetic requirement for TAG/structural lipid biosynthesis ranges from 5.84 ATP/C and 2.84 NADPH/C to 6.3 ATP/C and 2.9 NADPH/C (Subramanian *et al.*, 2013), (Raven, 1982). On the other hand, protein biosynthesis in a freshwater algae requires about 4.98 ATP/C and 2.77 NADPH/C (Raven,

1982). Typical algal (total) biomass productivities range from 15-30 g dry weight m<sup>-2</sup> day<sup>-1</sup> (Batan et al., 2010), containing about (Campbell et al., 2011) 4.3 to 7.0 kcal/gdw (Illman et al., 2000), (Williams et al., 2010). Typically, the oleaginous strains have higher energy densities (Subramanian et al., 2013). Consistent with the thermodynamic requirements, there is a trade-off between cell division rates and the accumulation of the storage products. Typically, organisms with higher lipid densities have slower growth rates. Nonetheless, it is important to note that product generation is not purely governed by thermodynamic constraints but by a number of other factors including the light absorption spectrum, size/orientation of the light harvesting complex, efficiency of photon absorption and energy transfer, kinetic rate constants of the various electron transfer reactions, availability of substrate metabolites, metabolic compartmentalization and feedback control, respiration rates, and the partitioning of reduced carbon between new cell growth and division and energy storage (Cardol et al., 2011), (Radakovits et al., 2010). Thus, both kinetic and thermodynamic constraints need to be taken into account while optimizing biofuel production by microalgae and improving microalgal strain performances definitely requires a better understanding of the mechanisms and regulation of carbon fixation and carbon allocation between biosynthetic pathways.

#### 4. Fermentative metabolism and carbon allocation for high H<sub>2</sub> production

Though microalgae have a dominant aerobic metabolism that uses oxygen dependent respiration and photosynthesis for energy generation, certain algal species have retained their original ability of anaerobic fermentation. Together with using microalgae for lipid/biomass production, microalgae can also be used for bioH<sub>2</sub> production by harnessing the fermentative metabolic capacity. H<sub>2</sub> is a clean fuel and burns without releasing CO<sub>2</sub>. Unlike the previously described eukaryotic algae, prokaryotic cyanobacteria, prefer to follow dark fermentation preceding the photoautotrophic growth and produce H<sub>2</sub>. Cyanobacteria possess a heteropentameric enzyme called hydrogenase catalyzing the reversible reaction

#### $2H^+ + NAD(P)H \leftrightarrow H_2 + NAD(P)^+$

The second part of the thesis (Chapter 4) addresses fermentative metabolism in the prokaryotic microalgae cyanobacteria, and how metabolic engineering can be effectively used to improve fermentative H<sub>2</sub> production by cyanobacteria.

#### 4.1) Microalgal energetics and the central carbon metabolism under fermentative conditions

Anaerobic metabolism under dark anoxic conditions is initiated by degradation of photosynthetically assimilated glycogen into monomeric glucose molecules that can either enter the pentose phosphate pathway or undergo glycolytic breakdown (Fig. 1-2).



Figure 1-2. **Fermentative metabolic pathways in** *Synechococcus* **7002.** Fermentative metabolic pathways in *Synechococcus* **7002** adapted from (McNeely et al., 2010). Abbreviations: OPP, oxidative pentose phosphate pathway; G6P, glucose-6-phosphate; F-6-P, fructose-6-phosphate; FBP, fructose bisphosphate; GAP, glyceraldehyde 3-phosphate; DHAP, dihydroxyacetone phosphate; BPG, bisphosophoglycerate, PEP, phosphoenol pyruvate; PYR, pyruvate; Ac-CoA, acetyl-CoA; HXK, hexokinase; PFK, phosphofructokinase; ALDO-TPI, aldolase-triose phosphate isomerase; GAPDH, glyceraldehyde 3- phosphate dehydrogenase; PGK, phosphoglycerate kinase; GPM, phosphoglycero mutase; ENO, enolase; PPSA, phosphoenol pyruvate synthase; PK, pyruvate kinase; PDH, pyruvate dehydrogenase; PFR, pyruvate ferredoxin oxidoreductase; LDH, lactate dehydrogenase, HOX, hydrogenase.

The OPP pathway generates 6 NADPH and 1 NADH as opposed to the glycolytic pathway which generates only 2 NADH molecules per equivalent of glucose. As OPP pathway produces greater NADPH, redirection of carbon through this pathway has been shown to enhance  $H_2$  production (Kumaraswamy *et al.*, 2013). Pyruvate produced via the above catabolic routes then enters various fermentative processes to replenish the available NAD<sup>+</sup>. The major end products of fermentation include acetate, formate, lactate, ethanol together with hydrogen and CO<sub>2</sub> (Gfeller *et al.*, 1984). Of these, lactate, ethanol and H<sub>2</sub> production compete with each other for the reductant generated through catabolism and in turn lower the H<sub>2</sub> productivity. The biggest drawback for H<sub>2</sub> production from cyanobacteria are the fluxes and yield for H<sub>2</sub> production. If improved, H<sub>2</sub> offers a carbon-free, energy intensive molecule that would be the first choice of renewable energy.

#### 5. Scope of the thesis

The scope of this thesis is two-fold: Chapters 2 and 3 are focused towards understanding the cellular distribution of carbon, referred to as carbon partitioning in microalgal cells and provide insights into the metabolic pathways that control the flow of carbon to the respective sink. How a change in carbon partitioning feedback controls photosynthetic electron transport is also addressed. Further, in Chapter 4, I present the results of an iterative metabolic engineering strategy guided by prior knowledge of cyanobacterial fermentative metabolism to increase biohydrogen production. Thus, the overall goal of my research has been understanding and harnessing the potential of algae as biofuel platforms.

The following three manuscripts presented as three separate chapters in the dissertation are:

Krishnan, A.; Kumaraswamy, G.K.;Vinyard, D.J.; Gu, H.; Ananyev, G.; Posewitz, M.;Dismukes, G.C. "Metabolic and photosynthetic consequences of a lack of starch biosynthesis in the green alga Chlamydomonas reinhardtii sta6 mutant." The Plant Journal, 2015, epub DOI: 10.1111/tpj.12783

2) Krishnan, A.; Gu, H.; Kumaraswamy, G.K.; Ananyev, G.M.; Posewitz, M.; Dismukes, G.C. "The carbon decision tree of Chlamydomonas reinhardtii: How starchless strains redirect carbon away from starch." Submitted to *The Journal of Experimental Botany*.

3) Krishnan, A.; Zhang, S.; Liu, T.; Bryant, D.; Dismukes, G.C. "Consequences of *ccmR* deletion on respiration, fermentation and H<sub>2</sub> metabolism in cyanobacterium *Synechococcus sp.* PCC 7002" Submitted to *Biotechnology and Bioengineering*.

An overall summary in Chapter 5 highlights the main conclusions of the work.

# Chapter 2 - Metabolic and photosynthetic consequences of blocking starch biosynthesis in the green alga *Chlamydomonas reinhardtii sta6* mutant

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#### Summary

The ADP-glucose pyrophosphorylase (AGPase) deletion strain of *Chlamydomonas reinhardtii*, *sta6*, cannot make starch but is known to accumulate lipids under nitrogendeprived photoautotrophic growth. However, *sta6* has a lower rate of biomass accumulation under nitrogen replete photoautotrophic conditions, which offsets its biotechnological value. To investigate the underlying metabolism under nutrient replete photoautotrophic growth, we have monitored pool sizes of the adenylates, pyridine nucleotides, and central carbon metabolites via LC-tandem-MS and compared fluxes through the photosynthetic electron transport chain and biomass partitioning (total carbohydrates, proteins and lipids). Deletion of starch synthesis in *sta6* accounts for 20% of the biomass loss, while neither protein nor lipid content changed relative to control strains with a functional AGPase. Although the rate of photosynthetic electron transfer (PET; water  $\rightarrow CO_2$  via PSII + *cytb6f* + PSI) is greatly reduced in *sta6*, the addition of artificial electron acceptors to PSII or PSI completely restored the partial fluxes through each photosystem to control levels, confirming native functionality of both photosystems. RuBisCO protein levels were normal, while the Calvin-Benson-Basham (CBB) metabolite levels - particularly RuBP - were greatly reduced, indicating diminished substrate for carboxylation in *sta6*. Carbon metabolites in the lower branch of glycolysis accumulated greatly, notably the precursors to fatty acid synthesis (malonyl-CoA, acetyl-CoA) and proteins (glutamate and glutamine), but neither lipid nor protein content changed. Energy imbalance was found to be a key indicator of the metabolic blockage and diminished biomass in the *sta6* starchless mutant, as seen by greatly reduced reoxidation of NADPH and 3-fold larger ATP pool size.

#### 1. Introduction

Microalgal strains partition photosynthetically fixed carbon into terminal products for biosynthesis and storage, including proteins, lipids and starch in amounts that vary appreciably by genus, species and culturing conditions (Becker, 1994); (Subramanian et al., 2013); (Benamotz et al., 1985). Along with growth rate and tolerance to environmental stresses, these traits determine their suitability for biotechnological applications. The search for natural strains that accumulate desired products has been augmented by genetic approaches that aim to create transgenic lines better suited in one or more of these desired traits. For example, in Chlamydomonas reinhardtii (Chlamydomonas throughout) (Chlorophyceae), genetically blocking alternative competing pathways, including starch biosynthesis increased lipid content by 2-fold to 8-fold relative to control strains in N-deprived media (Radakovits et al., 2010; Work et al., 2010); (Wang et al., 2009); (Li et al., 2010); (Li et al., 2010). The most widely characterized Chlamydomonas starchless mutant, sta6, contains a deletion in the small subunit of ADP glucose pyrophosphorylase (AGPase) which results in higher lipid production, and has been extensively investigated (Work et al., 2010); (Goodson et al., 2011); (Li et al., 2010); (Li et al., 2010) (Siaut et al., 2011); (Fan et al., 2012); (Wang et al., 2009) (Blaby et al., 2013). Goodson and coworkers observed that following growth of stab the removal of all nitrogen and addition of acetate to the medium led to cells so engorged with lipids that they floated to the surface (Goodson et al., 2011) (Goodenough et al., 2014). However, sta6 cells have been reported to display diminished photosynthetic activity compared to WT cells under some autotrophic and mixotrophic conditions (Li et al., 2010) (Work et al., 2010). Notably, sta6 displayed an altered energy partitioning at PSII and a reduced photochemical yield attributed to lower Fv'/Fm' (chlorophyll fluorescence parameter representing the maximum quantum efficiency of PSII photochemistry under continuous illumination) and qP (approximates the proportion of PSII reaction centers that are open under a given light intensity) relative to the reference strains (Li et al., 2010), indicating that the starchless phenotype exhibits consequences

as far upstream as water oxidation by PSII. The reduced levels of water oxidation and  $CO_2$  fixation in *sta6* renders this strain undesirable from an overall bioenergy perspective. To date, identification of metabolic bottlenecks induced by the absence of starch accumulation and the precise mapping of photosynthetic electron transport (PET) chain alterations in *sta6* have not been determined and are explored in this research.

Deletion of AGPase in *sta6* eliminates the conversion of glucose 1-P to ADP-glucose, the activated substrate used in covalent glucose linkage reactions. In the majority of prokaryotes, AGPase is a homotetrameric protein, while in eukaryotes it is a heteromeric enzyme (Ballicora *et al.*, 2004) with separate small and large subunits. However, in both cases AGPase targets glucose for incorporation into carbon and energy storage polysaccharides. In *Arabidopsis*, deletion of the small subunit of AGPase led to a 33% reduction in photosynthetic  $O_2$  evolution, while loss of the large subunit resulted in a moderate decrease of 8% (Sun *et al.*, 1999). Similarly, in cyanobacterial mutants lacking AGPase, photosynthetic  $O_2$  evolution is highly inhibited (Grundel *et al.*, 2012) (Suzuki *et al.*, 2010) indicating the importance of AGPase activity not only in producing starch/glycogen but also maintaining photosynthetic activities by providing photosynthetic ATP and NADPH sinks. Herein we adopt a highly quantitative metabolomics approach to determine changes in intracellular pool sizes of key metabolites, adenylate energy charge, and pyridine nucleotide redox poise. We

combine this approach with measurements of proteins, lipids, carbohydrates, biomass, and photosynthetic electron transport to identify the metabolic and photosynthetic chokepoints in the production of energy carriers and carbon precursors to starch and lipid biosynthesis in *sta6*, and the starch accumulating control strain, cw15.

#### 2. Experimental Procedures

#### 2.1) Strains and Culture conditions:

*Chlamydomonas* strains CC-4349 *cw15* mt<sup>-</sup> (Goodenough 330A, henceforth, *cw15*) and *cw15 sta6* (BAFJ5, henceforth, *sta6*) (Wang *et al.*, 2009) (Genty *et al.*, 1989; Zabawinski *et al.*,

2001) were obtained from the *Chlamydomonas* Resource Center (University of Minnesota). cw15 was chosen as the control strain as it was earlier thought to be the clonal parent to *sta6*. During the writing of the manuscript, Blaby and co-workers (Blaby et al., 2013) showed that it is not the original parent, as they do not share the same mating type. However, cw15 was kept as the control strain in this study because it a) has an intact starch biosynthetic pathway (Blaby et al., 2013), b) analogous to *sta6* lacks a cell wall, and c) does not require arginine for growth, making it suitable for autotrophic photosynthetic studies. Liquid cultures were grown in phosphate buffered, Sueoka's high salt medium (HS) supplemented with 5mM NaHCO3 and 9.4 mM NH4Cl (Harris et al., 2009). The cultures were maintained on an orbital shaker (100 rpm) under continuous illumination (100  $\mu E m^{-2} s^{-1} PAR$ ) at 25°C. For growth rate measurements, flasks were inoculated at a density of  $\sim 6 \times 10^5$  cells mL<sup>-1</sup> and cell density was measured daily using a Neubauer counting chamber. Growth data was fitted to a Gompertz function (Zwietering et al., 1990) to calculate the specific growth rate and doubling time. For all other characterizations, cells were grown in semi-continuous cultures at a density of 8.8-11 x  $10^6$  cells mL<sup>-1</sup>. For biomass measurements, precultures were grown to ~15-20 x 10<sup>6</sup> cells mL<sup>-1</sup> then resuspended at  $2.5 \times 10^6$  cells/ml in fresh HS medium. Samples were taken immediately after resuspension (0h) and at 96 h.

#### 2.2) Chlorophyll measurements:

Chlorophyll concentration was determined spectrophotometrically by methanol extraction using extinction coefficients from Porra (Porra *et al.*, 1989).

#### 2.3) In vivo measurements of oxygen evolution:

Light saturated OER were measured using a membrane covered Clark O<sub>2</sub> (*Hansatech*) electrode at 25°C in HS media containing 5 mM NaHCO<sub>3.</sub> Samples were illuminated with red light. Light dependent respiration (LDR) was measured within 30 s post darkness and subtracted from the light dependent OER to get the gross OER. Appropriate electron acceptors and inhibitors were added directly to the sample chamber. Titration curves for PSII and PSI electron acceptors, 2, 5-dichloro-

p-benzoquinone (DCBQ) and N-dimethyl-4-nitrosoaniline (PNDA) are shown in Fig. 2-1. LDR was not subtracted while reporting data using artificial electron acceptors.



Figure 2-1 Titration of light saturated  $O_2$  evolution rate to determine the optimal electron acceptor concentration for: A) DCBQ and B) PNDA.

#### 2.4) Relative electron transport rate and non-photochemical measurements:

Electron transport rates and NPQ were measured on a pulse-amplitude modulated (PAM) fluorometer (model *FL-3000*; *Photon Systems Instruments*) (Schreiber *et al.*, 1997). Cultures were diluted to 2.5 µg Chl mL<sup>-1</sup> in HS media containing 5mM NaHCO<sub>3</sub> and incubated in the dark for 15 min before each experiment. Samples were illuminated for 10 minutes before saturating pulse was applied. Electron transport rates and NPQ were calculated using, ETR = PAR x 0.5 x 0.84 x  $\Phi_{PSII}$ , where  $\Phi_{PSII}$  is equal to  $(F_m'-F_t')/F_m'$  and NPQ =  $(F_m/F_m') - 1$  (Genty *et al.*, 1989).

#### 2.5) Fast Repetition Rate (FRR) Fluorometry measurements:

Chlorophyll-a variable fluorescence yields were measured using a home-built laser fast repetition rate (FRR) fluorometer (Ananyev *et al.*, 2005). Whole cells were concentrated to 50 µg mL<sup>-1</sup> chlorophyll, dark adapted for 120 s, then subjected to a train of 50 single turnover laser flashes (STF) (665nm, 32,000 µE m<sup>-2</sup> s<sup>-1</sup>) at either 4Hz (Low STF frequency) or 100Hz (High STF frequency). To maximize the signal to noise ratio, the pulse train was repeated 75 times with 120 s of dark incubation between consecutive experiments. Variable fluorescence yield was calculated using  $F_v=(F_m-F_o)/F_m$  (Maxwell *et al.*, 2000).

#### 2.6) Time resolved $O_2$ evolution:

Time-resolved OER were measured using a home-built bare-Pt electrode controlled by Labview software (Zakrzhevskii *et al.*, 1978) which allows kinetic resolution of the transit time to electron acceptor pools downstream of PSII (PQ, NADP<sup>+</sup>, CO<sub>2</sub>). Cells were concentrated to 20  $\mu$ g ml<sup>-1</sup> chlorophyll, 8 microliters were loaded onto the 8 mm diameter electrode, dark incubated for 15 min, and illuminated with red light (400  $\mu$ E m<sup>-2</sup> s<sup>-1</sup>). The full kinetics of O<sub>2</sub> evolution was recorded over 2 min. O<sub>2</sub> signals thus generated reflect the size of the electron acceptor pools and transit times to fill them (Zakrzhevskii *et al.*, 1978). O<sub>2</sub> peak associated with the reduction of the PQ pool was isolated by subtracting the slower filling pools by linear extrapolation to the baseline and the total O<sub>2</sub> evolved was determined from the current using Faraday's law of electrolysis.

#### 2.7) Pyridine nucleotide fluorescence kinetics:

NAD(P)H fluorescence induction kinetics were measured in whole cells with a home-built setup similar to an instrument described previously (Mi *et al.*, 2000). A UV-LED (365 nm wavelength) in combination with an optical filter (2 mm UG 11, Schott) was used as the excitation light source for NAD(P)H modulated at 100kHz. Bright red LED (660 nm, Hewlett-Packard) was used as actinic light. For fluorescence emission detection, we used a low noise photomultiplier (R2059, *Hamamatsu, Japan*) with a maximum sensitivity at 420 nm protected by a combination of a long-pass filter (KV418, *Schott*), a short-pass filter (DT Cyan), and a blue-green glass filter (BG39, *Schott*) to eliminate both the UV-measuring light and chlorophyll fluorescence. An NMR tube with an outer diameter of 5 mm was used as the sample holder and the detection and excitation pathways were perpendicular. A multifunctional DAC board PCI-6036E (*National Instruments*) was used for data acquisition.

#### 2.8) Metabolite extraction and analysis

Cultures (8.8-11 x 10<sup>6</sup> cells mL<sup>-1</sup>) were quenched in 60:40 MeOH/water mixture and centrifuged at 5000 rpm for 2 min at 0°C. The supernatant was discarded and the pellet was resuspended in icecold 60:40 methanol:water mixture. The suspension was sonicated for one min then incubated at -20°C for 20 min. Further, the samples were centrifuged for 5 min at 5000 rpm at 0°C. The supernatant was vacuum dried (Centri Vap Benchtop Vacuum Concentrator, *Labconoco*) and the resulting pellet was resuspended in LC-MS grade water. Samples were injected into an *Agilent* 6490 QQQmass analyzer coupled to HPLC (*Agilent Technologies*) for metabolite analysis (Kenchappa *et al.*, 2013). Concentrations of each metabolite per cell were calculated using the calibration curve. Fold changes were calculated as a ratio of concentration of metabolites in low light and high light conditions, respectively.

#### 2.9) Lipid, glucose and protein analysis

Total lipids were converted to fatty acids for GC-FID analysis as described previously (Radakovits *et al.*, 2012) and analyzed using an Agilent 7890A gas chromatograph and DB5ms column with flame ionization detection. Total sugar was analyzed using the anthrone assay (Morris, 1948) (Meuser *et al.*, 2012). Protein levels were determined using the DC protein assay kit (Bio-Rad) according to the manufacturer's instructions.

#### 3. Results

#### 3.1) Photoautotrophic growth and biomass accumulation

Fig. 2-2 A compares sta6 and cw15 growth rates in photoautotrophic HS medium supplemented with 5 mM NaHCO3, illuminated using 100  $\mu$ E m-2s-1 (white light). The *sta6* mutant grew slower relative to cw15, with 30% longer doubling time (p= 0.03). Both strains reached stationary phase simultaneously, but sta6 had a ~1/3 lower cell density relative to the control strain. As the cultures grew, the light penetration decreased because of self-shading with transmitted intensity dropping as low as 25  $\mu$ E m-2s-1 in stationary phase cultures.



Figure 2-2 **Photoautotrophic growth phenotypes for** cw15 and sta6. Growth rates measured under photoautotrophic conditions. The inset shows the doubling time based on fits to Gompertz curves. The asterisk indicates a statistically significant difference compared with cw15 (P<0.05). B) Biomass composition of cw15 and sta6 under N-replete photoautotrophic conditions measured after 4 days of growth. The y axis represents the dry weight in grams. Values are means and standard errors of three biological replicates.

At stationary phase (after 96 h of growth), the three major biochemical fractions (total reducing carbohydrate, lipids and proteins) were quantified in *sta6* and *cw15* (Fig. 2-2 B and Table 2-1). In addition, we also measured the biomass distribution in two other starch synthesizing reference strains, CC124 and D66 to assess biomass distribution across multiple control strains. The dry cell weights (DCWs) of the *sta6* mutant cultures were ~20-35% lower than the DCWs of all the three reference strains. Total reducing carbohydrate (TRC) content in *sta6* was between 2-5 times lower than the control strains. Additionally, *sta6* has slightly increased protein content relative to D66 and CC124, however, it is not statistically different when compared to *cw15*.

Biomass		15		
(HS medium)	stao	cw15	CC124	D66
DW (0hr)	$0.13\pm0.04$	$0.07\pm0.00$	$0.18\pm0.03$	$0.12\pm0.04$
DW (96hr)	$1.08\pm0.05$	$1.34\pm0.13$	$1.29\pm0.11$	$1.49\pm0.06$
DW Flux (gl <sup>-1</sup> d <sup>-1</sup> )	$0.23\pm0.02$	$0.32\pm0.02$	$0.27\pm0.02$	$0.34\pm0.03$
Lipid (96hr)	$0.072\pm0.016$	$0.059\pm0.010$	$0.106\pm0.008$	$0.109\pm0.023$
%Lipid	$6.6\pm1.2$	$4.4\pm0.3$	$8.3\pm1.2$	$7.3\pm1.5$
Total reducing				
Carbohydrate (TRC)	$0.106\pm0.006$	$0.428\pm0.072$	$0.246\pm0.029$	$0.493 \pm 0.142$
(96hr)				
%TRC	$9.88 \pm 0.14$	$31.8\pm2.4$	$19.1\pm1.2$	$33.0\pm8.8$
Protein (96hr)	$0.688 \pm 0.153$	$0.761\pm0.293$	$0.645\pm0.109$	$0.702\pm0.083$
%Protein	$62.4 \pm 18.0$	$56.1 \pm 16.5$	$49.9\pm7.2$	$47.5\pm7.3$
Mass balance	$80.5\pm16.3$	$92.2\pm19.2$	$77.2\pm8.7$	$87.8\pm 6.0$

Table 2-1 Biomass composition of *sta6* and three reference strains, *cw15*, CC124 and D66, under NH4Cl replete, photoautotrophic culturing conditions. Values are reported as g L-1 and data represent three independent biological replicates.
These results indicate that the loss of starch biosynthesis, does not result in the repartitioning of carbon from TRC to lipids and proteins under nutrient replete conditions; instead cellular biomass is reduced by an amount that is similar to the attenuated TRC levels in *sta6* relative to the control strains. Diminished biomass accumulation implies that a primary adaptation in this starchless mutant is diminished/altered photosynthetic electron transport.

3.2) Photosynthetic oxygen evolution rate, PSII quantum yield of charge separation and electron transport rate

To quantify the effect of a lack of starch biosynthesis on photosynthesis; photosynthetic O<sub>2</sub> evolution rates (OER) were measured for *sta6* and *cw15* with a Clark electrode and illuminated with a range of light intensities ( $\lambda_{max} = 620$  nm). At light levels up to 60 µE m<sup>-2</sup>s<sup>-1</sup>, *sta6* and *cw15* have comparable OER but as the light intensity increased above this threshold, the differences between *cw15* and *sta6* increased and at a light intensity of 680 µE m<sup>-2</sup>s<sup>-1</sup>, *sta6* shows a 3.5-fold lower OER than *cw15* (Fig. 2-3A). It important to note that cultures were grown using 60-80 µE m<sup>-2</sup>s<sup>-1</sup> of white light, and then OER assayed across a range of light intensities. The lower OER in *sta6* (Fig. 2-3A), indicates that either PSII itself is impaired and/or that reactions downstream of PSII are affected at light intensities >60 µE m<sup>-2</sup>s<sup>-1</sup>. To differentiate between the two processes we probed both the light-induced electron transport rate (ETR) through PSII and the quantum yield of PSII charge separation and the turnover efficiency of the water-oxidizing complex (WOC).



Figure 2-3 **Oxygen, electron transport rate and NPQ**. A ) Oxygen evolution rates at different light intensities (16, 30, 61, 100, 150, 340 and 680  $\mu$ E m<sup>-2</sup> s<sup>-1</sup>) measured (at t = 6 T<sub>d</sub> Figure 2-2 A), and B) Relative electron transport rate (ETR) and C) NPQ measured by PAM fluorometer as a function of increasing actinic light intensities. Data represent the mean of three biological replicates with standard error.

Light-induced electron transport rate (ETR) through PSII was measured using a pulse amplitude modulated (PAM) fluorometer (Schreiber et al., 1997) (Fig. 2-3 B). This method reports on the fraction of open PSII centers capable of photoreducing  $Q_A$  as a function of the light intensity of concurrent background illumination. Although widely used to measure ETR (flux), this signal is a static measure of Q<sub>A</sub> population. It is proportional to the sum of ETR flux and photochemical quenching that does not lead to electron transfer beyond PSII. For this reason, the PAM type measurement of ETR includes non-flux contributions to photochemical quenching within PSII that produce no  $O_2$ . The ETR curves show a trend qualitatively similar to OER, with *sta6* and *cw15* showing no significant differences at the lowest light intensities, but with differences between sta6 and cw15 emerging at ~75 µE m<sup>-2</sup>s<sup>-1</sup>. sta6 shows substantially lower ETRs at light intensities above 200  $\mu$ E m<sup>-2</sup>s<sup>-1</sup>. At the maximum light intensity used for the measurement (~ 400  $\mu$ E m<sup>-2</sup>s<sup>-1</sup>) ETR in sta6 is ~2-fold lower than cw15. ETR in sta6 also saturates at a lower light intensity (200  $\mu E$  m<sup>-2</sup>s<sup>-1</sup>). The % differences observed in OER are larger than the differences in ETR (Fig. 2-4) above light intensities of 100  $\mu$ E m<sup>-2</sup>s<sup>-1</sup>. This may be attributed to alternate oxygen consuming pathways (also see section 2.3). To investigate the possibility of non-photochemical quenching (NPQ)-dependent decrease in ETR, NPQ was quantified simultaneously with ETR using (Schreiber et al., 1997) and calculated as given in "Experimental Procedures". sta6 and cw15 do not show significant differences in NPQ (Fig. 2-3 C), thus the difference in ETR are not due to differences in excitation quenching in the antenna.



Figure 2-4 **Discrepancy between OER and ETR.** % difference between OER and ETR between *cw15* and sta6 calculated as  $(cw15 \cdot sta6)/(cw15) \times 100$  at light intensities between 16-400 µE m<sup>-2</sup> s<sup>-1</sup>.



Figure 2-5 *sta6* is limited by electron acceptor. A) Average chl variable fluorescence yield (Fv/Fm) from PSII produced by 50 single turnover pulses at two flash rates, 4 Hz and 100 Hz. The average Kok hit parameter,  $\gamma_{avg}$ , was determined by fitting the data to the VZAD model.  $\gamma_{avg} = 1$ -  $\alpha$ (miss) -  $\beta$ (double hit) -  $\delta$ (backward) -  $\varepsilon$ (inactivated). Cycle period, P, was determined by Fourier transformation. The shaded region represents the reduced PQ pool. B) OER of *cw15* and *sta6* in the presence of inhibitors to alternate electron flow pathway. Control has 5 mM bicarbonate, SHAM (20  $\mu$ M) inhibits AOX, PGAL (2 mM) inhibits PTOX, Myxothiazol (2  $\mu$ M) inhibits respiration. C) Light saturated O<sub>2</sub>-evolution rates in the presence of artificial electron acceptors (40  $\mu$ M DCBQ or 400  $\mu$ M PNDA or 5mM NaHCO<sub>3</sub>) Data represent the mean of three biological replicates with standard error.

To examine if an impaired PSII-WOC activity is the cause of the low OER and ETR phenotype, the quantum yield of PSII charge separation and flux through the WOC were measured using Chl variable fluorescence emission (Fv/Fm) with a fast repetition rate fluorometer (FRRF) at two flash frequencies (Ananyev et al., 2005). Compared to cw15, sta6 exhibits slightly lower steady-state dark-adapted (F<sub>v</sub>/F<sub>m</sub>) under both low (4 Hz, <F<sub>v</sub>/F<sub>m</sub>>  $\approx$  0.43-0.41) and high (100 Hz, <F<sub>v</sub>/F<sub>m</sub>>  $\approx$ 0.24 -0.18) flash frequencies (Fig.2-5 A). Following an initial dark adaptation, the rate of damping of the period-four oscillations in  $F_v/F_m$  was similar for both strains and decreased at higher flash rates (Fig. 2-5 A), indicating both strains have normal initial WOC functioning. Fits of the oscillations to an advanced Kok model (VZAD algorithm) (Vinyard et al., 2013) produce almost identical hit parameter ( $\gamma_{avg}$ ) and period (P) at both flash frequencies, indicating no significant difference in the initial turnover probability of the WOC in the two strains. However, the initial rate of damping of individual oscillations in  $F_v/F_m$  drops off rapidly with higher flash rate ( $\gamma_{avg}$  dropping to 0.57 at 100 Hz, Fig. 2-5 A) and the non-oscillating portion of Fv/Fm in sta6 is missing the portion attributed to filling of the PQ pool (hatched area in Fig. 2-5 A) (Kolling et al., 2009). Thus, sta6 is characterized by a slower rate of emptying the reduced plastoquinol  $PQH_2$  pool relative to cw15.

Thus, both strains have a normal functioning WOC and PSII charge separation, but *sta6* has an inhibited PQH<sub>2</sub> oxidation rate, reflecting a blockage in electron flow downstream of the PQH<sub>2</sub> pool. This blockage is responsible for the sharp break in OER flux above light intensity > 100  $\mu$ E m<sup>-2</sup>s<sup>-1</sup> (Fig. 2-3A).

## 3.3) Alternative oxygen consuming pathways

As previously mentioned, at higher light intensities there is an imbalance in the extent of inhibition reported by OER vs. ETR, with OER predicting a lower intensity threshold and larger activity gap at saturation between *sta6* and *cw15* (Fig. 2-4). This difference indicates the presence of alternative  $O_2$  consuming pathways that influence OER and ETR differently at light intensities above 100 µE

m<sup>-2</sup>s<sup>-1</sup>. The main pathways that consume O<sub>2</sub> are respiration and alternative oxidases (AOX) in the mitochondrion, and plastoquinol terminal oxidase (PTOX), Mehler reaction and photorespiration in the chloroplast. Inhibition of AOX and PTOX (also known as chlororespiration) by specific inhibitors propyl gallate (PGAL, 2mM) and salicylhydroxamic acid (SHAM, 20µM) did not have any significant effect on the OER (at 680 µE m<sup>-2</sup>s<sup>-1</sup>) in both *cw15* and *sta6* (Fig. 2-5 B). However, inhibition of mitochondrial respiration by myxothiazol (inhibiting the cytochrome bc1 complex of the respiratory chain) leads to a 70% loss of OER in *sta6*, while only minor loss (10%) in *cw15* (Fig. 2-5B) (at 680 µE m<sup>-2</sup>s<sup>-1</sup>). Thus, only in *sta6* does mitochondrial respiration act as an important shunt to partially consume the excess light-induced reductant generated in the chloroplast. Further measurements of Mehler reaction and photorespiration were not performed, though Fig. 2-4 clearly indicates their possible contributions to reductant consumption.

# 3.4) Normal PET rates through PSII and PSI can be restored in sta6.

The foregoing results suggest that under high light intensities, photosynthesis in *sta6* is limited by the reactions occurring downstream of PSII, either within the electron transport chain itself (light dependent reactions) or by the carbon fixation reactions (light independent reactions). To distinguish between these options, artificial electron acceptors were used to alleviate choke points in the PET chain and the corresponding OER was measured at 680  $\mu$ E m<sup>-2</sup>s<sup>-1</sup>. As shown in Fig. 2-5 C, in the presence of 2, 5-dichloro-p-benzoquinone (DCBQ) (40  $\mu$ M), which accepts electrons from the Q<sub>B</sub> site of PSII, the OER of both *sta6* and *cw15* increased to an approximately equal level of ~200  $\mu$ mol O<sub>2</sub> mg Chl<sup>-1</sup> hr<sup>-1</sup>. With N-dimethyl-4-nitrosoaniline (PNDA) (400  $\mu$ M), which accepts electrons from PSI via ferredoxin (FDX), both strains had an equal OER of ~110  $\mu$ mol O<sub>2</sub> mg Chl<sup>-1</sup> hr<sup>-1</sup>. This experiment reveals that *sta6* has a structurally intact electron transport chain similar to *cw15* through FDX, with normal fluxes and similar electron acceptor pool sizes (for PQ and Fdx). Bicarbonate addition to ensure saturating levels for carbon fixation reactions (Fig. 2-5 C), did not reverse the disparity in OER between cw15 and sta6. Thus, the OER in sta6 is limited downstream of FDX by carbon fixation reactions.

To isolate the location of flux chokepoints, we measured the light induced yield of  $O_2$  from a thin solution of cells (0.8 mm) using a bare platinum (rate) electrode illuminated at 400 µE m<sup>-2</sup>s<sup>-1</sup>. This method resolves kinetic features of the electron acceptor pools downstream of PSII (Zakrzhevskii *et al.*, 1978). A typical  $O_2$  evolution trace after dark adaptation displays three distinct time-resolved features. The first is a sharp transient peak from reduction of the PQ pool, the second is a broad peak arising from the reduction of NADP<sup>+</sup>, the third is a continuous slope from CO<sub>2</sub>-dependent  $O_2$  evolution. The initial peak has a rise time of 300 ms due to  $O_2$  diffusion to the electrode and is determined by the finite thickness of the sample. The area under this peak (see experimental procedure) is directly proportional to the difference in electron fluxes filling and emptying the oxidized PQ pool. The PQ peak intensity and area are indistinguishable for *sta6* and *cw15*. In *cw15* a second  $O_2$  feature appears as an inflection (Fig. 2-6 top), but is masked by a much larger continuous slope that disappears when CO<sub>2</sub> is removed (not shown) or when the PET inhibitor 3-(3,4-dichlorophenyl)-1,1-dimethylurea (DCMU) is added (Fig. 2-6 bottom). This CO<sub>2</sub>-dependent water oxidation rate is 7-fold greater in *cw15* than in *sta6* over the indicated illumination period (400 µE m<sup>-2</sup>s<sup>-1</sup>).



Figure 2-6 **Time resolved oxygen evolution rate.** Light saturated OER on a bare-Pt electrode recorded as nanoAmperes ( $\mu$ g Chl)<sup>-1</sup> in the presence of bicarbonate as the sole electron acceptor. Data represent the average of three biological replicates. Light was turned on or off at points indicated by solid arrows. Transient O<sub>2</sub> evolution shows two distinct phases, spike 1 representing the PQ pool reduction while the continuous slope representing carbon fixation (top panel). The inset shows the PQ pool size equivalent calculated as indicated in experimental procedures. Bottom panel shows the effect of 10 µM DCMU on the OER. Dotted line represents the dark current recorded on the electrode in the absence of illumination.

To quantify the transient flux into and out of the NADP(H) pool the relative light-induced yield of the reaction: PSI donor  $\rightarrow$ NADP<sup>+</sup> was monitored by the fluorescence change from reduced NAD(P)H *in vivo*. NAD(P)H kinetics in whole cells was measured using a home-built fluorometer (excitation at 365 nm, detection at 420 nm) similar to (Mi *et al.*, 2000). Samples were exposed to a dark-light-dark regime of 5-4-7 min using saturating light (400  $\mu$ E m<sup>-2</sup>s<sup>-1</sup>, 660 nm). An average curve from three biological replicates for *sta6* and two for *cw15* is shown in Fig. 2-7 A. The rise and decay times and the amplitude changes of the fluorescence yield are given in Table 2-2. Light induced changes in NAD(P)H fluorescence display dynamics similar to previous reports (Cerovic *et al.*, 1993; Mi *et al.*, 2000). We observe a rise in NADPH fluorescence emission during actinic illumination, signifying light-driven reduction of NADP<sup>+</sup>, followed by a steady-state phase where the NADP<sup>+</sup> reduction and NAD(P)H oxidation rates are matched. Post illumination, in darkness the fluorescence decays followed by recovery to the initial steady-state dark level. In the presence of the artificial PSI electron acceptor PNDA, both *sta6* and *cw15* produce no change in NAD(P)H fluorescence upon illumination.



Figure 2-7 **NADPH fluorescence and photoinhibition**. A) Effect of actinic illumination on induction of continuous level of blue-green fluorescence arising from NAD(P)H. Actinic illumination was provided by saturating red light (660 nm, Intensity =400  $\mu$ E m<sup>-2</sup> s<sup>-1</sup>). *sta6* has been offset by 3 units in both the figures. Fluorescence decay during illumination (lifetime denoted by t<sub>2</sub> in *sta6*) was fitted to a linear decay slope and extrapolated to "light on" time to calculate the maximal photoreduced NADPH (Y1) in the absence of oxidation of NADPH. Y2 denotes the oxidized NADPH post illumination. t<sub>1</sub>, t<sub>2</sub>, t<sub>3</sub>, t<sub>4</sub> are the rise and decay times as described in the text. Magnification of the decay kinetic is given in the inset. For both A and B, light was turned on or off at points indicated by solid arrows B) Photoinhibition at 400  $\mu$ E m<sup>-2</sup> s<sup>-1</sup>. t<sub>1/2</sub> for photoinhibition is given in the inset table. Data represent the mean of three biological replicates with standard error.

	cw15	sta6	Fold Change		
			(sta6/cw15)		
Rise time, t <sub>1</sub> (s)	7.2 (0.3)	7.2 (0.42)	1.0 (0.07)		
Reoxidation time, t <sub>2</sub> (min)	$\leq t_1$	2.9 (0.67)			
Decay time, $t_3$ (s)	14 (0.6)	20 (2.4)	1.4 (0.12)		
Re-reduction rate, $t_4$ (s)	39 (0.6)	-			
Maximum photoreduced NADPH , $Y_1$ (rel. u.)	1.5 (0.04)	2.6 (0.3)	1.7 (0.12)		
Oxidized NADPH post illumination, $Y_2$ (rel. u)	2.8 (0.3)	1.9 (0.2)	0.68 (0.15)		

Table 2-2 **Kinetics of NADP+ reduction and NADPH reoxidation.** Values represent the average of three biological replicates and standard error of mean is given in parentheses. The NAD(P)H fluorescence yield rises from the dark adapted level to the maximal photoreduced level in identical times for both strains ( $t_1$ = 7.2 sec). During the illumination period the larger photoinduced amplitude in *sta6* decays slowly to 50% with a lifetime of ~2.9 minutes ( $t_2$ ), while the lower photoinduced amplitude for *cw15* does not decay appreciably on the measurement timescale. The larger amplitude and gradual decaying slope of light-induced NADPH in *sta6* is attributed to slower NADPH reoxidation. This was verified by turning off the actinic light and monitoring the decay rate of the NADPH fluorescence amplitude ( $Y_2$ ), which is indeed slower in *sta6* ( $t_3$  = 20 sec) than *cw15* ( $t_3$  = 14 sec) and "overshoots" to a level lower than the initial dark steady-state. The  $t_3$  decay rate in *cw15* is followed by recovery to the initial steady-state level within 39 sec ( $t_4$ ) as the non-equilibrium level of oxidized NADP+ is (re)reduced in the dark. Such oscillating behavior is commonly associated with regenerative cycles. This latter phase is absent in *sta6*. Taken together, the lack of overshoot, slower  $t_3$  decay, and slow decay of the actinic phase ( $t_2$ ) all support the conclusion of a slower reoxidation rate of actinically generated NADPH in *sta6*. For *cw15*, the lack of observable decay of the actinically generated NADPH phase indicates a rate of NADPH reoxidation faster than the time resolution of the instrument (~1 sec).

#### 3.5) sta6 is more sensitive to photoinhibition

Sensitivity to photoinhibition upon prolonged high light exposure was determined by measuring the loss of PSII quantum yield (as Fv/Fm) in samples incubated under light (400  $\mu$ E m<sup>-2</sup> s<sup>-1</sup>, 620 nm) over a period of 6 h (Fig. 2-7 B). *sta6* exhibits a 35% faster photoinhibition rate (halftime = 0.28 h vs. 0.43 h) than in *cw15*, indicating the importance of starch biosynthesis for tolerance to high light stress.

## 3.6) sta6 is inefficient in draining upper glycolytic intermediates at high light intensity

Given that OER and ETR are significantly inhibited at high light intensities and the metabolic block occurs downstream of the PET, comparative metabolic analysis of low (LL) and high light (HL) treated samples was performed to reveal the metabolic consequences of the decrease in NADPH reoxidation. The pool sizes of the various metabolites involved in the central carbon metabolism were measured using LC-MS/MS (Bennette *et al.*, 2011) under ambient growing conditions (~60-80  $\mu$ E m<sup>-2</sup>s<sup>-1</sup>, low light) and after a 2 h exposure to high light (400  $\mu$ E m<sup>-2</sup>s<sup>-1</sup>). Concentrations of the metabolites are presented in Table 2-3. Metabolites pertaining to the CBB cycle and upper glycolysis are summarized graphically in Fig. 2-8 A and 2-8 B. Simultaneously, the biomass contents (lipids, TRC and protein) were also measured (Fig. 2-8 C) using the same cultures.

Upon transfer to HL, neither protein nor lipids showed a significant change in the two strains. However, the TRC content per cell increased by 25% in *cw15* while being unchanged in *sta6* (Fig. 2-8 C). Thus, under 2 h exposure to HL, fixed carbon predominantly assimilates as starch, while blocking starch biosynthesis at AGPase in *sta6*, carbon assimilation into all biopolymers is impeded.

Metabolite analysis revealed that under HL, *cw15* shows an increase in the CBB cycle intermediates including RuBP (5-fold), R5P (4-fold) and X5P (9-fold) compared to LL. GAP and 3PG levels also increase by about 2-fold (Fig 2-8 A, Table 2-3). This increase correlates with the enhanced carbon

assimilation in the *cw15* (Fig. 2-8 C). Even though the TRC content increases in *cw15* (Fig. 2-8 C), there is no increase in the pool size of the upper glycolytic intermediates indicating an efficient drain of the upper glycolytic metabolites into carbohydrate biosynthesis. By contrast in *sta6*, we observe large increases in levels of many CBB cycle metabolites, especially GAP and 3PG levels increase by 11-fold and 7-fold respectively. Additionally, we observe enhanced levels of the upper glycolytic intermediates G1P, G6P and F6P by 5-fold, 4-fold and 8-fold respectively, even though there is no significant increase in the TRC content. Thus, under high light conditions and in the absence of AGPase, there is an inefficient drain of upper glycolytic intermediates that normally flow into starch biosynthesis.

A lower ATP recycling rate can limit electron transport chain flux (Sharkey *et al.*, 1986). To test this, the adenine nucleotide concentrations were quantified using LC-MS and the cellular energy charge (CEC) was calculated (Table 2-3). *sta6* exhibits appreciably larger concentrations of all adenine nucleotides (ATP, ADP and AMP) between 1.5- 2.8 fold (ATP) higher in *sta6* relative to *cw15* under both LL and HL. Additionally, *sta6* has a marginally larger pool size of the phosphorylated pyridine nucleotides, 1.3 (NADP<sup>+</sup> + NADPH) -fold larger relative to *cw15* (Table 2-3). This increase in both NADP(H) and adenylates likely reflects the cellular energy charge did not differ in the two strains in either of the light regimes, the method we used may not be rapid enough to prevent equilibration. Stable CEC is known to be essential for normal cellular function (Kramer *et al.*, 2011).

Metabolites	Abbrev.	<i>cw15</i>	cw15	sta6	sta6	Fold Change-	Fold Change-	p-value (cw15)	p-value
		<b>SS</b> <sup>(1)</sup>	$HL^{(2)}$	$SS^{(1)}$	$HL^{(2)}$	<i>cw15</i> ( <i>HL</i> /SS) <sup>(3)</sup>	sta6 (HL/SS) <sup>(3)</sup>	(4)	$(sta6)^{(4)}$
Glycolysis and CBB intermediates									
ADP-Glucose	ADP-	0.019	0.010	0.001	0.001	0.515	0.675	0.043	0 470
	glc	(0.003)	(0.001)	(0.001)	(0.000)	(0.102)	(0.311)	0.075	0.770
	CID	0.115	0.086	0.063	0.357	0.745	5.63	0 179	0.022
Glucose-1-phosphate	GIP	(0.006)	(0.017)	(0.004)	(0.091)	(0.152)	(1.19)	0.177	0.022
Character Carls and a to	CCD	0.101	0.076	0.064	0.291	0.753	4.53	0 373	0.046
Glucose-6-phosphate	GOP	(0.001)	(0.025)	(0.004)	(0.104)	(0.246)	(1.14)	0.575	0.040
		0.058	0.055	0.038	0.313	0.950	8.27	0 927	0.050
Fructose-6-phosphate	FOP	(0.006)	(0.029)	(0.001)	(0.118)	(0.509)	(1.63)	0.727	0.050
Glyceraldehyde		0.077	0.195	0.040	0.474	2.51	11.8	0.059	0.041
phosphate	GAP	(0.002)	(0.048)	(0.002)	(0.183)	(0.626)	(2.58)	0.057	0.071
	3PG	3.00	5.69	0.386	2.87	1.89	7.45	0.030	0.035
3-Phosphoglycerate		(0.091)	(0.812)	(0.058)	(1.091)	(0.277)	(1.04)	0.050	0.055
	PEP	0.736	1.27	0.160	0.702	1.72	4.38	0.067	0.110
Phosphoenolpyruvate		(0.041)	(0.210)	(0.014)	(0.264)	(0.302)	(1.695)	0.007	0.110
	AcCoA	0.116	0.231	0.205	0.119	1.99	0.581	0.018	0 125
Acetyl-CoA		(0.003)	(0.030)	(0.014)	(0.042)	(0.263)	(0.209)	0.010	0.125
Ribulose-1,5-bis-	חחוח	0.082	0.337	0.019	0.099	4.10	5.10	0.000	0.038
phosphate	KUBP	(0.011)	(0.019)	(0.002)	(0.037)	(0.583)	(0.906)	0.000	0.050
	D 5D	0.012	0.039	0.013	0.046	3.18	3.62	0.015	0 108
Ribose-5-phosphate	КЭР	(0.002)	(0.006)	(0.002)	(0.016)	(0.680)	(1.37)	0.015	0.100
Valess 5 alessalests	X5P	0.015	0.071	0.010	0.091	4.77	9.34	0 105	0.059
Xylose-5-phosphate		(0.003)	(0.026)	(0.002)	(0.031)	(2.08)	(3.76)	0.105	0.057
Sedoheptulose-7-	070	0.074	0.575	0.061	0.507	<u>9 21 (2 6)</u>	7.74 (2.01)	0.063	0.027
phosphate S/	5/P	(0.022)	(0.145)	(0.017)	(0.174)	0.21 (3.0)	7.74 (3.01)	0.005	0.027
Lipid Precursor									
Malanyl Co A	MaCaA	0.001	0.001	0.018	0.028	0.968	1.59	0.9195	0.026
Maionyi CoA	MaCOA	(0.000)	(0.000)	(0.000)	(0.002)	(0.296)	(0.071)		

	1		1						
	Abbrev							p-value (cw15)	p-value
Metabolites	AUDIEV.	cw15	<i>cw15</i>	sta6	sta6	Fold Change-	Fold Change-	(4)	$(sta6)^{(4)}$
		<b>SS</b> <sup>(1)</sup>	$HL^{(2)}$	<b>SS</b> <sup>(1)</sup>	$HL^{(2)}$	<i>cw15</i> ( <i>HL</i> /SS) <sup>(3)</sup>	sta6 (HL/SS) <sup>(3)</sup>		(*****)
TCA cycle and Amino acids									
Succinate	<u>ettee</u>	0.712	1.49	0.304	0.594	2.09	1.95	0.006	0.161
	SULL	(0.058)	(0.131)	(0.044)	(0.164)	(0.251)	(0.608)		
	NAAT	1.86	2.63	1.17	1.67	1.41	1.42	0.028	0.360
Malate	MAL	(0.108)	(0.201)	(0.152)	(0.456)	(0.136)	(0.429)		
. Vata alasta vata	AVC	0.452	1.13	0.169	0.343	2.50	2.03	0.112	0.159
α-Ketoglutarate	AKG	(0.017)	(0.334)	(0.030)	(0.096)	(0.746)	(0.675)		01107
01	CL LI	3.32	5.80	2.28	5.35	1.44	2.34	0.004	0.077
Glutamate	GLU	(0.177)	(0.378)	(0.106)	(1.29)	(0.147)	(0.578)	01007	0.077
Classica in a	GLN	0.046	0.062	0.082	0.057	1.34	2.08	0.003	0.056
Glutamine		(0.002)	(0.001)	(0.019)	(0.018)	(0.071)	(0.276)		
<u>Sugars</u>									
q	SUCR	0.011	0.014	0.007	0.044	1.20	6.10	0.598	0.062
Sucrose		(0.004)	(0.002)	(0.001)	(0.017)	(0.421)	(2.488)	0.070	0.002
Energy Carriers									
	AMP	1.66	1.60	2.09	2.67	2.16	2.71	0.137	0.217
AMP		(0.171)	(0.830)	(0.521)	(1.19)	(0.656)	(1.329)		
	ADP	2.20	2.84	3.43	3.98	1.28	1.15	0.549	0.717
ADF		(0.206)	(0.363)	(0.396)	(2.23)	(0.231)	(0.572)		
ΛТD	ATP	3.40	3.92	6.48	6.21	1.15	0.958	0.923	0.727
AIP		(0.183)	(0.826)	(1.35)	(1.21)	(0.256)	(0.285)		
<u>Cellular energy charge</u> (calculated)									
$=\frac{ATP + (0.5 * ADP)}{ATP + ADP + AMP}$	CEC	0.682 (0.060)	0.637 (0.023)	0.620 (0.022)	0.638 (0.076)				

Metabolites	Abbrev.	cw15 $SS^{(1)}$	<i>cw15</i> HL <sup>(2)</sup>	sta6 SS <sup>(1)</sup>	sta6 HL <sup>(2)</sup>	Fold Change- cw15 (HL/SS) <sup>(3)</sup>	Fold Change- sta6 (HL/SS) <sup>(3)</sup>	<i>p-value</i> ( <i>cw15</i> ) (4)	<i>p-value</i> ( <i>sta</i> 6) <sup>(4)</sup>
Pyridine nucleotides									
(NAD <sup>+</sup> + NADH)		1.92 (0.043)	3.06 (0.286)	2.74 (0.118)	2.348 (0.670)	1.43 (0.048)	0.767 (0.300)	0.454	0.752
(NADP <sup>+</sup> + NADPH)		0.636 (0.052)	0.873 (0.073)	0.781 (0.027)	1.13 (0.262)	1.22 (0.088)	1.29 (0.246)	0.213	0.783

Table 2-3 LC-MS/MS results for various metabolites. Standard error of mean are represented in parentheses. (1)(2) Represent the concentration of the metabolites in *sta6* and *cw15* under low light conditions(1) (steady state) and high light conditions(2) and calculated as  $\mu$ moles per 10<sup>8</sup> cells. (3)Represents the fold change in concentration of the metabolites between LL and HL in *sta6* and *cw15* calculated as [HL]/[LL]. (4) Represents the p-value



Figure 2-8 **Metabolite and biomass distribution**. Summary of metabolomics data for CBB cycle and gluconeogenic pathway. A) The fold changes in the metabolites occurring during low light to high light transition in *sta6* and *cw15*. Dotted line in A represents no-fold change B) Schematic representation of the fold changes of the metabolites given in A. C) Biomass distribution (Total reducing carbohydrates, TRC; lipids and proteins) during the low light to high light shift. White bars represent the biomass content under low light condition and grey bars represent the biomass after HL shift. Data represent the mean of three biological replicates with standard error.

#### 3.7) Photosynthetically fixed carbon in sta6 is redirected towards lower glycolysis

Metabolites of central carbon metabolism outside of the CBB cycle involved in biosynthesis of amino acids, fatty acid and alternative CO<sub>2</sub> fixation (C4) were quantified and compared across the two strains (Table 2-3). In *sta6*, the high energy phosphate ester, phosphoenolpyruvate (PEP) increased by 4-fold, further signaling the accumulation of excess cellular phosphate and energy. Amongst the TCA cycle metabolites, succinate and  $\alpha$ -ketoglutarate increased uniformly in both strains. Although *sta6* does not accumulate protein, the amino acid precursors glutamine and glutamate, increased both at LL and HL intensities compared to *cw15*. Significantly, from the perspective of lipid biosynthesis, the level of the fatty acid precursor malonyl-CoA is 18-fold higher in *sta6* compared to *cw15*, and increased further to 28-fold upon high light shift, although the total lipid content in *sta6* was indistinguishable from *cw15*.

## 4. Discussion

The use of various starchless mutants as test strains in search of higher lipid producers has been investigated in multiple prior studies (Work et al., 2010; Goodson et al., 2011; Li et al., 2010a; Li et al., 2010b; Siaut et al., 2011; Fan et al., 2012; Wang et al., 2009; Blaby et al., 2013; Goodenough et al., 2014). These studies also found that loss of starch synthesis and lower growth rates were strongly correlated, but the consequences on the individual PET reactions and carbon partitioning via pathways and into terminal products had not been examined systematically at the molecular level until now. In Scheme 2-1 we summarize our data describing the molecular phenotype arising from loss of AGPase in the *Chlamydomonas* starchless mutant, *sta6*.



Scheme 2-1 Schematic representation of metabolism in *sta6*. Schematic representation of light dependent and independent reactions of photosynthesis in the *sta6* strain (deletion in the small subunit of AGPase, denoted by X), while cw15 is the control strain. Thick and thin arrows represent large and small electron fluxes through WOC, PET and the CBB cycle, with *sta6* having a 2-fold slower electron flux than cw15. Small and large font sizes indicate (qualitative) differences in relative contents. Circle areas are quantitatively proportional to the total pool sizes in the cell (dashed circle for cw15, solid circles for *sta6*). Dashed circle in the sta6 panel shows the pool size of the metabolite in cw15 super imposed onto *sta6* pool size indicating that sta6 has a larger pool size of NADP(H) (1.2-fold) and adenylates (1.6-fold) vs cw15, but similar plastoquinone pool sizes. The ratio of NADPH/NADP+ and plastoquinol/plastoquinone pool sizes is qualitatively represented by the dark and light gray colors.

Starch (or glycogen in cyanobacteria) biosynthesis is essential for (at least) four purposes: a) consuming the photosynthetically generated reductant (NADPH) and ATP (Kramer and Evans, 2011), b) providing the gluconeogenic route needed for replenishing the CBB cycle intermediates, c) as a low osmotic potential storage product for carbon and energy need during dark, and d) photoprotection against oxidative damage. In the absence of starch biosynthesis arising from a nonfunctional AGPase, all of these functions are disrupted. Especially at high light, starting with the primary impact on the rate of  $CO_2$ -dependent water oxidation, sta6 has a 3.5-fold slower rate than cw15 under the conditions in Fig. 2-3 A. This leads to a buildup in the reductant poise (NADPH/NADP<sup>+</sup>), the extent of carbohydrate phosphorylation, and the adenylate nucleotides, the latter presumably in an attempt to compensate for the slower turnover rate (Scheme 2-1). Accumulation of several phosphorylated carbohydrates (Table 2-3) (Scheme 2-1) and the upper glycolysis intermediates (G1P, G6P and F6P) occurs which do not drain efficiently to terminal sinks. This outcome together with the 2.8 fold larger adenylate pool size will result in a corresponding decrease in the available free  $Mg^{2+}$  in the stroma of the chloroplast. As  $Mg^{2+}$  is a required activator for RuBisCO and other CCB enzymes, allosteric down regulation of RuBisCOdependent  $CO_2$  fixation activity is expected in *stab* if not offset. This down regulation appears to be directly responsible for the slowing of NADPH oxidation and ultimately slowing all PET reactions as far upstream as water oxidation by PSII. The CCB cycle appears to be the main target impacted by the AGPase mutation in sta6 as none of the individual PET enzyme fluxes exhibit inhibition relative to the cw15 control when measured in isolation. Our results with PET electron acceptors (Fig. 2.5 C) confirm that, like typical wild algal strains (Rochaix, 2011), reoxidation of the PQ pool is retained in *sta6* as the slowest kinetic step of the PET chain. *sta6* uses mitochondrial respiration to relieve its slowly responding, over reduced, PET chain (Fig. 2.5 B). By contrast, inhibition of mitochondrial respiration in cw15 leads to a comparatively insignificant effect on OER. Thus, respiration appears to be critical for sustaining the low level of photosynthetic oxygen evolution in *sta6*. Consistent with this observation, P700<sup>+</sup> kinetics are impaired when respiration is inhibited in *sta6* (Johnson *et al.*, 2012). Under conditions where *sta6* is unable to maintain redox homeostasis, *sta6* does not depend on PTOX or AOX but potentially Mehler reaction or photorespiration to reduce  $O_2$  and relieve the excess reductant. Thus alternate oxygen consuming pathways become important as redox valves in *sta6*.

The large 4-fold increase in PEP concentration in *sta6* is interesting to note as this molecule is known to function in another  $CO_2$  fixation pathway via phosphoenolpyruvate carboxylase (PEPC) which may therefore become more important in *sta6*. PEPC catalyzes the reaction combining bicarbonate (HCO<sub>3</sub><sup>-</sup>) and PEP to form oxaloacetate. This is an anaplerotic reaction important for amino acid (Ala, Asp, Glu) synthesis. PEPC dependent carbon fixation can even exceed the amount of carbon fixed via the CBB cycle in some unicellular algae under nitrogen limited conditions (Guy *et al.*, 1989).

*sta6* maintained smaller metabolite pool sizes of hexose phosphates (hexose-P) under low light. In the absence of starch biosynthesis with limited carbon fixation, the fixed carbon might be dynamically redirected to TCA cycle to sustain the high levels of protein required for growth leading to a smaller pool size of hexose-P. While, under HL, the TCA cycle appears to become saturated (Table 2-3) forcing an accumulation of the hexose-P. Future metabolic flux analysis can guide in understanding this dynamic redirection of metabolites.

Even though over-reduction of the PET chain is relieved in part by mitochondrial respiration in *sta6* (Fig. 2.5 B), prolonged exposure to high light leads to an accelerated photoinhibition relative to *cw15* (Fig. 2.7 B), indicating that starch production is an important safeguard against photodamage. The light-driven accumulation of excess NADPH and reduced ferredoxin are widely observed to lead to the formation of reactive oxygen species (ROS) in many aquatic phototrophs and plants. A similar phenotype is observed in several other characterized AGPase mutants (Sun et al., 1999; Grundel et al., 2012; Suzuki et al., 2010). Given that starch is the least energy intensive

terminal product among starch, proteins and lipids, it may serve as a low-barrier energy storage buffer to both protect against light energy fluctuations and to enable the biosynthesis of these more energy intensive biopolymers during low light periods, similar to that postulated in higher plants (Caspar *et al.*, 1985), (Stitt *et al.*, 1989), (Ludewig *et al.*, 1998), (Weise *et al.*, 2011) (Geigenberger, 2011).

Carbon that enters central metabolism from the CBB cycle may take two possible routes: the upper gluconeogenesis pathway leading into C6 and C5 carbohydrates, or the lower glycolytic pathway leading to acetyl-CoA that can branch either into the TCA cycle to synthesize amino-acids or to malonyl-CoA to synthesize lipids. Several groups have reported increased lipid yields in sta6 after switching cells to nitrogen deprived growth medium (Work et al., 2010; Goodson et al., 2011; Li et al., 2010a; Li et al., 2010b; Siaut et al., 2011; Fan et al., 2012; Wang et al., 2009). Increased lipid accumulation is not observed under nutrient replete, photoautotrophic conditions in sta6, rather growth is stunted. However, there is a major increase in both acetyl-CoA and malonyl-CoA levels in sta6. Malonyl-CoA formation is catalyzed by ACCase through an ATP-dependent carboxylation reaction of biotin that transfers a carboxyl group to acetyl-CoA (Berg et al., 2002). The ACCase reaction is considered to be the committed step in fatty acid synthesis as there is no other use for this intermediate other than fatty acid production. It is regulated by reversible phosphorylation catalyzed by an AMP-dependent protein kinase (AMPK) (Berg et al., 2002). ACCase is inhibited by its' phosphorylation, while the unphosphorylated form has carboxylase activity. AMPK itself acts as an adenylate nucleotide sensor, being activated by high AMP levels and inhibited by high ATP levels (Berg et al., 2002; Hardie et al., 2002). As such the kinase activity becomes self-limiting at high ATP concentrations. Accumulation of malonyl-CoA in sta6 indicates a reversal of inhibition of ACCase relative to cw15. Given that ATP accumulates in sta6 (Table 2-3), we predict that the higher ATP content likely inhibits AMPK and, in turn, enhances the carboxylase activity of ACCase. Thus, a simple overexpression of ACCase may not be enough to enhance ACCase activity, but needs to occur in parallel with lowering the level of phosphorylation of ACCase, which is achieved by suppression of AMPK phosphorylation activity. Taken together, higher ATP accumulation in *sta6* cells metabolically poises it for higher levels of fatty acyl biosynthesis as seen by elevated levels of acetyl-CoA and especially malonyl-CoA, but downstream blockage in the fatty acid synthesis complex prevents utilization. Thus under photoautotrophic nutrient replete conditions, the activity of the fatty acid synthesis complex, or glycerolipid biosynthetic enzymes appear to be limiting lipid biosynthesis. If the activity of these enzymes are increased (as under nitrogen starvation), it may enable *sta6* to utilize the enhanced Ma-CoA. Transcriptomic studies on *C.reinhardtii* have shown a significant increase in the acyl transferases such as acyl-CoA:diacylglycerol acyltransferase (DGAT) and phospholipid:diacylglycerol acyltransferase (PDAT)) and suggested the involvement of transcription factor NRR1 in potentially regulating the expression of certain DGAT genes (Boyle *et al.*, 2012). Thus, factors beyond precursor availability control lipid biosynthesis.

In conclusion, our data indicate that in *C. reinhardtii*, starch biosynthesis plays a critical role in regulating multiple functions largely through the accumulation/utilization of excess redox and adenylate cofactors. Future research efforts should examine mechanisms to effectively leverage the greatly increased malonyl-CoA levels for enhanced lipid biosynthesis which if successful could allow higher NADPH reoxidation rate and restore high levels of photosynthetic growth.

# Chapter 3: The carbon decision tree of *Chlamydomonas reinhardtii*: How starchless mutant strains redirect carbon metabolism.

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## Summary

Imposed nutrient starvation is a widely used tool to shift microbial cellular metabolism to favor selected bioproducts, but typically causes reduced phototsynthetic rates and biomass accumulation. Under nitrogen (N) starvation, the microalga *Chlamydomonas reinhardtii* redirects biosynthesis from proteins to starch and lipids, and in mutants lacking starch biosynthesis enzymes, preferentially to lipids. To understand the molecular basis for this altered carbon partitioning, a comparative analysis of the intracellular pools of central carbon metabolites and their <sup>13</sup>C fluxes upon N deprivation in strains with and without modified starch synthesis was performed. Both transient <sup>13</sup>C-metabolites fluxes produced from fixed NaH<sup>13</sup>CO<sub>3</sub> and steady-state metabolite pool sizes reveal sustained fluxes through gluconeogenesis as system wide responses to N starvation in all strains. Simultaneous measurements of adenylate nucleotides (ATP, ADP, AMP) revealed a tight correlation between elevated cellular energy charge (CEC) and the flux into gluconeogenesis. Gluconeogenic flux is maintained even in mutants blocked at starch polymerization, but carbon is redirected to the oxidative pentose phosphate (OPP) pathway via G6P, initially forming GAP and DHAP, and ultimately into fatty acid precursors (Mal-CoA). Activation of the OPP pathway produces a significant loss of fixed carbon as CO<sub>2</sub> and the co-generation of excess reductant (3

NADPH per G6P). Thus, the starchless mutant strains use the OPP pathway as a shunt to reallocate photosynthate fated for starch, instead into lipids at the expense of reduced biomass accumulation.

# 1. Introduction

The use of microalgae as cell factories for making fuel precursors is being extensively investigated, with a focus on improving cell growth rates and boosting product yields, typically lipids and isoprenoids (Chisti, 2013); (Melis, 2013). The fastest growing strains often exhibit the lowest lipid productivity. Indeed, these rates have been shown to be inversely correlated (Rodolfi et al., 2009);(Hu et al., 2008); (Wijffels et al., 2010). There is both a kinetic and thermodynamic origin to this inverse correlation. Under rapid growth conditions in nutrient replete media, functional proteins and membrane lipids are the synthesized first, with amino acid biosynthesis being the major sink for primary photosynthate (50-60% of the dry weight). This is typically followed by glycogen/starch accumulation as energy storage carbohydrates. Storage forms of lipids such as triacylglycerides (TAGs) are the slowest to accumulate. This sequence is not surprising as TAGs have twice the energy content of starch per carbon atom and correspondingly consume more reductant for biosynthesis. By contrast, nutrient starvation especially nitrogen (N) starvation, inhibits protein biosynthesis and stimulates the accumulation of non-polar lipids in the form of TAGs (up to 50%) and starch (up to 60%) (Siaut et al., 2011); (Breuer et al., 2012), but severely decreases photoautotrophic growth rates and thus total biomass yields (Chisti, 2007); (Hu et al., 2008). This loss of photosynthesis leads to greater reliance on reduced carbon sources for growth and repair (homeostasis).

Rerouting of carbon flux under nitrogen deprivation results in accumulation of starch and lipid at the expense of protein biosynthesis in many algal strains (Rodolfi *et al.*, 2009); (Hu *et al.*, 2008) including the *Chlorophyceaen* alga *Chlamydomonas reinhardtii (henceforth, C. reinhardtii)* (Ball *et al.*, 1990). Msanne et al, have shown that starch and lipid accumulation are temporally separated under N starvation (Msanne *et al.*, 2012), and this has been shown to be regulated at the mRNA transcript level (Blaby *et al.*, 2013); (Goodenough *et al.*, 2014). In *C. reinhardtii,* starch accumulates first in response to nitrogen starvation, primarily serving as a short-term carbon and

energy store, while TAGs accumulate after more prolonged periods of N starvation (Siaut *et al.*, 2011); (Zhu *et al.*, 2014); (Blaby *et al.*, 2013).

A key unknown is the complex regulation of carbon partitioning into the various terminal sinks that has hindered efforts to substantially alter carbon allocation between lipids and other products while maintaining growth rate (Melis, 2013). The partitioning of assimilated carbon between TAGs and carbohydrates is a highly regulated process, as indicated by the activity of the various transcription factors and the observed changes in the transcriptome and proteome (Wase et al., 2014); (Blaby et al., 2013); (Miller et al., 2010); (Goodenough et al., 2014). At the metabolite level, both of these sinks share the same precursors. The primary product of CO<sub>2</sub> fixation, 3PG can feed directly into starch biosynthesis through gluconeogenesis or can be directed into lower glycolysis to produce acetyl-CoA and malonyl-CoA, precursors for fatty acids (FA) (Fig. 3-1). In *Chlamydomonas*, both starch and fatty acid synthesis occurs in the same intracellular compartment, the chloroplast, and thus direct competition for these metabolites should occur. Hypothesizing a direct competition for carbon precursors, several groups have already reported success in increasing lipids by eliminating starch biosynthesis (Work et al., 2010); (Blaby et al., 2013); (Goodenough et al., 2014); (Wang et al., 2009); (Goodson et al., 2011); (Siaut et al., 2011); (Breuer et al., 2014); (Li et al., 2010); (Miller et al., 2010). However, the work with genetically complemented strains showed increased levels of both lipids and starch (Work et al., 2010), as well as modulated levels of photosynthetic activities as a function of sink strength, suggesting that the carbon partitioning between lipids and starch is more complex than mere competition for precursors (Work et al., 2010); (Ramazanov et al., 2006); (Breuer et al., 2012); (Miller et al., 2010).



Figure 3-1 **Metabolic pathways in** *Chlamydomonas reinhardtii* .Metabolic pathways in *Chlamydomonas reinhardtii* adapted from (Johnson et al., 2013). Abbreviations: CBB, Calvin Benson Bassham cycle; OPP, oxidative pentose phosphate pathway; TCA, tricarboxylic acid cycle; G1P, glucose-1-phosphate; G6P, glucose-6-phosphate; F-6-P, fructose-6-phosphate; FBP, fructose bisphosphate; GAP, glyceraldehyde 3-phosphate; DHAP, dihydroxyacetone phosphate; 3PG, 3 phosophoglycerate; RUBP, ribulose bisphosphate; Glu, glutamate; Gln, glutamine; PEP, phosphoenol pyruvate; PYR, pyruvate; Ac-CoA, acetyl-CoA; CIT, citrate; AKG,  $\alpha$ -ketoglutarate; SUCC, succinate; FUM, fumarate; MAL, malate; OAA, oxaloacetate; ISOCIT, isocitrate; GLYX, glyoxylate

Understanding the carbon decision tree - the temporal and spatial coordination of biochemical pathways responsible for biosynthesis and catabolism of the major carbon sinks is still fragmentary in microalgae. Here, we apply LC-tandem-MS methods to determine steady-state metabolite concentration changes during nitrogen starvation and time-resolved <sup>13</sup>C metabolite kinetics. We identify the main metabolic nodes where redirection of carbon between pathways occurs during nutrient stress, leading to accumulation of starch, or lipids, in both wild type and starchless mutant strains. In contrast to most prior work, we have used autotrophic conditions to strictly avoid any dependence on external reduced carbon. We find that under nitrogen starvation there is selective inhibition of the TCA cycle and active allocation of carbon into gluconeogenesis, irrespective of the presence or absence of starch. Rather than directly channeling the fixed carbon to fatty acids, the starchless mutants use the oxidative pentose phosphate pathway (OPP) to redirect the carbon entering gluconeogenesis to alternate sinks. In addition, we provide evidence indicating ATP/AXP (cellular energy charge) as a potential effector for regulating this metabolic switch.

## 2. Experimental Procedures:

## 2.1) Strains and Culture conditions:

*cw15 sta6* (BAFJ5, henceforth, *sta6*) (Wang *et al.*, 2009), (Zabawinski *et al.*, 2001), CC-124 wild type mt- [137c] (henceforth CC124), CC-4425 *cw15* nit2-203 mt+ [D66] (henceforth D66), and CC-4349 *cw15* mt<sup>+</sup> (Goodenough 330A, henceforth, *cw15*) were obtained from the Chlamydomonas resource center (University of Minnesota). The starchless mutant strain (*sta7-10*) in the CC425 background (henceforth *sta7*), and its genetic complement (*c19*) were isolated as described previously (Work *et al.*, 2010). The lineages of the six strains is summarized in Supporting Information (Fig. 3-2).



Adapted from (Haris, 2009) and Chlamydomonas center

Figure 3-2 Relationships among the strains chosen for carbon partitioning analysis.

sta6 and sta7 are unable to produce starch owing to genetic lesions that block starch synthesis in these strains. In sta6, the small catalytic subunit of ADP glucose pyrophosphorylase is disrupted via insertion of pArg7.8 into the genome of CC-4568 strain and as a result, the ADPglucose fated for starch biosynthesis is not synthesized. This mutant accumulates less than 1% of the starch observed in the WT cells under nutrient starvation (Ball et al., 1990), (Work et al., 2010) The sta7 mutant contains a disrupted isoamylase gene due to insertion of the pJD67 into the genome of the CC-425 background strain. Isoamylase is an important debranching enzyme needed for the proper formation of crystalline starch. In the absence of isoamylase, sta7 has a severely attenuated starch accumulation phenotype and instead accumulates a glycogen like product (Mouille et al., 1996), (Dauvillee et al., 2001; Work et al., 2010). c19 clone was previously isolated by complementing sta7 with the isoamylase gene (Work et al., 2010). c19 was found to accumulate higher levels of starch and lipids as compared to sta7 under nitrogen starvation in the presence of acetate (Work et al., 2010) and is used as one of the starch synthesizing controls in this study. Previous research has shown significant variability in starch and lipid accumulation in different laboratory strains (Siaut et al., 2011). Therefore, for generalizing our results we examined three starch synthesizing control strains cw15, D66 and CC124. Comparing a range of strains should enable the identification of a global response pattern to nitrogen starvation while the addition of starchless mutant strains will help in elucidating carbon partitioning.

Liquid cultures were grown in phosphate buffered, Sueoka's high salt medium (HS) supplemented with 9.4 mM NH<sub>4</sub>Cl (Harris *et al.*, 2009). The cultures were maintained on an orbital shaker (120 rpm) under continuous illumination (80-100  $\mu$ E m<sup>-2</sup> s<sup>-1</sup> PAR) at 26°C and maintained under 1% CO<sub>2</sub>/air atmosphere. For nitrogen starvation experiments, cultures were grown to late log phase in nitrogen replete HS media, harvested, washed with NH<sub>4</sub>Cl free HS media (HS-N) and resuspended at ~2.5 x 10<sup>6</sup> cells ml<sup>-1</sup> in either HS media replete with NH<sub>4</sub>Cl (HS+N) or HS-N media. Samples for analysis were taken at multiple time points as indicated in the main text. Cell counts were analyzed using a Z2 counting chamber.

# 2.2) Lipid, glucose and protein analysis

Total lipids were quantified using GC-FID after converting lipids to fatty acid methyl ester via transesterification (Work *et al.*, 2010,); (Radakovits *et al.*, 2012). The anthrone assay was used to quantify the total reducing carbohydrate content (Morris, 1948). Total protein content was determined using the DC protein assay kit (Bio-Rad, Hercules, CA) according to the manufacturer's instructions.

## 2.3) Metabolite extraction and analysis

LC-MS/MS analysis was done as described previously (Krishnan, 2015). Briefly, 1.8 ml of *C. reinhardtii* cultures were injected into 0.2 ml of an 80:20 MeOH:water mixture for fast quenching and centrifuged at 5000 rpm for 2 min at -10°C. The supernatant was carefully removed and the pellet was resuspended in  $60\mu$ L ice-cold extraction solvent (60:40 MeOH:water). The suspension was sonicated in a bath sonicator for one min then incubated at -20°C for 20 min. To remove cell debris, the samples were centrifuged for 5 min at 5000 rpm at -10°C. Supernatant was vacuum dried (CentriVap Benchtop Vacuum Concentrator, *Labconoco*) and the resulting pellet was resuspended in LC-MS grade water. These samples were injected into an *Agilent* 6490 triple quadrupole mass analyzer coupled to a high performance liquid chromatograph (HPLC, *Agilent Technologies*) for targeted metabolite analysis (Kenchappa *et al.*, 2013). Standard solutions at 5 different concentrations were used to obtain calibration curves for each metabolite. Final concentrations of each metabolite per 10<sup>6</sup> cells were calculated using the calibration curve. Fold changes were calculated as a ratio of concentration of metabolites at the given test condition divided by the time 0 (T<sub>0</sub>). Cellular energy charge (CEC) was calculated as:

$$CEC = \frac{ATP + 0.5 \times ADP}{ATP + ADP + AMP}$$

To analyze metabolic flux, <sup>13</sup>C-labeling was performed using 10mM <sup>13</sup>C NaH<sup>13</sup>CO<sub>3</sub> in addition to the already existing 5mM <sup>12</sup>C NaH<sup>12</sup>CO<sub>3</sub> in the medium (see main text). Briefly, the cultures were grown to late log phase, washed with HS-N medium and resuspended in either HS+N or HS-N medium supplemented with 5mM NaH<sup>12</sup>CO<sub>3</sub> to acclimate the cells to bicarbonate as the carbon source. HS+N cultures were allowed to grow for 3 h and then used for <sup>13</sup>C-labeling. For HS-N labeling analysis, cultures were grown for 21 h in HS-N medium, washed and resuspended in fresh media (HS-N + 5mM <sup>12</sup>C NaH<sup>12</sup>CO<sub>3</sub>) to avoid any artifacts in <sup>13</sup>C-labeling caused by the accumulation of waste products or by nutrient depletion (Yuan *et al.*, 2008). These samples were allowed to grow for another 3 h, thus covering the total growth period of 24 h under –N conditions and then used for <sup>13</sup>C-labeling. 10mM <sup>13</sup>C NaH<sup>13</sup>CO<sub>3</sub> was added to the cell culture to initiate labeling. Samples for metabolite analysis were withdrawn at 0 s (prior to <sup>13</sup>C-label addition), 30 s, 1 min, 2 min, 4 min, 15 min and 30 min, and analyzed by LC-MS/MS as described above. The fraction of <sup>13</sup>C-labeled fraction of metabolites was determined as described previously (Hasunuma *et al.*, 2010). Briefly, the <sup>13</sup>C-labeled fraction % of a metabolite containing n carbon atoms is calculated as

13C fraction (%) = 
$$100 * \sum_{i=1}^{n} \frac{i \times mi}{n}$$

where mi is the total ion count for the isotopomer of the metabolite containing i  ${}^{13}$ C atoms. Metabolic turnover rate was calculated from the initial slope of the  ${}^{13}$ C fraction vs time from 0-2 min (Hasunuma *et al.*, 2010).

## 2.4) Pyridine nucleotide fluorescence kinetics:

Intrinsic NAD(P)H fluorescence emission was measured in whole cells with a home-built fluorometer as described previously (Williams *et al.*, 2010), (Krishnan *et al.*, 2015), (Mi *et al.*, 2000). For NAD(P)H excitation, an LED (365 nm wavelength) in combination with an optical filter (2 mm UG 11, Schott) was used, while emission intensity was detected by a filtered low noise

photomultiplier (PMT, R2059, *Hamamatsu, Japan*). To eliminate scattered measuring light and chlorophyll fluorescence emission into the PMT, a combination of a long-pass filter (KV418, *Schott*), a short-pass filter (DT Cyan), and a blue-green glass filter (BG39, *Schott*) were used that restricted transmission to  $420\mp10$  nm. The detection and excitation pathways were perpendicular. Red actinic illumination was provided by a bright LED ( $660\pm 5$  nm), Hewlett-Packard). An NMR tube (outer diameter 5 mm) was used as the sample holder. Cells were harvested every 24 h post nitrogen starvation, washed and resuspended in 20% ficoll solution made in HS+N media and containing 5 mM bicarbonate. Ficoll was used to slow the settling rate of cells in the fluorometer. The sample emission was normalized to the same total pyridine nucleotide (NADP<sup>+</sup> + NADPH) content determined by LC-MS/MS ( $0.25 \mu$ M ml<sup>-1</sup>). Samples were dark adapted for 25 min to get a stable base line before beginning the actual measurements.

## 3. Results:

## 3.1) N starvation induces C flux changes between lipids, proteins and carbohydrates.

Fig. 3-1 is a schematic representation of the central carbon metabolism of *C. reinhardtii* based on the model described by Johnson and Alric (Johnson *et al.*, 2013). To analyze carbon partitioning under N starvation, six *C. reinhardtii* strains were chosen: two starchless mutants (*sta6* and *sta7*) and four starch synthesizing strains (CC124, D66, *cw15* and *c19*) (*See Materials and methods*, Fig. 3-2). For quantifying biomass accumulation under N starvation, all the strains were precultured photoautotrophically to late log phase in HS+N medium and kept under a 1 % CO<sub>2</sub>/air atmosphere (*See Materials and methods*). Cultures were then centrifuged and resuspended in HS-N medium at fixed cell densities and allowed to grow for 96 h under 80-100  $\mu$ E m<sup>-2</sup> s<sup>-1</sup> of white light. The major biochemical products, including total reducing carbohydrates (TRC), lipids and proteins were quantified in all six strains at 0h, 24h and 96h after initiation of nitrogen starvation (Fig. 3-3). As controls, the strains were also grown in HS+N media.



Figure 3-3 **Changes in biomass composition of six** *C. reinhardtii* **cultures under nitrogen starvation**. A) Total biomass content (g L-1), B) total protein, C) total reducing carbohydrate, and D) total lipid content in (g L-1). E) is the growth curve displaying the changed in cell density over time for all the six strains in nitrogen replete media (+N) above the dotted line and N-free media (–N) below the dotted line. F) proteins, G) total reducing carbohydrates (TRC), and H) lipids during N starvation, each expressed as percentage of the total biomass. Values are the average of 3 measurements  $\pm$  SE.
As expected, after 96 h of nitrogen starvation, all strains grew much slower relative to nitrogen replete culturing, experiencing at most one cell doubling (Fig. 3-3 B). However, the calculated biomass (given by the sum of the three major fractions: TRC + proteins + lipids) increased at least 2.5 fold (Fig. 3-3 A), indicating active carbon fixation taking place under nitrogen starvation rather than merely remodeling of carbon from preexisting biomass. Among the starchless mutant strains, *sta6* and *sta7*, accumulated 2-5 fold less biomass than the starch-synthesizing control strains after 96 h of N starvation (Fig. 3-3 A).

As expected, in the absence of N, starch (measured as total reducing carbohydrates, TRC) and lipids become the dominant sinks in starch competent strains (Fig. 3-3 C & D), while in *sta6* and *sta7* there was no significant increase in TRC content over 96 h of N starvation (Fig. 3-3 C & G). This contrasts with the absolute protein content (Fig. 3-3 B), which did not change appreciably or had a minor loss (*sta7*). After normalization to total calculated biomass, the protein fraction of total cellular biomass decreased by up to 4.5 fold (Fig 3-3 F). Amongst all the strains, *sta6* accumulated the highest lipid content on a dry weight basis under these photoautotrophic conditions (Fig. 3-3 D & H). The other starchless mutant, *sta7*, had a smaller increase in lipid content versus *sta6* on a weight basis (g L<sup>-1</sup>), though it synthesized more lipids compared to starch competent strains on a % biomass basis (Fig. 3-3 D & H). Among the starch synthesizing strains, CC124 had the lowest lipid content (-N) at 96 h, consistent with the reported range found among the different *C. reinhardtii* control strains (Siaut *et al.*, 2011). The presence of a cell wall in CC124, which supports a higher osmotic pressure unlike the other strains, may contribute to its reduced lipid content (Johnson *et al.*, 2013).

#### 3.2) Metabolic responses to N starvation.

Given that the starchless mutants preferentially partition carbon into lipids, we sought to identify how this is represented in the profile of intracellular metabolites by determining the concentrations of the central carbon metabolites for all six strains (+/- N) using LC-MS/MS

methods (Bennette *et al.*, 2011), (Krishnan, 2015). Initially, a time-coarse profile was taken, with sampling at 0, 24 and 96 h after the initiation of N starvation. For 0 h measurement, the samples were suspended in HS+N medium at the same cell density as that in the HS-N medium. The experimental setup is shown in Fig. 3-4 F. Individual metabolite concentrations are given in Table 3-1. The time-course results of all the detected metabolites are summarized in Fig 3-4 A. Columns represent the fold-change in the metabolites occurring at the different starvation points for each individual strain. Fold-changes are calculated relative to the initial metabolite concentrations at time zero,  $T_0$ . Because we are comparing six strains of different lineages, we chose to bin these in units of 3-fold differences to focus on major trends. Greater detail can be found in Table 3-1.



Figure 3-4 Metabolite concentration changes at two time points following transfer of cells to N-free or to fresh N-replete media. A) Heat map showing the ratio of metabolite concentrations [N-free/N-replete] after 24h (first 6 columns) and 96h (last 6 columns) of nitrogen starvation. Metabolite pool sizes are given in Table 3-1. Panels B) C) D) and E) give changes in cellular energy charge (CEC), ATP, ADP, and AMP content at each time point during nitrogen starvation. Values are the average of 3 measurements  $\pm$  SE. Panel F) is a schematic represention of the experimental set up.

Pathway	Metabolites		Concentration (µmol /10 <sup>8</sup> cells)													
				т.	sta6		sta7		CC12 4		cw15		c19		D66	
			Time	Avg	SE	Avg	SE	Avg	SE	Avg	SE	Avg	SE	Avg	SE	
			0	0.208	0.016	0.567	0.013	0.944	0.077	0.578	0.095	0.726	0.024	0.414	0.063	
	Glucose-1- Phosphate	G1P	24-N	0.378	0.051	1.454	0.109	1.587	0.145	1.022	0.122	1.192	0.189	0.614	0.019	
	-		96-N	0.337	0.082	1.252	0.086	0.295	0.241	0.361	0.241	0.703	0.072	0.492	0.157	
~	~ ~ ~		0	0.198	0.012	0.576	0.031	0.950	0.061	0.681	0.118	0.828	0.069	0.424	0.061	
Gluconeoge nesis	Glucose-6- Phosphate	G6P	24-N	0.325	0.069	1.501	0.002	1.534	0.115	1.097	0.144	1.416	0.202	0.730	0.009	
	-		96-N	0.380	0.053	1.247	0.122	0.804	0.143	0.710	0.009	0.872	0.179	0.395	0.139	
	Fructose-		0	0.088	0.009	0.289	0.011	0.451	0.045	0.306	0.058	0.448	0.035	0.214	0.039	
	6- Di	F6P	24-N	0.164	0.040	0.684	0.049	0.834	0.036	0.584	0.066	0.734	0.093	0.475	0.011	
	Phosphate		96-N	0.190	0.033	0.569	0.004	0.470	0.094	0.416	0.001	0.273	0.035	0.255	0.087	
	Ribulose	RUBP	0	0.218	0.028	0.227	0.012	0.412	0.088	0.356	0.038	0.412	0.107	0.511	0.124	
	bisphosph		24-N	0.069	0.036	0.215	0.026	0.897	0.091	0.644	0.109	0.745	0.100	1.007	0.031	
	ate		96-N	0.061	0.002	0.309	0.030	0.653	0.197	0.832	0.184	0.630	0.081	1.098	0.324	
CBB cycle	3		0	0.526	0.023	0.493	0.121	0.404	0.030	1.583	0.164	0.711	0.297	0.237	0.033	
	Phosphogl	3PG	24-N	0.850	0.190	1.086	0.041	0.808	0.120	3.200	0.178	2.290	0.825	0.370	0.016	
	ycerate		96-N	0.104	0.005	0.532	0.291	0.201	0.176	0.655	0.061	0.426	0.051	0.097	0.074	
			0	0.026	0.005	0.302	0.019	1.155	0.239	0.637	0.064	0.883	0.024	0.494	0.097	

Dih ac	Dihydroxy	DHA	24-N	0.196	0.044	3.824	0.264	2.496	0.182	1.352	0.259	2.606	0.382	1.384	0.100
	phosphate	Р	96-N	0.110	0.047	5.572	0.236	1.841	0.093	1.666	0.054	1.555	0.107	0.858	0.093
	Fructose		0	0.019	0.001	0.037	0.016	0.716	0.072	0.134	0.009	0.219	0.057	0.327	0.046
	Bisphosph	FBP	24-N	0.035	0.003	0.205	0.038	1.229	0.127	0.453	0.046	0.573	0.237	2.301	0.577
	ate		96-N	0.034	0.012	0.309	0.051	1.199	0.274	0.725	0.035	0.410	0.048	0.611	0.090
	Glyceralde		0	0.000	0.000	0.002	0.000	0.007	0.002	0.004	0.000	0.004	0.000	0.004	0.001
	hyde 3 phosphate	GAP	24-N	0.001	0.000	0.015	0.002	0.011	0.002	0.007	0.001	0.005	0.003	0.005	0.001
			96-N	0.001	0.000	0.008	0.001	0.012	0.000	0.006	0.002	0.005	0.000	0.005	0.001
	Phosphoen		0	0.067	0.007	0.053	0.011	0.068	0.005	0.203	0.028	0.215	0.046	0.026	0.005
	ol	PEP	24-N	0.042	0.015	0.052	0.006	0.052	0.011	0.153	0.034	0.135	0.075	0.014	0.004
	pyruvate	PEP	96-N	0.023	0.068	0.095	0.036	0.129	0.046	0.133	0.093	0.138	0.007	0.015	0.016
Lower glycolysis															
			0	0.347	0.105	0.573	0.101	0.549	0.133	0.521	0.068	0.802	0.099	0.217	0.038
	Pyruvate	PYR	24-N	0.124	0.152	0.681	0.200	0.260	0.160	0.454	0.101	0.655	0.193	0.113	0.207
			96-N	0.297	0.333	1.294	0.484	1.263	0.010	0.989	0.068	1.620	0.247	0.647	0.253
			0	0.011	0.019	0.058	0.005	0.029	0.007	0.023	0.002	0.112	0.016	0.009	0.001
	Acetyl- CoA	Ac- CoA	24-N	0.013	0.001	0.094	0.004	0.014	0.000	0.012	0.002	0.122	0.040	0.010	0.002
			96-N	0.043	0.015	0.150	0.023	0.017	0.001	0.037	0.000	0.209	0.017	0.015	0.001
I CA cycle															
	Citrata	CIT	0	0.015	0.019	0.024	0.002	0.016	0.003	0.029	0.000	0.050	0.008	0.015	0.002
	Curate	un	24-N	0.024	0.021	0.043	0.004	0.018	0.065	0.028	0.001	0.056	0.027	0.020	0.008

			96-N	0.064	0.009	0.041	0.007	0.073	0.026	0.124	0.010	0.103	0.024	0.040	0.007
	<i>a</i> -		0	0.031	0.003	0.150	0.017	0.064	0.005	0.021	0.003	0.258	0.002	0.238	0.009
	ketoglutar	AKG	24-N	0.044	0.004	0.520	0.103	0.109	0.819	0.026	0.002	0.463	0.040	0.458	0.135
	ate		96-N	0.145	0.014	0.275	0.023	0.196	0.135	0.143	0.030	0.502	0.024	0.907	0.266
			0	0.094	0.011	0.282	0.036	0.219	0.069	0.100	0.010	0.441	0.013	0.050	0.003
	Succinate	SUCC	24-N	0.192	0.001	0.458	0.069	0.394	0.017	0.216	0.030	0.830	0.089	0.107	0.005
			96-N	0.082	0.020	0.239	0.054	0.192	0.037	0.141	0.028	0.493	0.070	0.071	0.006
			0	0.334	0.037	0.380	0.026	0.558	0.096	0.508	0.055	0.260	0.018	0.270	0.047
	Fumarate	FUM	24-N	0.618	0.055	1.131	0.274	0.935	0.105	0.917	0.107	0.841	0.139	0.670	0.016
			96-N	0.230	0.041	0.211	0.211	0.605	0.157	0.231	0.189	0.337	0.162	0.271	0.170
	Malate M		0	0.188	0.011	0.724	0.113	0.743	0.125	0.309	0.023	0.823	0.089	0.344	0.045
		MAL	24-N	0.357	0.021	1.758	0.157	1.167	0.152	0.552	0.060	1.786	0.297	0.857	0.054
			96-N	0.150	0.057	0.950	0.074	0.811	0.181	0.228	0.029	1.162	0.177	0.486	0.194
			0	0.082	0.010	0.027	0.006	0.050	0.003	0.030	0.006	0.021	0.001	0.063	0.022
	Glutamine	GLN	24-N	0.015	0.002	0.026	0.011	0.002	0.001	0.025	0.005	0.025	0.016	0.003	0.000
			96-N	0.019	0.000	0.006	0.000	0.004	0.000	0.012	0.000	0.020	0.002	0.008	0.001
others			0	0.846	0.096	3.695	0.224	2.133	0.341	0.992	0.113	3.919	0.224	1.317	0.140
	Glutamate	GLU	24-N	0.250	0.016	4.499	0.527	0.062	0.037	0.928	0.153	6.141	0.793	0.026	0.014
		SUCC FUM MAL GLN	96-N	0.547	0.175	1.159	0.085	0.375	0.265	1.417	0.163	3.204	0.548	0.277	0.102

	Malonyl- CoA		0	0.007	0.001	0.002	0.000	0.001	0.003	0.001	0.000	0.005	0.003	0.001	0.000
		Ma- CoA	24-N	0.013	0.000	0.007	0.002	0.001	0.000	0.001	0.000	0.003	0.010	0.001	0.000
	CON		96-N	0.033	0.000	0.053	0.003	0.001	0.000	0.001	0.000	0.008	0.003	0.002	0.001
			0	0.004	0.002	0.003	0.001	0.028	0.004	0.002	0.001	0.007	0.003	0.012	0.005
	Sucrose	SUCR	24-N	0.011	0.007	0.021	0.005	0.079	0.006	0.004	0.001	0.017	0.007	0.024	0.003
		Ma-CoA   SUCR   AMP   AMP   ADP   ATP	96-N	0.046	0.016	0.034	0.019	0.100	0.005	0.011	0.002	0.018	0.002	0.062	0.036
			0	0.450	0.173	0.585	0.077	0.309	0.013	0.303	0.014	0.803	0.197	0.401	0.064
	AMP	AMP	24-N	0.356	0.017	0.472	0.017	0.251	0.010	0.184	0.024	0.826	0.098	0.273	0.030
			96-N	0.243	0.040	0.337	0.012	0.224	0.005	0.185	0.036	0.434	0.086	0.207	0.029
			0	0.893	0.119	0.836	0.082	0.191	0.023	0.625	0.050	1.335	0.069	0.572	0.047
Adenylates	ADP	ADP	24-N	0.597	0.005	0.762	0.031	0.207	0.025	0.388	0.039	1.129	0.238	0.364	0.043
			96-N	0.226	0.102	0.345	0.010	0.160	0.017	0.328	0.134	0.772	0.123	0.306	0.027
			0	0.790	0.057	0.899	0.169	0.457	0.064	0.558	0.041	1.219	0.005	0.694	0.053
	ATP	ATP	24-N	1.292	0.012	2.776	0.290	0.462	0.017	0.302	0.095	1.172	0.466	0.448	0.161
			96-N	0.254	0.013	0.344	0.045	0.168	0.058	0.121	0.044	0.375	0.469	0.170	0.191

Table 3-1: LC-MS/MS results for various metabolites. Standard error of mean are represented in columns adjacent to the average metabolite concentration calculated as µmoles per 10<sup>8</sup> cells.

#### Metabolic response to N starvation in starch containing strains:

In the absence of N, cellular levels of glutamate and glutamine decreased (1-50 fold) indicating the inability of the strains to synthesize nitrogenous compounds (Fig. 3-4 A). Within the first 24 h of nitrogen starvation, the pool sizes of several metabolites increased and then subsequently decreased back to the  $T_0$  levels at 96 h, indicating a dynamic remodeling of metabolism over the course of N starvation (Fig. 3-4 A). Most metabolites of the TCA cycle increased by 1.5-3.5 fold within the first 24 h, except for citrate and acetyl-CoA. However, at 96 h, fumarate, succinate and malate decrease back to  $T_0$  levels or lower while the  $\alpha$ -ketoglutarate (AKG) and citrate (CIT) remained higher (2-5 fold higher than  $T_0$ ). The inverse behavior between CIT-AKG and the rest of the TCA cycle metabolites suggest a slower conversion of AKG into succinate or a very high influx of carbon into AKG derived from either the deamination of glutamic acid by glutamate dehydrogenase or from newly fixed carbon entering the TCA cycle. Nonetheless, the high accumulation of CIT and AKG indicates a kinetic bottleneck at AKG.

Metabolites corresponding to upper gluconeogenesis, G1P, G6P and F6P, increased by 1.5-3 fold at 24 h and then decreased back to  $T_0$  levels at 96 h of nitrogen starvation (Fig. 3-4 A). This increase correlates with an increase in the production of sucrose (albeit at low concentrations) at both 24 h and 96 h after N starvation, confirming the increased flux through gluconeogenesis. Contrary to the upper glycolytic intermediates, metabolites of lower glycolysis that lead to AcCoA and to the fatty-acid precursor MaCoA did not show any significant difference in the first 24 h. However, both AcCoA and MaCoA increased appreciably at 96 h of nitrogen starvation, indicating a major change in carbon allocation toward fatty acid synthesis following a 24 h lag.

The total adenylate nucleotides were also quantified at each of the time points and the cellular energy charge (CEC) was calculated (Fig 3-4 B-E). Overall, both CEC and the adenylates tend to decrease during the nitrogen starvation period across all the starch containing strains.

## Metabolic response to nitrogen starvation in starchless mutant strains:

A main aim of these experiments is to identify the underlying metabolic changes to account for the higher lipid fraction in the starchless mutant strains. At 24 h of N starvation, a select few metabolites displayed different concentration profiles between starch producing and starchless mutant strains (Fig. 3-4 A). At 24 h, AcCoA and MaCoA increase (1.6-fold and 1.5-3-fold higher, respectively) in both *sta6* and *sta7* strains (Fig. 3-4 A), indicating an earlier onset of fatty acid biosynthesis in these mutants. Even though starch biosynthesis is blocked in both these mutants, the pool sizes of several upper gluconeogenic intermediates (G1P, G6P, F6P) and sucrose increase at 24 h and do not diminish at 96 h (Fig. 3-4 A). The starchless mutant strains also show an increased CEC at 24 h (unlike the starch containing strains) caused by an increase in the total ATP content at this time point (Fig. 3-4 B-E). These results indicate that in the absence of starch synthesis, carbon accumulates as intermediates of gluconeogenesis and excess ATP is not utilized.

The two-point timeline of metabolite concentrations (Fig. 3-4 A) suggests a temporal separation of the central carbon pathways leading to starch, and both lipids and amino acid precursors. In all of the starch synthesizing strains, there is an earlier allocation of carbon into gluconeogenesis, followed by intermediates of lipid synthesis (AcCoA, MaCoA) and the upper TCA cycle intermediates involved in amino acid synthesis (AcCoA, CIT, AKG). By contrast, in the starchless mutant strains, carbon allocation to all three pathways appears simultaneously within the limited time resolution.

We undertook a more detailed time course in two representative strains, *sta6* and *cw15*. These strains were chosen as additional literature on their transcriptomic response under nitrogen starvation is available (Blaby *et al.*, 2013), (Goodenough *et al.*, 2014). Samples were taken at intervals of 3, 6, 10, 24, 31, 48, 72 and 96 h post nitrogen starvation (Fig. 3-5 A-E). The experimental setup is schematically represented in Fig. 3-5 F.





Figure 3-5 **Time resolved metabolite change.** A) Time course of the steady-state changes of metabolite concentrations following initiation of nitrogen starvation through 96 h. Two cell lines are shown *sta6* (red) and *cw15* (black). The pathways of central carbon metabolism are depicted and key nodes are highlighted where flux changes are prominent. Concentrations of three adenylate energy metabolites and CEC are listed. The Y axis units are  $\mu$ mol (10<sup>6</sup> cells)<sup>-1</sup>. Values are the average of 3 measurements ± SE. B) Schematic representation of the experimental set up for the time resolved metabolic analysis.

*Gluconeogenesis:* The first 10 h of nitrogen starvation acts as a preparatory phase, where the upper glycolytic metabolite levels increase in concentration, especially 3PG, DHAP, FBP, G6P, G1P and ADP-glucose (Fig. 3-5 A). Post 10 h, the gluconeogenic intermediates G6P, G1P and ADP-glucose start to decrease while, DHAP and FBP levels remain high. These results, taken together with high TRC content at both 24 h and 96 h (Fig. 3-3) indicate a sustained feeding of carbon into the gluconeogenic route via DHAP and FBP with a simultaneous drain of G6P into starch. As DHAP and FBP show a very different trend than both the preceding (3PG) and succeeding metabolites (G1P) (Fig. 3-5 A), we suggest that this junction is key to controlling carbon allocation to starch. We call this the 'Carbohydrate Node' and label it as such in Fig 3-5 A.

<u>TCA cycle</u>: Parallel to the increase in the gluconeogenic intermediates, there is also an initial increase in the TCA cycle metabolites, especially succinate, malate and fumarate while AKG and citrate decrease in concentration (Fig. 3-5 A). This occurs in coordination with a slight increase in protein content at 24 h (Fig. 3-3 B). Post 24 h, AKG and citrate start to accumulate, while succinate, malate and fumarate decrease. In addition, both glutamate and glutamine - products derived from AKG - show a large decreasing trend. Collectively, these temporally coordinated changes in metabolites indicate that the metabolic switch away from protein biosynthesis occurs at the level of CIT/AKG. This observation (Fig. 3-5 A) is in parallel to the results observed in Fig. 3-4 A. Henceforth, we call CIT-AKG node as the 'TCA cycle node' controlling carbon redirection.

<u>Lipid metabolites</u>: In parallel with the continuous accumulation of AKG over 96 h, the concentration of the FA precursors AcCoA and MaCoA increase continuously after a lag of 31 h following N starvation (Fig. 3-5 A). The concentration rise of MaCoA is particularly large reaching 12 fold increase. The lag time clearly shows the temporal separation of carbon allocation initially into carbohydrates/starch biosynthesis followed by build up of precursors to FA synthesis. To

emphasize this key metabolic control point, we highlight MaCoA as the 'FA node' in Figure 3-5 A.

<u>Adenylate nucleotides</u>: At the start of N starvation, there is an immediate increase in the level of ATP and CEC. The ATP and CEC levels then decrease (Fig. 3-5 B-E) in parallel with the consumption of the gluconeogenic precursors, G1P and G6P, eventually after 24-36 h dropping below the initial levels. The time resolved measurements give a better resolution of the changes in the in the coarse measurements. The results show that even the starch containing strains show an initial increase in the CEC levels till 10h (not seen in the coarse profiling) after which the CEC levels start to drop (seen in coarse profiling). Thus the time resolved measurements allow us to gain more resolved data that was not observed in the coarse profiling. Overall at the end of 96h of N starvation, the total adenylate nucleotide pool size decreased, reflecting the quantitative impact of N starvation on adenylate energy.

## Time-resolved metabolic changes in the starchless mutant sta6:

Higher time resolution studies (Fig. 3-5) reveal that within the first 10 h of N starvation, *sta6* starts from a lower initial level of gluconeogenic intermediates (FBP and G6P) which then barely increase compared to their levels in the control strain *cw15*. On the other hand, significant accumulation of G1P and 6PG is observed. As it is previously established that the irreversible step in gluconeogenesis of green algae is the hydrolysis reaction of FBP to F6P and inorganic phosphate (Huppe *et al.*, 1989), the very low levels of the downstream metabolite G6P and high levels of G1P, indicates that it must be consumed by downstream reactions other than starch formation. The second difference observed is an earlier increase in both AcCoA and MaCoA levels; MaCoA starts to rise sharply at 24 h and reaches a higher level in *sta6* compared to onset at 31 h in *cw15* (Fig. 3-5 A). This indicates an earlier entry and greater accumulation of carbon at the 'FA node' in *sta6*. The onset and subsequent plateau of accumulation of G1P, the immediate precursor to starch

biosynthesis (Fig. 3-5). This suggests that in the starchless mutant blocking of the carbon flux into the TCA cycle at AKG/GLU due to N starvation leads to carbon allocation into FA biosynthesis pathway.

Similar to the starch synthesizing strains, *sta6* also shows an initial increase in both the ATP content and the CEC levels by *ca*. 60%, followed by recovery to even lower levels at 96 h. However, unlike *cw15*, both ATP and CEC are sustained at higher levels throughout the course of N starvation in *sta6* (Fig. 3-5). Thus, cells of the starchless mutant have an excess of adenylate energy relative to the control strain, which can potentially lead to greater inhibition of many enzymes involved in energy metabolism.

#### 3.3) Carbon redirection away from starch occurs at the level of G6P in the sta6 strain.

To test how the loss of starch polymerization affects the flux through gluconeogenesis, we performed <sup>13</sup>C kinetic labeling experiment using LC-MS to determine the first order rate constant of the metabolites in *cw15* and *sta6*, using <sup>13</sup>C-NaH<sup>13</sup>CO<sub>3</sub> to follow formation of <sup>13</sup>C-labeled metabolites. This was done in the presence and absence of N. As significant differences in carbon allocation into the metabolic pathways were observed at 24 h of N starvation (Figs. 3-4 and 3-5), this time point was chosen to inject 10 mM <sup>13</sup>C-sodium bicarbonate (100%), followed by sampling at 0.5, 1, 2, 4, 15 and 30 min to assess metabolite turnover (Fig. 3-6 B). To mimic the results from the time resolved metabolite analysis, samples resuspended in HS+N medium were used as the control for <sup>13</sup>C analysis while for HS-N analysis, samples were grown in HS-N medium for 24 h. The choice of 3h for the HS+N analysis is described in the methods section (See Materials and Methods). The ratio of <sup>13</sup>C to the total carbon in each metabolite was calculated from the mass isotopomer distributions (See Materials and Methods). Fig. 3-6 A gives the concentration of <sup>13</sup>C-labeled metabolites compared under +N and -N conditions. From this one can extract the rate of formation of labeled metabolite and the fraction of the total metabolite pool that is labeled. The initial slope of the <sup>13</sup>C -fraction versus time has been used as an estimate of the initial formation



Figure 3-6 **13C analysis of N starvation.** Time course of formation of <sup>13</sup>C metabolites in intact cells of *cw15* and *sta6*, following transfer to N-replete media (solid black) and N-free media (grey dash). Y axis represents fraction of <sup>13</sup>C labeled metabolite. Values are the average of 2 measurements,  $\pm$  SE. B) Schematic of the procedure for <sup>13</sup>C-labeling from NaH<sup>13</sup>CO<sub>2</sub>.



Figure 3-7 **Initial rate of formation of the metabolite.** The initial formation rate  $(min^{-1})$  of <sup>13</sup>C-metabolites in C. reinhardtii following injection of NaH<sup>13</sup>CO<sub>3</sub> (10 mM) with N-replete media (white bars) or without nitrogen (grey bars, -N). The rate is calculated by fitting the initial labelled fraction (0 min-2 min) to a straight line. Values are the averages of 2 biological replicates ± S.E. Top panel is for *cw15* and the bottom panel is for *sta6*. B) Carbon partitioning at G6P under N starvation condition. Turnover rates from Fig. 6A and the pool sizes from Table 3-1 for 6PG and G1P are presented.

Fig. 3-7 A shows that the most rapidly labeled metabolites in both cw15 and sta6 under nitrogen replete (+N) conditions (PYR, PEP, 3PG) are not confined to the Calvin cycle, although the first Calvin cycle product, 3PG, is among the fastest labelled metabolites. This outcome indicates that other  $CO_2/HCO_3^-$  reduction sites must contribute to the rapid labeling across so many diverse metabolites. C. reinhardtii has two other CO<sub>2</sub>/HCO<sub>3</sub>-utilizing enzymes, pyruvate synthase  $(AcCoA + 2Fd_{red} + CO_2 + 2H^+ \leftrightarrow PYR + 2Fd_{ox} + CoA)$  and PEP carboxylase  $(PEP + HCO_3^- \leftrightarrow PYR + 2Fd_{ox} + COA)$ OAA + Pi) (Chen et al., 1971), that function to restore metabolites that are consumed during biosynthesis. These anaplerotic reactions are required for proper functioning of the TCA cycle under normal conditions. Their possible contribution during -N nutrient stress is indicated. Pyruvate synthase requires high reductant poise (Fd red), while PEP carboxylase is active only at high CEC, thus limiting both reactions to these special conditions. The slower labeling of metabolites such as DHAP and AKG than their downstream metabolites indicates the presence of subcellular compartmentation or enzymatic channeling of these metabolites (Yuan et al., 2008). The kinetic data in Fig. 3-6 also reveals that in *cw15* significant labeling of G1P and ADP-glucose was only observed under -N conditions, consistent with significant accumulation of total reducing carbohydrate under these conditions in cw15.

For both *sta6* and *cw15*, N starvation lowers the maximum <sup>13</sup>C enrichment and the metabolite turnover rates (Fig. 3-6, 3-7 A) for all the measured metabolites (except G1P, ADP-glucose and PYR). Especially in *sta6*, the maximum <sup>13</sup>C enrichment of 3PG is reduced by more than 2-fold, indicating a significant reduction in carbon fixation upon N starvation. However, in both the strains, the turnover rates of the sugar phosphates remain at par with 3PG (except G1P in *sta6*). This indicates that irrespective of ability/inability to synthesize starch, the turnover of the sugar phosphates leading to G6P remains high.

In *cw15*, G1P turns over at rates significantly higher under –N than under +N conditions (Fig. 3-7 A). Even though in *sta6*, the kinetics of G6P labeling are at par with the 3PG labeling

rate, the turnover rates for G1P do not increase significantly under –N conditions. This indicates a redirection of carbon flux into G6P. G6P can be converted to either G1P or 6PG, depending on relative activities of the enzymes phosphoglucomutase or glucose-6 phosphate dehydrogenase, respectively. 6PG is an intermediate that occurs in the oxidative pentose phosphate (OPP) pathway or the Entner–Doudoroff (ED) pathway. As *Chlamydomonas* lacks the genes for the ED pathway, we attribute the labeling of 6PG to flux through the OPP pathway. Flux through a metabolite is calculated by multiplying the metabolite turnover rate and its pool size. Since G6P is the only <sup>13</sup>C labeled precursor to synthesize G1P or 6PG, the enrichment kinetics and pool sizes of both can be used quantitatively to compare the fluxes between the two metabolites. Both the metabolite pool size and the metabolic turnover rate for 6PG are larger as compared to the pool size and the metabolic turnover rate for G1P (Fig. 3-7 B), indicating a faster carbon influx from G6P into 6PG than into G1P for the *sta6* mutant.

We conclude that in *sta6*, under N starvation a greater fraction of the flux is diverted through the OPP. Activation of the OPP pathway converts G6P to GAP +  $3 \text{ CO}_2 + 3 \text{ NADPH}$ , which in turn delivers additional GAP through lower glycolysis to make AcCoA and MaCoA, the carbon precursors to fatty acids. The OPP pathway generates three times more NADPH than does glycolysis, which is the essential nucleotide cofactor needed to power anabolic reactions, including fatty acid biosynthesis. To test this prediction, we investigated the intrinsic fluorescence emission from NAD(P)H to estimate the relative concentrations of NAD(P)H and NAD(P)<sup>+</sup>, using a previously developed protocol and homebuilt fluorometer (Krishnan *et al.*, 2015).

## 3.4) NADPH measurements:

Figure 3-8 shows the intensity of fluorescence emission at  $420 \mp 10$  nm from intact cells excited by weak light at 365 nm) following different durations of N starvation from 0 to 96 h. Emission intensity is proportional to the concentration of reduced pyridine nucleotides. Samples were concentrated to the same pyridine nucleotide concentration (0.25 µmol ml<sup>-1</sup>). For this the total pool

size of (NADP<sup>+</sup>+ NADPH) was measured at the each time point using LC-MS. The NADPH fluorescence time profile shows an illumination cycle of dark–light–dark induction transients produced by saturating red light (660 nm, Intensity =400  $\mu$ E m<sup>-2</sup> s<sup>-1</sup>) which photoreduces NADP<sup>+</sup> via Photosystem I excitation. The light induced increase provides a direct measure of the relative amount of NADP<sup>+</sup> availability in the dark prior to illumination.



Figure 3-8 **Kinetics of NADPH fluorescence induction**. Intensity of fluorescence emission at wavelength  $420 \mp 10$  nm from *Chlamydomonas reinhardtii* cells following different durations of N starvation from 0 to 96 h. Emission intensity is proportional to the concentration of reduced pyridine nucleotides, NAD(P)H (weak excitation at 365 nm). The time profile shows an illumination cycle of dark–light–dark induction transients produced by saturating red light (660 nm, Intensity =400  $\mu$ E m<sup>-2</sup> s<sup>-1</sup>). Light was turned on at points indicated by upward arrow and turned off at time point indicated by downward arrow. Samples were concentrated to a pyridine nucleotide (NADP<sup>+</sup>+NADPH) concentration 0.25  $\mu$ mol ml<sup>-1</sup>.

The data show that cw15 and sta6 have comparable initial light-induced amplitudes prior to N starvation, indicating comparable NADP<sup>+</sup> concentrations in the dark. As N starvation develops in cw15 the oxidized NADP<sup>+</sup> signal increases by ca 4 fold at 96h, while sta6 shows much smaller changes that decrease oxidized NADP<sup>+</sup>. The data indicate that N starvation causes a significant increase in the relative amount of oxidized NADP<sup>+</sup> in the control strain due to oxidation of NADPH, while sta6 is incapable of draining its reduced NADPH pool or has a higher activity of light independent pathways that reduce the oxidized NADP<sup>+</sup> pool (for example OPP).

## 4. Discussion

*Pathway of carbon assimilation and the carbon redirection nodes under N starvation:* Multiple starchless mutant strains showing enhanced lipid accumulation under N starvation condition have been generated and extensively investigated in the green microalga *C. reinhardtii* (Work *et al.*, 2010); (Blaby *et al.*, 2013); (Goodenough *et al.*, 2014); (Wang *et al.*, 2009); (Goodson *et al.*, 2011); (Siaut *et al.*, 2011); (Breuer *et al.*, 2014); (Li *et al.*, 2010); (Miller *et al.*, 2010). Herein we used time-resolved metabolite profiling as a tool to identify how two such strains (*sta6* and *sta7*) partition photosynthetically fixed carbon away from starch into lipids during N starvation. Scheme 3-1 gives a representation of the global metabolism occurring in the starchless and the starch synthesizing strains as a response to N starvation.



Scheme 3-1 **Global metabolic consequences of nitrogen starvation** (–**N**) **for two strains** as seen by: (A) steady-state concentration changes in key metabolic nodes. To generalize the metabolite trends and report the relative changes in the metabolite levels occurring during N starvation, the highest concentration reached by the metabolite was taken as 100% and the rest of the time points were normalized to this highest value. (B) Changes in downstream terminal product yields, and (C) transient flux changes in major metabolic pathways after 24 h of N starvation, as measured by <sup>13</sup>C-labeling. Black lines represent *cw15*, red lines represent *sta6*. In (C) the major flux change in sta6 through the oxidative pentose phosphate pathway is shown by thicknesses of arrows to represent large and small fluxes through the metabolites, dashed arrows represent fluxes that were not measured.

Time resolved profiling of metabolites shows that upon N starvation three metabolite nodes (flux choke points) exist whose steady-state concentrations change as N starvation develops, and are responsible for the changes in terminal products (proteins, carbohydrates and lipids): a) The Carbohydrate Node at GAP/FBP reports on the allocation of carbon between the upper and lower branches of the glycolytic/gluconeogenic pathways, which feeds into either starch or precursors to fatty acids and proteins; b) The TCA Node at CIT/AKG reports on metabolite fluxes to the biosynthesis of amino acids and from catabolism of proteins; and c) The Fatty Acid Node at Malonyl-CoA reports on metabolite fluxes to fatty acids and from catabolism.

We observe maintenance of gluconeogenic fluxes during N starvation even in mutants blocked at starch polymerization due by redirecting carbon into the OPP pathway via G6P, initially forming GAP and DHAP, and ultimately into fatty acid precursor (Mal-CoA) and a precursor to formation of both phospho-lipids and TAGs (glycerol-3-P) at the expense of the loss of fixed carbon as CO<sub>2</sub>. This rerouting of fixed carbon through OPP generates the additional reductant (NADPH) necessary for biosynthesis of fatty acids, lipids and TAGS, thereby ensuring these soluble carbon metabolites are converted to more safely stored energy-rich fats.

*Central carbon metabolism:* Early in N starvation, there is an initial transient increase in many of the monitored metabolites (Fig. 3-5) and this acts as a preparatory phase needed to switch the metabolic pathways from protein synthesis to starch accumulation. The carbon allocation into lipids and starch are temporally separated (Fig. 3-4, Fig. 3-5) with gluconeogenesis preceding fatty acid (FA) biosynthesis (Fig. 3-3, Fig. 3-4, Scheme 3-1). Transcriptomic studies show a similar - temporal separation between the gluconeogenic and FA pathways (Blaby *et al.*, 2013) in *cw15* under N starvation.

Even in the absence of the ability to synthesize starch and during N starvation, gluconeogenesis occurs (Fig. 3-4, Fig. 3-5, Scheme 3-1) and the temporal separation between the gluconeogenic and the FA pathways is retained. However, the time period between the active

allocation of carbon into the two pathways is significantly reduced in both *sta6* and *sta7* (Fig. 3-4, Fig. 3-5, Scheme 3-1) as opposed to the control strains.

The 'Carbohydrate Node': DHAP and FBP metabolite levels were found to be sensitive reporters of flux changes at the junction of the CBB cycle and gluconeogenesis (Fig. 3-4, Scheme 3-1). On the downstream side, the irreversible hydrolysis reaction of FBP to F6P and inorganic phosphate is catalyzed by FBP phosphatase in green algae (Huppe et al., 1989). The unidirectionality of this reaction means it serves as the gatekeeper for formation of gluconeogenic products. On the downstream side, the greatly depressed levels of FBP and G6P in the starchless strain following N starvation implies FBPase is regulated "on" (its most active form) and these metabolites are being drained by the OPP pathway. This conclusion is confirmed by both the transient <sup>13</sup>C flux measurements and longer term steady-state concentration of 6PG (Figs. 3-5 and 3-6, respectively). On the supply side of this node, concentrations are governed by two reversible enzymes: triose phosphate isomerase (TPI) isomerizes GAP to DHAP, while fructose bisphosphate aldolase (FBA) generates FBP by an aldol condensation of GAP and DHAP. Not much is known about the regulation of TPI and FBA in C. reinhardtii. Interestingly, in the proteomic study by Schmollinger et al. (Schmollinger et al., 2014), TPI and FBA were the only CBB cycle proteins that increase in concentration upon N depletion (70% and 12% respectively), suggesting the importance of this node in directing carbon into gluconeogenesis products in the absence of N.

*Pathway of carbon in starchless mutant strains:* A major objective of our study was to understand how the starchless mutant strains direct a greater fraction of carbon towards TAGs under N depletion. Goodenough et al. (Goodenough *et al.*, 2014) had hypothesized a G6P backflow mechanism in *sta6*, wherein the G6P generated feeds into the plastid-localized OPP and lower glycolysis pathways to generate additional carbon and reductant necessary for TAG biosynthesis. This was proposed to occur under special conditions of "acetate boost" (Goodenough *et al.*, 2014). Our results in *sta6* actually show the occurrence of the OPP pathway during N starvation and that this is not specific to "acetate boost" conditions. The pathway is shown in Scheme 3-1 C.

In response to N stress, the starchless mutant strains channel carbon through gluconeogenesis as a default response. The flux through both the gluconeogenic and the lower-glycolytic route remains high (Fig. 3-6, Fig. 3-7). However, in *sta6*, the carbon forming G6P gets funneled into 6PG at higher flux than G1P under –N conditions. Thus, the redirection of carbon in the starchless mutant strains occurs at the level of G6P (Fig. 3-7, Scheme 3-1). From here, the carbon cycles through the OPP pathway to generate GAP/DHAP that enters into lower glycolysis and into FA biosynthesis. Indeed, the formation of malonyl-CoA (FA node) in *sta6* has an earlier onset than in *cw15*. In addition, DHAP is also a substrate for the synthesis of the TAG backbone, glycerol 3-phosphate. Thus, the recycling of carbon via OPP generates both the precursors for the fatty acid chain, as well as the glycerol backbone of TAGs. We observe that under prolonged N starvation, *sta6* maintains a highly reduced pyridine nucleotide pool, consistent with our data demonstrating elevated OPP flux (~ 3x more reductant generated than glycolysis).

The gluconeogenic and CCB pathways shares a series of four reversible reactions involving 3PG, DHAP, GAP, FBP and F6P. Flux through gluconeogenesis may be necessary to replenish the CBB cycle intermediates and thus sustain carbon fixation. This may be a potential reason why the starchless mutant strains try to maintain a high flux through gluconeogenesis and recycle the fixed carbon through the OPP pathway. However, the use of the OPP pathway to redirect fixed carbon towards lipids has two disadvantages. For every molecule of G6P polymerized to starch one molecule of ATP is consumed to generate ADP-glucose. While the OPP pathway consumes a molecule of G6P by oxidative release of three  $CO_2$  molecules and one 3 C molecule GAP with a net gain of six NADPH molecules ( $3G6P \rightarrow F6P + G6P + GAP + 3 CO2 + 6 NADPH$ ). Thus, while the OPP pathway generates the NADPH needed for FA biosynthesis, it also results in a significant loss in fixed  $CO_2$  and does not regenerate or consume ATP. In addition to the lower levels of photosynthetic carbon fixation (Fig. 3-6 A), the loss of already fixed carbon as  $CO_2$  may be a contributing reason to the substantially lower levels of biomass accumulation in the starchless mutants upon nitrogen starvation (Fig. 3-3).

*ATP regulation of gluconeogenesis:* Reversible protein phosphorylation is known to play an important role in regulation of starch anabolism. Multiple enzymes including starch synthases, starch branching enzymes, phosphoglucose isomerase, AGPase (both small and large subunits) are targets for reversible protein phosphorylation (Tetlow et al., 2004; Tetlow et al., 2008; Grimaud et al., 2008; Heazlewood et al., 2008). The rise in ATP level that occurs before the conversion of gluconeogenic metabolites into starch (Fig. 3-43) may be essential to activate these enzymes and thus activate starch biosynthesis. In this aspect, the potential role of protein phosphorylation may be an important avenue to investigate to increase carbon allocation to desired sinks.

# Chapter 4: Fermentative metabolism in the carbon concentrating mechanism repressor (*ccmR*) mutant of *Synechococcus* sp. PCC 7002

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## Summary

CcmR, a LysR-type transcriptional regulator, represses the genes encoding components of the highaffinity carbon concentration mechanism in cyanobacteria. Unexpectedly, deletion of the *ccmR* gene was found to alter the expression of fermentative genes, especially the hydrogenase operon in the cyanobacterium *Synechococcus sp.* PCC 7002. To understand how CcmR influences fermentative metabolism, the kinetics of glycogen autofermentation were studied in *ccmR* deletion strains of *Synechococcus* 7002 following photoautotrophic growth. We quantified excreted metabolites and monitored the intracellular concentration of NAD(P)H by fluorescence. Inactivation of CcmR resulted in a 50% increase in H<sub>2</sub> production, while continuous removal of H<sub>2</sub> from the medium ("milking") to suppress uptake, boosted the yield by 2.6-fold (20 h) compared to the wild-type strain. Consistent with this greater reductant flux to H<sub>2</sub>, the mutant excreted less lactate (NAD(P)H consuming pathway) during glycogen catabolism. To increase the glycolytic rate and conserve NADH, the *ccmR* mutant was further engineered to introduce GAPDH overexpression (more NADH production) and LDH deletion (less NADH consumption). The triple mutant (*ccmR* deletion + GAPDH overexpression + LDH deletion) showed 6-8-fold greater  $H_2$  yield than the WT strain, achieving conversion rates of 17 nmol  $10^8$  cells<sup>-1</sup> h<sup>-1</sup> and conversion yield of 0.87 H<sub>2</sub> per glucose equivalent (8.9% theoretical maxima). Simultaneous monitoring of the intracellular NAD(P)H concentration and H<sub>2</sub> production rate by these mutants reveals an inverse correspondence between these variables, indicating hydrogenase-dependent H<sub>2</sub> production is a major reductant sink for balancing NAD(P)H energy resources in preference to loss of carbon as excreted lactate from these cells.

## 1. Introduction

Cyanobacteria are found to grow in a diverse range of habitats and are subjected to continuous fluctuations in the inorganic carbon (Ci) availability (Badger et al., 2006). Limited Ci availability can rate-limit photosynthesis and thus growth of cyanobacteria. Therefore, to accommodate for sudden shifts in carbon availability, cyanobacteria have evolved several adaptive mechanisms to maximize CO<sub>2</sub> uptake and/or carbon fixation, which are collectively termed as carbon concentration mechanisms (CCM) (Kaplan et al., 1999), (Kaplan et al., 1991), (Badger et al., 2003), (Price et al., 2008), (Burnap et al., 2015). Transcription factors belonging to the LysR-type transcriptional regulator family (LTTR) are known to regulate expression of genes encoding the structural components of CCM in cyanobacteria in response to dynamic changes in the external Ci availability (Figge et al., 2001), (Omata et al., 2001), (Wang et al., 2004), (Daley et al., 2012), (Burnap et al., 2015). The LTTR family is constituted of both transcriptional activators and repressors, including a) CcmR (Repressor of high affinity Ci uptake genes), b) CmpR (Activator of ABC-type bicarbonate transporter genes) c) vcf30 (Activator of Calvin-Benson-Bassham cycle genes) and d) NtcB (Activator of nitrate assimilation genes) (Figge et al., 2001), (Omata et al., 2001), (Wang et al., 2004), (Daley et al., 2012), (Burnap et al., 2015). Of these, CcmR coordinates the transcriptional repression of genes encoding the high-affinity Ci transporters under Ci sufficiency conditions (Woodger et al., 2007), (Wang et al., 2004), (Figge et al., 2001), (Omata et al., 2001). Important targets for CcmR have been shown to be high-affinity CO<sub>2</sub> transporter genes, ndhF-III, ndhD-III, chpY (cupA) (Woodger et al., 2007), (Wang et al., 2004), inducible HCO<sub>3</sub>transporter genes, *bicA* and *sbtA* and putative HCO<sub>3</sub><sup>-</sup> porin gene, *porB* (Woodger *et al.*, 2007). Loss of *ccmR* gene has been shown to be sufficient to fully induce high affinity CCM even under Ci sufficiency conditions (Woodger et al., 2007). However, the mechanism that CcmR detects the levels of Ci is still not fully understood, which was reported to be regulated by NADPH, 2oxoglutarate or by direct interaction with Ci itself in the freshwater cyanobacterium Synechocystis sp. PCC 6803 (Daley *et al.*, 2012). In summary, CcmR regulates the transition from a basal CCM to a CCM with high affinity for Ci when the external Ci conditions are low.

The photosynthetic metabolism and the genome wide expression pattern in the euryhaline cyanobacterium *Synechococcus sp.* PCC 7002 (henceforth, *Synechococcus* 7002) with inactivated *ccmR* gene has been previously studied (Woodger *et al.*, 2007), (Zhang et al., in preparation). Expression levels of the high affinity CCM genes including *ndhF-III*, *ndhD-III*, *cupA*, *sbtA* and *bicA* were found to be highly up-regulated (Zhang et al., in preparation), (Woodger *et al.*, 2007) in the *ccmR* mutant compared to the wild type strain. Interestingly, together with the CCM related genes, in our studies (Zhang et al., in preparation) the *ccmR* mutant also revealed alterations in mRNA levels of multiple fermentative enzymes. Of note was the 2-9 fold enhanced expression of 12 of the 15 genes related to [Ni-Fe] bidirectional hydrogenase and associated maturation factors (Zhang et al., in preparation). Fig. 1-2 is a schematic representation of the fermentative metabolism in *Synechococcus* 7002.

Therefore, in an effort to understand the metabolic consequences of the *ccmR* gene on fermentation, the kinetics of autofermentation were studied in the *ccmR* mutant as well as the wild type strain of *Synechococcus* 7002 by analyzing the excreted fermentative metabolites, glycolytic rates, cellular energy charge (CEC) and by continuous monitoring of intracellular NAD(P)H levels. Even though lactate production is thermodynamically the most favorable fermentative pathway ( $E_0$  (pyruvate/lactate)=-185mV) for carbon, a greater fraction of the catabolized pyruvate was converted and excreted as acetate in the *ccmR* mutant, demonstrating the preference of ATP producing pathway over the NAD(P)H consuming lactate pathway under autofermentative conditions. However, cellular energy charge did not change significantly, indicating the presence of alternate ATP consuming pathways in the *ccmR* mutant. Continuous and simultaneous monitoring of H<sub>2</sub> and NADPH via milking strategy revealed that the excess redox poise caused by the loss in lactate production was in part dissipated by the higher H<sub>2</sub> production. These results indicate that *ccmR* 

mutant remodels the fermentative metabolism to balance the ATP producing and the consuming pathways while using hydrogenase as a reductant vent. Further, we demonstrate the potential of using the *ccmR* mutant as a platform for metabolically engineering and improving  $H_2$  production by 6-8 fold.

## 2. Experimental Procedures:

#### 2.1) Growth conditions

All the strains were grown photoautotrophically in A<sup>+</sup> medium (Stevens *et al.*, 1973) bubbled with 2% CO<sub>2</sub> in air and supplemented with 2  $\mu$ M NiCl<sub>2</sub>. Cells were incubated at 38°C and illuminated with 150  $\mu$ E m<sup>-2</sup> s<sup>-1</sup> white light. Appropriate antibiotics were added in the following concentrations when needed: spectinomycin (50  $\mu$ g ml<sup>-1</sup>), kanamycin (100  $\mu$ g ml<sup>-1</sup>), gentamycin (20  $\mu$ g ml<sup>-1</sup>) and erythromycin (20  $\mu$ g ml<sup>-1</sup>).

## 2.2) Mutant construction and segregation:

The *ccmR* mutant was generated by deleting the coding sequence of SynPCC7002\_A0171 (*ccmR*) in the wild type *Synechococcus* 7002 strain, and replaced by a DNA fragment encoding an antibiotic resistance gene (*aphAII*, kanamycin resistance) using the homologous recombination method as previously described (Frigaard *et al.*, 2004). The  $\Delta ccmR + gap1^+$  strain was generated using the same approach by deleting the coding sequence of SynPCC7002\_A0171 in the *gap1^+* strain (Kumaraswamy et al., 2013), and replaced with the erythromycin resistance gene (*erm*). Upstream and downstream regions of SynPCC7002\_A0171 were amplified by PCR using primer sets ccmRD1-ccmRD2 and ccmRD3-ccmRD4 (Table 4-1). The *ldhA* gene (SynPCC7002\_G0164) was then deleted in the  $\Delta ccmR + gap1^+$  strain and replaced by a DNA fragment encoding *aacC1*, which confers resistance to gentamycin, to generate the  $\Delta ccmR + gap1^+ + ldhA^-$  strain, using the homologous recombination method as previously described (Frigaard *et al.*, 2004). Primer sets IdhAD1-IdhAD2 and IdhAD3-IdhAD4 (Table 4-1) were used by PCR to amplify the upstream and downstream region of SynPCC7002\_G0164. The complete segregation of alleles for *ccmR* or *ldhA* 

was then verified by PCR using primer set ccmRD1-ccmRD4 or ldhAD1-ldhAD4, with template DNAs from both the wild type strain and mutants. In addition, the *ldh*<sup>-</sup> strain (McNeely *et al.*, 2010) and *gap1*<sup>+</sup> strain (Kumaraswamy *et al.*, 2013) were used as control strains.

## 2.3) Transcriptomic profiling and analysis

Transcriptome profiling of the  $\Delta ccmR$  mutant as well as wild type strain was performed as described (Ludwig *et al.*, 2011). The *ccmR* mutant and wild-type strain were grown in A<sup>+</sup> under standard conditions and harvested at OD<sub>730 nm</sub> = 0.7 (Ludwig *et al.*, 2011). Total RNA was then extracted and rRNA depletion was performed in both strains as described (Gan et al., 2014). The cDNA libraries were then constructed and sequenced (Illumina) in the Genomic Core Facility at The Pennsylvania State University. The BWA software package was used to mapping the sequence reads against the *Synechococcus* sp. PCC 7002 genome (Li *et al.*, 2009). Self-developed scripts were then used to analyze the resulting alignment files and extract expression levels for each gene as described previously (Zhang *et al.*, 2015). The raw sequence data has been deposited to NCBI SRA database with accession number SRR2086086

## 2.4) Oxygen measurements

Light saturated oxygen evolution and dark respiration rates were measured using a membrane covered Clark-type O<sub>2</sub> (*Hansatech*) electrode at 38°C in A<sup>+</sup> media containing 25 mM NaHCO<sub>3</sub>. Samples were illuminated with red light at an intensity of 680  $\mu$ E m<sup>-2</sup> s<sup>-1</sup>. The gross oxygen evolution rate (the total oxygen evolved in light plus the respiration rate) is reported in the results.

## 2.5) In vitro hydrogenase assay

*In vitro* hydrogenase assay, was done as described in (King *et al.*, 2006). Autotrophically grown cells in the exponential phase of growth were washed in  $NO_3^-$  free A<sup>+</sup> medium and concentrated to a density of 10<sup>10</sup> cells/ml. Cells were incubated in anaerobic vials for two hours to activate hydrogenase after which the cells were lysed with BugBuster (EMD Millipore) and incubated with

1mM methyl viologen and 5mM sodium dithionite for 2 hrs. Headspace  $H_2$  was measured by gas chromatography.

#### 2.6) Autofermentation:

For autofermentation experiments, cells were grown to the early exponential phase (~7-8 x  $10^7$  cells ml<sup>-1</sup> and comparable to the conditions under which transcriptomic profiling was performed) at which point, the cells were harvested, washed with A<sup>+</sup> medium lacking NO<sub>3</sub><sup>-</sup> and resuspended in the same volume of NO<sub>3</sub><sup>-</sup> free A<sup>+</sup> medium and supplemented with 2 µM NiCl<sub>2</sub>. 4ml aliquots were placed into anaerobic vials. The headspace air in the vials was replaced with argon and autofermented for 5 days in the dark.

## 2.7) Analytical Assays

Hydrogen in the headspace of the assay vials were sampled with a gas-tight syringe and measured by gas chromatography using Perkin-Elmer Clarus 680 series equipped with a thermal conductivity detector and using argon as the carrier gas. After measuring the headspace gases, 1 ml of cells from the fermentation vials was harvested and the spent media was separated from the cells by centrifugation. The cell pellet was analyzed for total reducing carbohydrate content using anthrone reagent (Hassid *et al.*, 1957) as described previously (McNeely *et al.*, 2010). The cell-free supernatant was analyzed by Perkin-Elmer HPLC with a Rezex ROA column (300 x 7.8 mm, Phenomenex) equipped with UV detection at 210 nm for excreted fermentation metabolites including lactate, acetate and succinate. All of the above steps were done at 24, 72 and 120h after the onset of fermentation.

#### 2.8) LC-MS

Adenylate and pyridine nucleotide levels were quantified using LC-MS as described previously (McNeely *et al.*, 2010) after 3 days of autofermentation. Briefly, aliquots of cultures from the fermentation vials were subjected to rapid vacuum filtration and the filter with cells were submerged and quenched in 80:20 MeOH:Water mixture incubated on dry ice. Cells were scrapped

off the filters and the solvent with cells was transferred into microcentrifuge tubes and incubated at -20°C for 20 min for metabolite extraction. The supernatant with the metabolites was separated from the cells by centrifugation and vacuum dried (Labconco Centri-Vap concentrator). The vacuum dried pellet was resuspended in LC-MS grade water and analyzed using Agilent 1200 series HPLC coupled to a 6490 QQQ mass analyzer equipped with an ion spray source. The acquired data was analyzed using Agilent Mass Hunter software.

## 2.9) Real-time $H_2$ and NAD(P)H detection

Real-time dissolved H<sub>2</sub> and NAD(P)H were determined using a home-built reverse Clark-type electrode coupled to a fluorometer capable of detecting NAD(P)H fluorescence as described in (Ananyev *et al.*, 2012). The Clark-electrode continuously consumes the generated H<sub>2</sub> and therefore, the electrode measures the rate of H<sub>2</sub> production rather than the concentration of dissolved H<sub>2</sub>. The rate is presented as electrical current (nA) and is converted to gas nmoles using Faraday's second law of electrolysis. For instantaneous NAD(P)H measurement the sample was illuminated by pulsed UV light ( $\lambda$ max=365 nm, 1000  $\mu$ W/cm<sup>2</sup>, T=250ms) and recorded by a photodiode. The fluorescence emission was filtered by an emission filter ( $\lambda$ max= 450 nm). For measurements, *Synechococcus spp.* 7002 cells were taken from photoautotrophically growing cultures at a density of 7-8 x 10<sup>7</sup> cells/ml and concentrated to 10<sup>10</sup> cells/ml and transferred to sample cavity of the electrode. The sample was covered with a coverslip to prevent diffusion of air into the cavity. Cellular respiration consumes the internal O<sub>2</sub> pool and creates anoxic conditions within 2-3 minutes. Experiments were performed in triplicates. All figures presented in the results here are a representative single trace.

## 3. Results

## 3.1) Mutant construction

To investigate the possible implications of CcmR on fermentative metabolism and  $H_2$  production, a *ccmR* deletion mutant of *Synechococcus* 7002 ( $\triangle ccmR$ ) was constructed (Fig. 4-1 A) and the complete segregation of the alleles was confirmed (Fig. 4-1 B).



Figure 4-1 Scheme for mutant construction and verification of mutants. A) General scheme showing the homologous recombination methods by which the target gene is replaced with antibiotic resistance cassettes. In detail, the upstream and downstream flanking regions for a gene of interest was amplified by PCR using primer pairs D1-D2 and D3-D4, after which these fragments were ligated to the antibiotic cassette. The resulting construction was then used to generate fully segregated mutants as described in the Materials and Methods. B) Agarose gel electrophoretic analysis of PCR products using template DNA derived from wild type (WT) or  $\Delta ccmR$  mutants, verifying the complete segregation of alleles of the interested gene in mutants, lane 1  $\Delta ccmR$ , using primer set ccmRD1-ccmRD4
Name	Sequence (5' to 3')
ccmRD1	ATCTAGGGCTTGGCGATCGCGGATAAAACT
ccmRD2	TAGGGTCGACGGGTTTACTCCTGAACTTGT
ccmRD3	GCCAGAATTCAAGCTCAAAATTGTTGGGCC
ccmRD4	TCAAGCCAGAGGTGTAAAGGAAATTGGGCG
ldhAD1	CAGACCCAGAAGACCAGAGGCGCTTAAGCT
ldhAD2	TATGATGTCGACGGTGCTTTGGGGGTAATGG
ldhAD3	GCGCTGGAATTCAAACTGCGCCAAGAATAG
ldhAD4	TAGGATCAATTTACGTCTTTGTTGGCGCAA

Table 4-1 Primer sequences used for the construction of mutants.



Figure 4-2 **Photoautotrophic phenotypes for**  $\Delta ccmR$  and Wt.A) Growth curve of  $\Delta ccmR$  and Wt. Data represents the mean of three biological replicates with standard error grown under 150  $\mu$ E m<sup>-2</sup>s<sup>-1</sup> illumination with constant bubbling of 2% CO<sub>2</sub>. The inset shows the calculated doubling time based on fits to Gompertz curve. B) Gross photosynthetic oxygen evolution rate and C) respiration rate measured at 38°C under light saturating conditions of 680  $\mu$ E m<sup>-2</sup>s<sup>-1</sup> of red light and in the presence of 25 mM bicarbonate.

Fig. 4-2 A compares photoautotrophic growth of the  $\Delta ccmR$  strain and the wild-type (WT) strain when grown under an irradiance of 150 µE m<sup>-2</sup>s<sup>-1</sup> (white fluorescent light) and with constant bubbling with 2% CO<sub>2</sub> in air. The growth-rates were not significantly different between the mutant (T <sub>doubling</sub>: 3.6 ± 0.3 h) and the WT (T <sub>doubling</sub>: 3.8 ± 0.4 h) strains. The light-saturated (680 µE m<sup>-2</sup>s<sup>-1</sup> red light) photosynthetic oxygen evolution rate (Fig. 4-2 B) and the respiration rates (Fig. 4-2 C) were also measured. The oxygen evolution rate was lower in  $\Delta ccmR$  (720 ± 21.5 µmol O<sub>2</sub> (mg Chl)<sup>-1</sup> (h)<sup>-1</sup>) vs. in WT (850 ± 35.6 µmol O<sub>2</sub> (mg Chl)<sup>-1</sup> (h)<sup>-1</sup>). By contrast, the respiration rate was higher in the  $\Delta ccmR$  mutant (89 ± 6.02 µmol O<sub>2</sub> (mg Chl)<sup>-1</sup> (h)<sup>-1</sup> vs. in WT (59 ± 7.22 µmol O<sub>2</sub> (mg Chl)<sup>-1</sup> (h)<sup>-1</sup>). These results suggest that under high light and Ci concentration, the loss of CcmR does not confer a growth/photosynthetic advantage (or disadvantage) and instead shows a 15% decrease in the photosynthetic oxygen evolution rate and a 50% increase in the respiration rate.

#### 3.2) Transcript profiling:

To gain a deeper understanding of the possible relations between CcmR, fermentative metabolism and other metabolic pathways, rRNA-depleted total RNAs from the  $\Delta ccmR$  and WT strain were used for transcript profiling. The whole cell transcriptomic profiling was done with cells grown under photoautotrophic and Ci sufficient conditions. For analysis, an expression cut-off threshold of 2-fold higher or lower than the expression in the WT in combination with a *p*-value <0.05 was used. A few important genes based on the metabolic pathways were extracted and are displayed in Table 4-2.

	N	( D	Fold Change		Gene		
Gene name	w I	∆ccmR	( <i>∆ccmR</i> / WT)	p-value	Symbol		Metabolic pathway
Hydrogenase							
SYNPCC7002 _A0194	1.13E-05	1.76E-05	1.566	0.05-0.02	hypE	hydrogenase expression/formation protein HypE	H2ase
<i>SYNPCC7002</i> _ <i>A0195</i>	3.26E-05	0.000188424	5.773	<0.001	hoxE	hydrogenase subunit E ([NiFe] hydrogenase subunit)	H2ase/complex1
SYNPCC7002 _A0196	4.25E-05	0.000169283	3.987	< 0.001	hoxF	hydrogenase large diaphorase subunit F	H2ase/complex1
SYNPCC7002 _A0197	2.28E-05	6.13E-05	2.687	< 0.001	hoxU	hydrogenase small diaphorase subunit U	H2ase/complex1
SYNPCC7002 _A0198	2.11E-05	5.68E-05	2.695	< 0.001	hoxY	hydrogenase small subunit Y	H2ase
SYNPCC7002 _A0199	2.80E-05	4.82E-05	1.719	<0.001	hyp3	CBS domain-containing protein; possibly associated with hydrogenase assembly	H2ase
SYNPCC7002 _A0200	4.39E-05	7.75E-05	1.764	< 0.001	hoxH	hydrogenase large subunit H	H2ase
SYNPCC7002 _A0201	1.65E-05	2.39E-05	1.453	0.05-0.02	hoxW	hydrogenase maturation protease	H2ase
SYNPCC7002 _A0202	5.20E-06	1.17E-05	2.243	0.01-0.001	hypA	hydrogenase nickel insertion protein HypA	H2ase
SYNPCC7002 _A0203	1.53E-05	2.69E-05	1.758	< 0.001	hypB	hydrogenase isoenzymes formation protein; HypB	H2ase
SYNPCC7002 _A0204	1.56E-05	3.62E-05	2.320	< 0.001	hypF	[NiFe] hydrogenase maturation protein HypF	H2ase
SYNPCC7002 _A0205	1.56E-05	4.43E-05	2.838	< 0.001	hypC	hydrogenase assembly chaperone HypC/HupF	H2ase
SYNPCC7002 _A0206	6.35E-05	0.000148048	2.330	<0.001	hypD	hydrogenase expression/formation protein HypD	H2ase

SYNPCC7002 _A1746	7.16E-05	0.000172573	2.409	< 0.001	hypA	hydrogenase nickel insertion protein	H2ase
<i>SYNPCC7002</i> _ <i>A1747</i>	8.38E-05	0.000170479	2.035	< 0.001	hypB	hydrogenase accessory protein HypB	H2ase
SYNPCC7002 _A1849	0.000107742	8.43E-05	0.783	0.01-0.001	lexA	LexA repressor	transcription/H2ase

## CCM related genes

SYNPCC7002 _A0171	0.000132872	3.89E-06	0.029	<0.001	rbcR	transcription regulator of rubisco operon, RbcR (LysR family)	CO2-conc- fix/transcription
SYNPCC7002 _A0172	8.61E-05	0.000723789	8.409	<0.001	ndhF-III	NADH2 dehydrogenase (plastoquinone) chain 5	CO2-conc-fix
<i>SYNPCC7002</i> _ <i>A0173</i>	4.33E-05	0.000380139	8.774	<0.001	ndhD-III	proton-translocating NADH-quinone oxidoreductase (NADH dehydrogenase subunit 4)	CO2-conc-fix
SYNPCC7002 _A0174	3.93E-05	0.000282337	7.187	< 0.001	cupA	CO2 hydration protein	CO2-conc-fix
SYNPCC7002 _A0175	1.99E-05	0.000138477	6.948	<0.001	'cupS'	conserved hypothetical protein	CO2-conc-fix
SYNPCC7002 _A0470	3.38E-05	0.008090281	239.388	< 0.001	'sbtA'	sodium-dependent bicarbonate transporter	CO2-conc-fix
SYNPCC7002 _A0690	0.000133161	0.000116644	0.876	0.1-0.05	-	probable bicarbonate transporter, ICT family protein	CO2-conc-fix
<i>SYNPCC7002</i> _ <i>A1414</i>	0.000274987	0.000195901	0.712	< 0.001	ppc	phosphoenolpyruvate carboxylase	CO2-conc-fix
SYNPCC7002 _A1796	0.002537278	0.000720798	0.284	<0.001	rbcS	Ribulose bisphosphate carboxylase, small subunit	CO2-conc- fix/calvinCycle
SYNPCC7002 _A1797	0.00316582	0.000951094	0.300	<0.001	rbcX	RbcX protein, possible rubisco chaperone	CO2-conc-fix
SYNPCC7002 _A1798	0.009051184	0.002814401	0.311	<0.001	rbcL	ribulose-1,5-bisphosphate carboxylase, large subunit	CO2-conc- fix/calvinCycle

<i>SYNPCC7002</i> _ <i>A1799</i>	0.000334491	0.000109466	0.327	<0.001	ccmN	Carbon dioxide concentrating mechanism protein	CO2-conc-fix
<i>SYNPCC7002</i> _ <i>A1800</i>	0.002824397	0.000850003	0.301	<0.001	ccmM	carbon dioxide concentrating mechanism protein	CO2-conc-fix
SYNPCC7002 _A1801	0.000615544	0.000154927	0.252	<0.001	ccmL	carbon dioxide concentrating mechanism protein	CO2-conc-fix
<i>SYNPCC7002</i> _ <i>A1802</i>	0.000599657	0.000164497	0.274	<0.001	ccmK	Carbon dioxide concentrating mechanism protein	CO2-conc-fix
SYNPCC7002 _A1803	0.000575682	0.000153132	0.266	<0.001	ccmK	carbon dioxide concentrating mechanism protein	CO2-conc-fix
SYNPCC7002 _A1805	0.001395156	0.000960067	0.688	< 0.001	ndhF4	NADH dehydrogenase subunit F4	CO2-conc-fix
SYNPCC7002 _A1806	0.000811097	0.000611033	0.753	< 0.001	ndhD4	NADH dehydrogenase subunit D4	CO2-conc-fix
SYNPCC7002 _A1807	0.000482383	0.000393298	0.815	< 0.001	'cupB'	CO2 hydration protein	CO2-conc-fix
SYNPCC7002 A1997	8.32E-05	4.91E-05	0.590	< 0.001	icfA	carbonic anhydrase	CO2-conc-fix
SYNPCC7002 _A2025	4.45E-05	3.38E-05	0.760	0.05-0.02	-	conserved hypothetical protein	CO2-conc-fix
SYNPCC7002 _A2371	0.000123051	0.012784438	103.895	< 0.001	bicA	bicarbonate transporter, BicA	CO2-conc-fix
<i>SYNPCC7002</i> _ <i>A2389</i>	0.000175333	0.000122027	0.696	<0.001	ccmk1	carbon dioxide concentrating mechanism protein	CO2-conc-fix
SYNPCC7002 _A2612	4.13E-05	3.17E-05	0.768	0.05-0.02	ccmK	carbon dioxide concentrating mechanism protein	CO2-conc-fix
SYNPCC7002 _A2613	0.000140382	8.31E-05	0.592	<0.001	ccmK	carbon dioxide concentrating mechanism protein	CO2-conc-fix

Uxidative pentose phosphate pathway										
SYNPCC7002 _A0010	4.22E-05	5.44E-05	1.291	0.05-0.02	fbaB	fructose-bisphosphate aldolase class I	glycolysis/pentosePpathw ay/calvinCycle			
<i>SYNPCC7002</i> _ <i>A0130</i>	3.32E-05	2.45E-05	0.738	0.05-0.02	eda	2-dehydro-3- deoxyphosphogluconate aldolase/4-hydroxy-2- oxoglutarate aldolase	pentosePpathway			
SYNPCC7002 _A0162	0.000230215	0.000460892	2.002	< 0.001	pfkA	6-phosphofructokinase PfkA	glycolysis/pentosePpathw ay			
<i>SYNPCC7002</i> _ <i>A0221</i>	0.000238881	0.001024968	4.291	<0.001	gnd	6-phosphogluconate dehydrogenase, decarboxylating	pentosePpathway			
SYNPCC7002 _A0324	0.000517045	0.000242559	0.469	< 0.001	rpe	ribulose-phosphate 3- epimerase	pentosePpathway			
SYNPCC7002 _A0329	0.000241769	0.000439657	1.818	< 0.001	fbp	fructose-1,6- bisphosphatase	glycolysis/pentosePpathw ay/calvinCycle			
SYNPCC7002 _A0928	9.56E-05	0.000177956	1.861	< 0.001	pgl	6- phosphogluconolactonase	pentosePpathway			
SYNPCC7002 _A0964	0.000240036	0.000380438	1.585	<0.001	pgi	glucose-6-phosphate isomerase	glycolysis/pentosePpathw ay/StarchSucroseMetaboli sm			
SYNPCC7002 _A1022	0.002646753	0.002032889	0.768	< 0.001	tkt	transketolase	pentosePpathway/calvinC ycle			
SYNPCC7002 _A1269	0.000144137	5.95E-05	0.413	< 0.001	rpiA	ribose 5-phosphate isomerase A	pentosePpathway/calvinC ycle			
SYNPCC7002 _A1301	0.001543337	0.00063825	0.414	<0.001	-	Bacterial fructose-1,6- bisphosphatase, glpX- encoded superfamily	glycolysis/pentosePpathw ay/calvinCycle			
<i>SYNPCC7002</i> _ <i>A1352</i>	0.004438214	0.002276943	0.513	<0.001	fba	fructose-bisphosphate aldolase, class II, Calvin cycle subtype	glycolysis/pentosePpathw ay/calvinCycle			
<i>SYNPCC7002</i> _ <i>A1459</i>	9.68E-05	0.00063825	6.596	< 0.001	zwf	glucose-6-phosphate 1- dehydrogenase	pentosePpathway			
<i>SYNPCC7002</i> _ <i>A1460</i>	0.000175044	0.001153575	6.590	< 0.001	tal	transaldolase	pentosePpathway			

#### **A**---: **J** = **4**:hambata nath . .

SYNPCC7002	4.74E-05	2.36E-05	0.499	< 0.001	talC	transaldolase	pentosePpathway
_M2330 SYNPCC7002 _A2726	0.000650495	0.000381036	0.586	<0.001	prs	ribose-phosphate pyrophosphokinase	pentosePpathway/Purine Metabolism
Glycolysis							
SYNPCC7002 _A0010	4.22E-05	5.44E-05	1.291	0.05-0.02	fbaB	fructose-bisphosphate aldolase class I	glycolysis/pentosePpathw ay/calvinCycle
SYNPCC7002 _A0073	0.000434723	0.000186929	0.430	< 0.001	eno	2-phosphopyruvate hydratase (enolase)	glycolysis
SYNPCC7002 _A0106	0.001326698	0.000519214	0.391	<0.001	gap	glyceraldehyde-3- phosphate dehydrogenase, type I	glycolysis/calvinCycle
SYNPCC7002 _A0162	0.000230215	0.000460892	2.002	<0.001	pfkA	6-phosphofructokinase PfkA	glycolysis/pentosePpathw ay
SYNPCC7002 _A0329	0.000241769	0.000439657	1.818	<0.001	fbp	fructose-1,6- bisphosphatase	glycolysis/pentosePpathw ay/calvinCycle
SYNPCC7002 _A0964	0.000240036	0.000380438	1.585	<0.001	pgi	glucose-6-phosphate isomerase	glycolysis/pentosePpathw ay/StarchSucroseMetaboli sm
SYNPCC7002 _A1301	0.001543337	0.00063825	0.414	<0.001	-	Bacterial fructose-1,6- bisphosphatase, glpX- encoded superfamily	glycolysis/pentosePpathw ay/calvinCycle
SYNPCC7002 _A1352	0.004438214	0.002276943	0.513	<0.001	fba	fructose-bisphosphate aldolase, class II, Calvin cycle subtype	glycolysis/pentosePpathw ay/calvinCycle
SYNPCC7002 _A1585	0.001736868	0.000665766	0.383	<0.001	pgk	phosphoglycerate kinase	glycolysis/calvinCycle
SYNPCC7002 _A2233	0.001014449	0.000605949	0.597	<0.001	gpm	2,3-bisphosphoglycerate- independent phosphoglycerate mutase	glycolysis
SYNPCC7002 _A2560	7.13E-05	3.62E-05	0.507	< 0.001	gpmB	phosphoglycerate mutase	glycolysis
SYNPCC7002 _A2697	2.02E-05	0.000327798	16.212	<0.001	gap	glyceraldehyde-3- phosphate dehydrogenase, type I	glycolysis

<b>Pyruvate</b>	Metabol	lism
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<i>SYNPCC7002</i> _ <i>A0110</i>	0.000303584	0.000195602	0.644	<0.001	-	dihydrolipoamide S- acetyltransferase; 2-oxo acid dehydrogenases acyltransferase (catalytic domain)	pyruvate
<i>SYNPCC7002</i> _ <i>A0250</i>	0.000266322	0.000955879	3.589	< 0.001	ppsA	phosphoenolpyruvate synthase	pyruvate
SYNPCC7002 _A0291	0.000260545	0.00023568	0.905	0.05-0.02	-	conserved hypothetical PEP-utilizing protein	pyruvate
SYNPCC7002 _A0353	0.00028943	0.000144758	0.500	< 0.001	'pdhA'	pyruvate dehydrogenase E1 component, alpha chain	pyruvate
SYNPCC7002 _A0655	0.000236859	0.000151637	0.640	< 0.001	pdhB	pyruvate dehydrogenase E1 beta chain	pyruvate
SYNPCC7002 _A1443	6.41E-05	0.000370568	5.779	<0.001	nifJ	pyruvate:ferredoxin (flavodoxin) oxidoreductase	pyruvate
SYNPCC7002 _A1658	0.000280764	0.000175863	0.626	< 0.001	pyk	pyruvate kinase	pyruvate

Calvin Cycle

SYNPCC7002 _A0010	4.22E-05	5.44E-05	1.291	0.05-0.02	fbaB	fructose-bisphosphate aldolase class I	glycolysis/pentosePpathw ay/calvinCycle
<i>SYNPCC7002</i> _ <i>A0106</i>	0.001326698	0.000519214	0.391	<0.001	gap	glyceraldehyde-3- phosphate dehydrogenase, type I	glycolysis/calvinCycle
SYNPCC7002 _A0162	0.000230215	0.000460892	2.002	< 0.001	pfkA	6-phosphofructokinase PfkA	glycolysis/pentosePpathw ay
SYNPCC7002 _A0329	0.000241769	0.000439657	1.818	< 0.001	fbp	fructose-1,6- bisphosphatase	glycolysis/pentosePpathw ay/calvinCycle
SYNPCC7002 A0595	0.000189198	0.000126214	0.667	< 0.001	tpiA	triosephosphate isomerase	calvinCycle

SYNPCC7002 _A1301	0.001543337	0.00063825	0.414	< 0.001	-	Bacterial fructose-1,6- bisphosphatase, glpX- encoded superfamily	glycolysis/pentosePpathw ay/calvinCycle
SYNPCC7002 _A1352	0.004438214	0.002276943	0.513	<0.001	fba	fructose-bisphosphate aldolase, class II, Calvin cycle subtype	glycolysis/pentosePpathw ay/calvinCycle
SYNPCC7002 _A1585	0.001736868	0.000665766	0.383	< 0.001	pgk	phosphoglycerate kinase	glycolysis/calvinCycle
SYNPCC7002 _A1796	0.002537278	0.000720798	0.284	< 0.001	rbcS	Ribulose bisphosphate carboxylase, small subunit	CO2-conc- fix/calvinCycle
SYNPCC7002 _A1798	0.009051184	0.002814401	0.311	< 0.001	rbcL	ribulose-1,5-bisphosphate carboxylase, large subunit	CO2-conc- fix/calvinCycle
SYNPCC7002 _A2665	0.002403539	0.001985334	0.826	< 0.001	prk	phosphoribulokinase	calvinCycle
<i>SYNPCC7002</i> _ <i>A2857</i>	0.000419991	0.000256018	0.610	< 0.001	prk	Phosphoribulokinase	calvinCycle

## NADPH dehydrogenase

SYNPCC7002 _A0195	3.26E-05	0.000188424	5.773	<0.001	hoxE	hydrogenase subunit E ([NiFe] hydrogenase subunit)	H2ase/complex1
SYNPCC7002 _A0196	4.25E-05	0.000169283	3.987	<0.001	hoxF	hydrogenase large diaphorase subunit F	H2ase/complex1
SYNPCC7002 _A0197	2.28E-05	6.13E-05	2.687	<0.001	hoxU	hydrogenase small diaphorase subunit U	H2ase/complex1
SYNPCC7002 _A0560	2.34E-05	1.35E-05	0.575	0.01-0.001	ndhL	NADH dehydrogenase subunit L (inorganic carbon tranpsort protein)	complex1
SYNPCC7002 _A0854	0.001572223	0.001497524	0.952	0.02-0.01	ndhF	NADH dehydrogenase subunit 5	complex 1
SYNPCC7002 _A0923	0.000186599	0.000220427	1.181	0.01-0.001	ndhE	NADH dehydrogenase subunit E	complex 1

SYNPCC7002 _A0924	0.000300406	0.000351725	1.171	< 0.001	ndhG	NADH dehydrogenase subunit G	complex1
SYNPCC7002 _A0926	0.000461008	0.000545234	1.183	< 0.001	ndhA	NADH dehydrogenase subunit A	complex 1
SYNPCC7002 _A1143	0.000142404	0.000122625	0.861	0.05-0.02	'ndhN'	conserved hypothetical protein	complex 1
<i>SYNPCC7002</i> _ <i>A1432</i>	3.29E-05	5.98E-05	1.817	<0.001	-	Na+-transporting NADH:ubiquinone oxidoreductase subunit 2	complex1?
SYNPCC7002 _A1973	6.07E-05	8.82E-05	1.455	< 0.001	ndhD2	NADH dehydrogenase subunit D2	complex 1
SYNPCC7002 _A2000	0.001096194	0.001832202	1.671	< 0.001	ndhD1	NADH dehydrogenase subunit D1	complex 1
SYNPCC7002 _A2311	0.000279609	0.000122326	0.437	<0.001	ndbA	type II NADH dehydrogenase A	complex1-B
SYNPCC7002 _A2327	0.000178222	9.66E-05	0.542	< 0.001	ndhD	NADH dehydrogenase subunit 4	complex1
SYNPCC7002 _A2541	0.000608612	0.000524896	0.862	< 0.001	ndhH	NADH dehydrogenase subunit H	complex 1
SYNPCC7002 _A2547	0.001838255	0.001596522	0.868	< 0.001	ndhB	NADH dehydrogenase subunit B	complex 1
SYNPCC7002 _A2748	0.000179666	0.000234484	1.305	< 0.001	ndhC	NADH dehydrogenase subunit C	complex 1
SYNPCC7002 _A2749	0.000434434	0.000523401	1.205	< 0.001	ndhK	NADH dehydrogenase subunit K	complex 1
SYNPCC7002 _A2750	0.000303584	0.000397186	1.308	< 0.001	ndhJ	NADH dehydrogenase subunit J	complex1

### Terminal oxidase

SYNPCC7002	2.05E-05	4.19E-05	2.042	< 0.001	ctaEII	cytochrome oxidase II	cytOxidase
_A0725						small subunit	
SYNPCC7002 _A0726	0.000172445	0.000275458	1.597	< 0.001	ctaDII	cytochrome oxidase II large subunit	cytOxidase
<i>SYNPCC7002</i> _ <i>A0727</i>	6.01E-05	7.69E-05	1.279	0.01-0.001	ctaCII	Cytochrome C oxidase subunit II	cytOxidase

SYNPCC7002 _A1162	0.000238881	0.000837441	3.506	< 0.001	ctaCI	cytochrome oxidase subunit II	cytOxidase
SYNPCC7002 _A1163	0.000267477	0.000804243	3.007	<0.001	ctaDI	cytochrome oxidase large subunit (subunit I)	cytOxidase
SYNPCC7002 _A1164	0.000129983	0.00028473	2.191	<0.001	ctaEI	cytochrome oxidase small subunit (subunit III)	cytOxidase
ATP synthase							
SYNPCC7002 _A0733	0.001631726	0.001300726	0.797	< 0.001	atpG	ATP synthase F1, gamma subunit	ATPase
SYNPCC7002 _A0734	0.002833929	0.002185124	0.771	< 0.001	atpA	ATP synthase F1, alpha subunit	ATPase
SYNPCC7002 _A0735	0.000810519	0.000591293	0.730	< 0.001	atpH	ATP synthase F1, delta subunit	ATPase
SYNPCC7002 _A0736	0.000904396	0.000637353	0.705	< 0.001	atpF	ATP synthase B chain (Subunit I)	ATPase
SYNPCC7002 _A0737	0.00100925	0.000731266	0.725	<0.001	atpG	ATP synthase B chain (Subunit II)	ATPase
SYNPCC7002 _A0738	0.001033513	0.000750407	0.726	<0.001	atpE	ATP synthase C chain (Lipid-binding protein)	ATPase
SYNPCC7002 _A0739	0.002569052	0.0019638	0.764	<0.001	atpB	ATP synthase F0, A subunit	ATPase
<i>SYNPCC7002</i> _ <i>A0740</i>	0.00240845	0.002217425	0.921	< 0.001	atp1	ATP synthase subunit I	ATPase
SYNPCC7002 _A0749	0.003338554	0.002128297	0.637	< 0.001	atpD	ATP synthase beta chain	ATPase
SYNPCC7002 _A0750	0.000681114	0.000377148	0.554	<0.001	atpC	ATP synthase F1, epsilon subunit	ATPase
SYNPCC7002 _G0144	8.69E-05	0.000287721	3.309	<0.001	atpD	ATP synthase F1, beta subunit	ATPase
SYNPCC7002 _G0145	2.48E-05	6.73E-05	2.709	<0.001	atpC-II	ATP synthase epsilon subunit	ATPase
SYNPCC7002 _G0146	1.99E-05	5.77E-05	2.896	<0.001	-	F0F1-ATPase subunit, putative	ATPase
SYNPCC7002 _G0148	3.99E-05	0.000107073	2.686	<0.001	atpB-II	ATP synthase F0, A subunit	ATPase

SYNPCC7002 _G0149	2.22E-05	4.46E-05	2.004	< 0.001	atpH-II	ATP synthase c subunit	ATPase
SYNPCC7002 _G0150	6.56E-05	9.90E-05	1.510	< 0.001	atpF-II	ATP synthase b subunit	ATPase
SYNPCC7002 _G0151	6.15E-05	8.64E-05	1.405	< 0.001	atpA-II	ATP synthase F1, alpha subunit	ATPase

Color ording .	Fold Change ≥2 fold	Fold Change ≤2 fold	
Color couling :	colored in red	colored in blue	

Table 4-2: Transcript levels for selected protein-coding genes for WT and  $\Delta ccmR$ .

<u>Fermentative metabolism and Pyruvate metabolism</u>: Even though the transcriptomic profiling was done under photoautotrophic growth conditions, expression of multiple genes of the fermentative metabolism were differentially expressed. The expression levels of genes encoding subunits of the hydrogenase (Hox) and associated chaperone proteins (10 of the 16 known genes) were significantly enhanced by 2-6 fold (Table 4-2). The hydrogenase genes are co-localized in one operon and the increase in the transcription of multiple genes of this operon suggests a clear regulatory activity of CcmR on this operon.

Under fermentative conditions pyruvate, the end product of glycolysis, is an important intermediate of glycogen catabolism (Fig. 1-2). Pyruvate can be synthesized from precursor phosphoenolpuruvate by either pyruvate kinase (pk) which produces ATP from ADP or by phosphoenol puruvate synthase (*ppsA*) that generates ATP from AMP and 2Pi. The expression level of *ppsA* is enhanced by 3.5-fold with unchanged levels of *pk* in the mutant, suggesting *ppsA* being the preferred pathway for the conversion of phosphoenol pyruvate to pyruvate. Once generated, pyruvate may be converted to lactate via lactate dehydrogenase (*ldh*) or to acetyl-CoA by the action of either a) pyruvate: ferred oxin oxidored uctase (pfor), or via pyruvate dehydrogenase (*pdh*), both of which decarboxylate pyruvate to acetyl-CoA. PDH reduces NAD<sup>+</sup>, while PFOR reduces ferredoxin. Of these, PFOR is sensitive to oxygen and was found to be the dominant enzyme of the two under anaerobic (fermentative) conditions (McNeely et al., 2011). The transcript level of PFOR (also called *nifJ*, SYNPCC7002\_A1443) is 5.7-fold higher, while the LDH transcript remains unchanged in the *ccmR* mutant. This observation suggests that catabolic generation of  $CO_2$ and Fd<sub>red</sub> with retention of reduced carbon (as acetyl-CoA) is the preferred regulatory response to the lack of sufficient intracellular  $C_i$  (as opposed to production of the lower energy NADPH reductant via PDH). Interestingly, the expression of the gap1 gene involved in glycolysis is 16fold higher in the  $\Delta ccmR$  strain than WT. This reversible enzyme is the major site for NADH/NAD<sup>+</sup> dependent carbon fluxes going into gluconeogenesis and glycolytic catabolism.

**Oxidative phosphorylation:** The transcriptome reveals a considerable remodeling of the respiratory metabolism under photoautotrophic Ci sufficient conditions.sdhB (SYNPCC7002 A1094), one of the two subunits of respiratory Complex II and 4 of the 6 genes of terminal oxidase (Complex IV) show at least 2-fold or higher expression levels in the  $\Delta ccmR$  strain (Table 4-2). The  $\Delta ccmR$  strain also shows increased transcript levels (>2-fold) for genes coding for the Na<sup>+</sup>-dependent ATPase subunits: atpD (SYNPCC7002\_G0144), atpC-II (SYNPCC7002\_G0145), atpB-II (SYNPCC7002\_G0148) and *atpH*-II (SYNPCC7002\_G0149). This expression difference of the oxidative phosphorylation genes together with the 50% higher respiration rate in the mutant (Fig. 4-2 C) indicates a significantly greater reliance on respiration for energy production in the  $\Delta ccmR$ . Calvin-Benson-Bassham (CBB) Cycle: The genes belonging to the CBB cycle, including the small and large subunits of RuBisCO, rbcS (SYNPCC7002\_A1796) and rbcL (SYNPCC7002\_A1798) show a 3.5 and 3.2-fold lower expression levels in the  $\Delta ccmR$  mutant under photoautotrophic Ci sufficient condition. The implication of this finding could be that intracellular  $CO_2$  levels are sufficient to induce normal expression levels of these inducible genes. CO<sub>2</sub> generated from the higher respiratory flux may play such a role.

<u>Membrane Transporters:</u> The transcriptome also suggests considerable remodeling of the ion transport system in the  $\triangle ccmR$  strain. The mRNA levels for genes encoding bicarbonate transport systems *bicA* (bicarbonate transporter: SYNPCC7002\_A2371), *sbtA* (sodium-dependent bicarbonate transporter: SYNPCC7002\_A0470) and the Na<sup>+</sup>/H<sup>+</sup> antiporter (SYNPCC7002\_A2372-A2375 and SYNPCC7002\_A2380) are enhanced by 103-, 240- and 100-fold respectively. This, together with the increase in the expression of the Na<sup>+</sup>-dependent ATPase show an internal compensatory mechanism to overcome the possible excess sodium import caused by sodium-dependent bicarbonate transport. Furthermore, the transcripts levels for three genes, co-located in one apparent operon, that encode the ABC transporter family proteins (SynPCC7002\_A1733, SynPCC7002\_A1735) were highly up-regulated (16-34-fold) (Table 4-2).

Thus, it is very likely that the expression of this operon is also regulated by *ccmR* and may also be involved in carbon assimilation or fixation.

In summation, the loss of CcmR results in a remodeling of metabolism with alterations in Ci uptake, bicarbonate transport, and fermentative metabolism, specifically [NiFe]-hydrogenase expression.

### 3.3) In vitro and in vivo hydrogenase activity and NAD(P)H availability

To establish whether the increased expression level of the *hox* operon translates into a higher hydrogenase activity, we measured extracellular H<sub>2</sub> using the standard methyl viologen (MV) assay on cell lysates from both the WT and the  $\Delta ccmR$  strains. After 2 h of dark anaerobic incubation with MV,  $\Delta ccmR$  produced a 50% higher yield of H<sub>2</sub> per mg total protein as compared to the WT strain (Fig. 4-3 A), validating that the increase in Hox gene expression does translate into higher hydrogenase activity.



Figure 4-3 **Hydrogenase activity and mutant verification**. A) In vitro hydrogenase specific activity from WT and  $\triangle ccmR$  cell lysates after 2 h assayed by methyl viologen/dithionite under anaerobic conditions. B) Kinetics of autofermentative H<sub>2</sub> excretion measured over 120 h in sealed vials. Data represents the average and standard deviation of three biological replicates. \*\* represents a t-test *p*-value less than 0.005. C) Agarose gel electrophoretic analysis of PCR products using template DNA derived from wild type (WT) or mutants, verifying the complete segregation of alleles of the targeted gene in each mutant: a)  $\triangle ccmR + gap1^+$  mutant (lane 1  $\triangle ccmR + gap1^+$ ), using primer set ccmRD1-ccmRD4; b)  $\triangle ccmR + gap1^+ + \triangle ldh$  mutant (lane 1  $\triangle ccmR + gap1^+ + \triangle ldh$ ), using primer set ccmRD4.

The hydrogen production rate of whole cells produced under dark fermentative conditions (in vivo activity) was measured either as the concentration of  $H_2$  gas in the headspace by gas chromatography, or as dissolved H<sub>2</sub> in solution using a homebuilt electrochemical cell (Ananyev, 2012). The former method measures the  $H_2$  concentration in equilibrium with cells and thus includes any loss due to the hydrogenase uptake reaction. It was monitored using sealed opaque anaerobic vials over a period of 5 days (Fig. 4-3 B). In these steady-state headspace measurements,  $\Delta ccmR$  mutant produced about 1.5-2-fold more H<sub>2</sub> than the WT. The glycolytic rate of the  $\Delta ccmR$ mutant was not significantly different than the WT (described later in Section 3.5) even though the gap1 gene expression was 16-fold higher (Table 4-2). Earlier work has shown that improving gap1 gene expression by more than 25-fold leads to an improvement in the glycolytic rates and  $H_2$ production (Kumaraswamy et al., 2013). GAPDH1 being a highly regulated enzyme, we hypothesized that the inherent improvement in expression of gap1 in the  $\triangle ccmR$  mutant may not have been sufficient to show a phenotypic difference in the mutant, perhaps owing to insufficient metabolites or altered regulation by changes in the NADH/NAD<sup>+</sup> redox poise (Koksharova et al., 1998), (Garrigues et al., 2001), (Clermont et al., 1993). To test if the glycolytic rate can be enhanced by increasing the GAPDH1 expression level further, we engineered two additional gene changes in the  $\triangle ccmR$  background. They are: a) overexpression of glyceraldehyde 3-phosphate dehydrogenase (GAPDH1) ( $\Delta ccmR + gap1^+$ ), and b) both overexpression of GAPDH1 and deletion of lactate dehydrogenase (LDH) denoted as  $(\Delta ccmR + gap l^+ + \Delta ldh)$ . Mutant construction is described in Experimental Procedures and Fig. 4-1 A. Complete segregation of alleles in the mutants was confirmed by PCR (Fig. 4-3 C). The  $gap l^+$  mutant was previously shown to improve both the glycogen accumulation rate under photoautotrophic growth and the glycogen catabolic rate under dark anaerobisis (Kumaraswamy et al., 2013). The  $\Delta ldh$  knockout was previously shown to improve H<sub>2</sub> production by 5-fold (McNeely et al., 2010). The single mutants,  $gap I^+$  and  $\Delta ldh$ were used as appropriate controls for the experiments. Even though individually,  $\Delta ccmR$ ,  $\Delta ldh$  or  $gap1^+$  did not exhibit any growth defect (Fig. 4-2 A), (McNeely et al., 2010), (Kumaraswamy et al., 2013), the double and the triple mutants showed a 25% and 50% slower growth rate, respectively (Fig. 4-4), although they reached the same final cell density at stationary phase.



Figure 4-4 **Photoautotrophic phenotypes for mutants and Wt**. A) Growth curve of  $\triangle ccmR + gap1^+$  and  $\triangle ccmR + gap1^+ \triangle ldh$  and Wt. Data represents the mean of three biological replicates with standard error grown under 150  $\mu$ E m<sup>-2</sup>s<sup>-1</sup> illumination with constant bubbling of 2% CO<sub>2</sub>. The inset shows the calculated doubling time based on fits to Gompertz curve.

 $H_2$  production by the double and triple mutants was also measured via vial experiments (Fig. 4-3 B). The double mutant,  $(\triangle ccmR + gap1^+)$ , produced about 2.5-3 fold higher H<sub>2</sub> than the WT strain and about 2-fold higher than the single mutants  $\Delta ccmR$  and  $gap1^+$ , respectively. The triple mutant ( $\Delta ccmR + gap1^+ + \Delta ldh$ ) produced about 6-8 fold higher H<sub>2</sub> than WT, 4-6 fold higher H<sub>2</sub> than  $\triangle ccmR$  and  $gap 1^+$ , and about 1-2 fold higher H<sub>2</sub> than the *ldh* strain. Thus, the overall the H<sub>2</sub> production rate was  $(\triangle ccmR + gap1^+ + \triangle ldh) > \triangle ldh > (\triangle ccmR + gap1^+) > \triangle ccmR > gap1^+ >$ WT, an inverse of the trend mirrored by the concentration of intracellular NAD(P)H. The dissolved H<sub>2</sub> production rate was measured continuously by electrochemical consumption under conditions when H<sub>2</sub> is not allowed to accumulate in the headspace. In this configuration the hydrogenase enzyme does not suffer from product inhibition that is otherwise observed in vial measurements. The kinetics from such a technique reflects the organismal rate of  $H_2$  production. For this configuration, a homebuilt fuel-cell coupled to a fluorometer was used (Ananyev et al., 2012). The fuel cell consumes the produced H<sub>2</sub> (termed "milking"), shifting the equilibrium of the reaction  $2H^+ + 2e^- \leftrightarrow H_2$  towards right, while the fluorometer measures the instantaneous changes in the intracellular NAD(P)H concentration in real-time parallel to the H<sub>2</sub> production. For the milking experiments, the measurements were limited to the 3 mutant strains in  $\Delta ccmR$  background.



Figure 4-5 **Simultaneous detection of H<sub>2</sub> and NADH.** A) Excreted (extracellular) hydrogen oxidation current during 20 hrs of continuous dark anaerobic autofermentation is directly proportional to the rate of H<sub>2</sub> production ("Milking" conditions). Inset shows the yield of H<sub>2</sub> (nmol) at 20 h for the four strains (from integration of the curves and using Faraday's laws of electrolysis). (b) Simultaneous determination of the relative NAD(P)H fluorescence yield is directly proportional to the total concentration of intracellular reduced pyridine nucleotides under the same conditions as (A). The real time Data in (A) and (B) is one representative sample.

Within the first 20 h, all the strains produced about 50-fold higher H<sub>2</sub> (Fig. 4-5 A) than under the fixed headspace vial experiments (Fig. 4-3 B). Comparing across strains, the hydrogen production in the  $\triangle ccmR$  strain was about 2.3-fold higher than the WT, while the double and the triple mutants were 3.7 and 7.3 fold higher than the WT (Fig. 4-5 A). Simultaneous NAD(P)H fluorescence indicated that the highest intracellular levels were maintained in the WT cells with the lowest hydrogen production rate (Fig. 4-5 B), while in the  $\triangle ccmR$  mutant, the residual NADPH is lower, suggesting an improved consumption of reductant by hydrogenase. The double and the triple mutants show significantly lower amounts of residual reductant levels and corresponding higher H<sub>2</sub> production rate. Thus, taken together, there is a direct correspondence between higher H<sub>2</sub> production rate and lower intracellular NAD(P)H concentration is the four strains. Accordingly, the deletion of the *ccmR* gene, which upregulates the expression of hydrogenase genes, is an apparent response to elevated NAD(P)H availability and serves the purpose of a vent for elimination of excess reductant capacity as volatile H<sub>2</sub> gas.

#### 3.4) Autofermentation products:

Because *nifJ* (PFOR) showed a 5.7-fold increase in expression (Table 4-2), additional fermentative pathways were also studied by quantifying excreted fermentative metabolites (acetate, lactate and pyruvate) using cell-free supernatants (Fig. 4-6 A-C) collected during the vial autofermentation experiments.



Figure 4-6 **The kinetics of autofermentative metabolite excretion measured over 120 h in sealed vials.** A) lactate, B) acetate and C) pyruvate present in the spent medium determined using HPLC. Data represents the mean and standard deviation of three biological replicates. B.D. = below detection limits

The  $\triangle ccmR$  mutant excreted about ~0.7-fold lower lactate as compared to the WT, while the double mutant  $(\triangle ccmR + gap1^+)$  excreted about 1.5-fold higher lactate, as compared to both the WT and the gap  $I^+$  mutant strain. Finally, as expected, the  $\Delta ldh$  and the triple mutant ( $\Delta ccmR +$  $gap1^+ + \Delta ldh$ ) did not show detectable levels of lactate in the extracellular medium (Fig. 4-6 A). All strains except for the triple mutant ( $\Delta ccmR + gap l^+ + \Delta ldh$ ), generated the same amount of acetate (~ 100 nmoles per 10<sup>8</sup> cells), as compared to the WT strain (Fig. 4-6 B), while the triple mutant produced 4-fold higher acetate levels compared to the WT. While all the strains had very low levels of excreted pyruvate (Fig. 4-6 C), the triple mutant  $(\Delta ccmR + gap1^+ + \Delta ldh)$ , lacking the ability to excrete carbon as lactate, excreted about 3-7 fold higher levels of pyruvate as compared to the WT, suggesting that the loss of *ldh* and the overexpression of *gap1*, the cells are forced to excrete carbon through alternate pathways. The lactate/acetate ratio was found to be lower in the  $\Delta ccmR$  (1.5-2.5) than WT (2.5-3.5), suggesting a selective redirection of the glycolytically generated pyruvate into acetate rather than lactate in the mutant. Interestingly, the introduction of gap1 overexpression selectively redirects the fermented carbon into lactate instead of acetate, and the same occurs when gap1 is introduced in the  $\triangle ccmR$  background. However, on limiting the flux through lactate by deleting the lactate dehydrogenase gene in the triple mutant, the cells continue to break down glycogen, but now excrete significantly more pyruvate and acetate into the medium. 3.5) Glycolytic rate:

To investigate if the higher H<sub>2</sub> production in  $\Delta ccmR$  was caused by higher hydrogenase activity or as a by-product of higher glycolytic rate leading to greater reductant production, we quantified the total intracellular reducing carbohydrate (TRC) as glucose equivalents using the anthrone assay (Fig.4-7). The cell pellets stored during the course of autofermentation were used for this purpose. Amongst the mutants, the triple mutant ( $\Delta ccmR + gap1^+ + \Delta ldh$ ) and the double mutant ( $\Delta ccmR + gap1^+$ ) accumulated the highest amounts of total reducing carbohydrate (TRC) equivalents during photoautotrophic growth (~1.2 µmoles glucose/10<sup>8</sup> cells) while the  $\Delta ccmR$  and the WT strains had about ~0.8 µmoles glucose/10<sup>8</sup> cells at the start of dark anaerobisis (autofermentation). As shown previously,  $gap1^+$  had a higher amount of TRC (0.95 µmoles glucose/10<sup>8</sup> cells) than WT. Linear fits to the TRC data gave the glycolytic rate (Fig. 4-7 B). In comparison, the  $\Delta ccmR$  strain had a marginally lower glycolytic rate (~15%) than the WT, although not statistically significant (p=0.3291). The introduction of gap1 overexpression increased the glycolytic rate in the  $\Delta ccmR$  strain by about 60% more than the WT, consistent with previous results on single gap1 ovexpression (Kumaraswamy et al., 2013). The  $\Delta ldh$  strain had a 50% lower glycolytic rate, while as expected the  $gap1^+$  had a 40% higher glycolytic rate. In summary, the results indicate that the loss of CcmR produces nearly no change in rate of anaerobic glycogen breakdown, in contrast to aerobic respiration, for which this rate is atleast 50% higher than the WT (Fig. 4-2 C, from respiration measurements). Additionally, these results also indicate that the gain in hydrogen in the  $\Delta ccmR$  mutant is indeed an effect of the increased hydrogenase activity, while that in the double and the triple mutant is a combined effect of higher glycolysis and higher hydrogenase activity.



Figure 4-7 **Autofermentative catabolism of glycogen**.A) Autofermentative catabolism of glycogen. The total intracellular reducing carbohydrate content was measured via anthrone assay, following the dark anaerobic incubation at indicated time points. Data represents the mean and standard deviation of three biological replicates. B) Glycolytic rate. Data represents the fitted slope and standard error of three biological replicates calculated using GraphPad Prism6 statistical software.

#### *3.6) Cellular energy charge:*

To determine if the altered acetate production is accompanied by an increase in the cellular energy charge (CEC) or in the redox poise, the steady-state pool sizes of the adenylate energy carriers were measured by LC-MS/MS after 3 days of fermentation (Table 4-3). CEC was calculated as CEC = (ATP + (0.5ADP))/(ATP + ADP + AMP), and redox poise as (NAD(P)H)/(NAD(P)). The *ccmR* mutant had neither a higher CEC nor a higher redox poise, even though lactate production was suppressed as compared to the WT. Both the double and the triple mutant had significantly higher CEC (Table 4-3) with unchanged redox poise.

	Wt $\triangle ccmR$ $gap1^+$ $ldh^-$		$\Delta ccmR+$	$\Delta ccmR+$		
					$gap1^+$	gap1+ldh-
AMP	2.7 (0.50)	3.4 (0.21)	3.2 (0.33)	3.3 (0.24)	3.2 (0.50)	1.9 (0.14)
ADP	0.39 (0.11)	0.53 (0.03)	0.99 (0.34)	0.58 (0.06)	1.8 (0.07)	1.0 (0.07)
ATP	0.09 (0.02)	0.08 (0.01)	0.38 (0.10)	0.16 (0.01)	0.98 (0.05)	0.82 (0.10)
CEC	0.11 (0.01)	0.09 (0.01)	0.17 (0.01)	0.11 (0.01)	0.32 (0.06)	0.035 (0.07)
NAD+	0.07 (0.01)	0.08 (0.01)	0.11 (0.02)	0.06 (0.01)	0.15 (0.01)	0.03 (0.00)
NADH	0.12 (0.01)	0.14 (0.05)	0.16 (0.03)	0.08 (0.02)	0.2 (0.03)	0.06 (0.01)
NADP+	0.01 (0.00)	0.03 (0.00)	0.04 (0.01)	0.03 (0.00)	0.12 (0.01)	0.03 (0.00)
NADPH	0.0 (0.00)	0.02 (0.00)	0.03 (0.01)	0.02 (0.00)	0.08 (0.01)	0.01 (0.01)
Redox	1.5	1.6	1.3	1.2	1	1.2
poise						

Table 4-3 LC-MS/MS results for adenylate and pyridine nucleotides. Values represent the concentration of the metabolites after 3 days of autofermentation calculated as  $\mu$ moles per 10<sup>8</sup> cells. Experiment was performed on 4 biological replicates. Standard error of mean are represented in parentheses. CEC and redox poise calculation described in the Results section.

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#### 4. Discussion

# 4.1) CcmR deletion remodels respiratory metabolism to enhance ATP production, but does not improve photoautotrophic growth under Ci sufficiency

CcmR is a LysR-type transcription factor, which under Ci sufficient conditions, regulates the expression of the genes involved in carbon uptake (Woodger et al., 2007), (Price et al., 2008), (Daley et al., 2012), (Burnap et al., 2015). Loss of CcmR mimics a low Ci environment and is sufficient to derepress the genes involved in the high affinity Ci transport (Lieman-Hurwitz et al., 2009). The results presented here show that an increase in the Ci uptake genes (Table 4-2) does not guarantee an increase in overall growth under Ci sufficient conditions (Fig. 4-2 A). This is not unexpected as the CCM pathway has been previously suggested to decrease growth in air, not because of the lack of carbon but because of the high energetic demand for maintaining the high affinity CCM (Burnap et al., 2015). Depending on the mechanism, CCM-dependent Ci uptake may use 1 ATP, 1 proton or at least one NADPH per bicarbonate imported (Burnap et al., 2015), (Raven et al., 2014). To fulfill this higher requirement for ATP/ $H^+$ , the oxidative phosphorylation pathway is preferred, a result supported by both the 50% higher respiration rate (Fig. 4-2 C), as well as enhanced expression of the subunits of respiratory Complex IV and Complex II (Table 4-2). Thus, the *ccmR* deletion strain appears to remodel its global photoautotrophic metabolism to enhance respiration and thus ATP production. The schematic representation of the overall changes occurring by the loss of *ccmR* is presented in Scheme 4-1.



Scheme 4-1 Schematic representation of key features of physiology and metabolism on the loss of ccmR in *Synechococcus* 7002. Small and large font sizes are used for the various processes, metabolites and the gene expression to indicate the quantitative differences relative to the WT, with the fold size of 'Reducing carbohydrate' being 1. The solid up and down arrows qualitatively represent either the increase or the decrease in the process, metabolites or the gene expression relative to the WT.

## 4.2) Deletion of CcmR alters fermentative metabolism, enhances $H_2$ production and improves the flux of carbon towards acetate

Anaerobic metabolism in  $\Delta ccmR$  also occurs in a way as to maximally conserve the adenylate energy. Glycolysis, the dominant ATP-producing pathway under fermentative conditions, generates pyruvate as its end product. The generated pyruvate may have at least 2 dominant fates in Synechococcus 7002 (Fig. 1-2): a) reduction of pyruvate to lactate via lactate dehydrogenase with a concomitant oxidation of NADH, or b) conversion of pyruvate to acetyl-CoA using either pyruvate dehydrogenase (reduces NAD<sup>+</sup> to NADH) or via nifJ (reduces ferredoxin) (McNeely et al., 2011). The acetyl-CoA can be converted to acetate producing ATP via substrate-level phosphorylation. At the metabolic level, the *ccmR* mutant strain still excretes lactate as the dominant fermentative end product than acetate (Fig. 4-6 A, B). However, comparing the fractions of excreted carbon as lactate and acetate between the mutant and the WT, a higher fraction of acetate is excreted in the  $\triangle ccmR$  strain. Consequently,  $\triangle ccmR$  redirects the catabolized carbohydrates into the acetogenic route (acetate) that generates ATP. In addition, transcripts for an alternate modified Embden-Meyerhof pathway using phosphoenolpyruvate synthase (*ppsA*), that converts phosphoenolpyruvate to pyruvate, is enhanced (3.6-fold) (Table 4-2). ppsA performs the same function as pyruvate kinase, but uses AMP as the substrate for ATP synthesis rather than ADP (Fig. 1-2) and is able to conserve a greater amount of energy as compared to the pyruvate kinase pathway. In addition, the rerouting of carbon flux through the *nifJ* pathway reduces ferredoxin which has a higher energy content than NADH produced by pyruvate dehydrogenase. This suggests that the normal glycolytic pathway may be replaced by the alternative pathway comprising *ppsA* and *nifJ* to conserve and maximize ATP and NADH, a result seen previously in Synechococcus 7002 under anoxic conditions (Ludwig and Bryant, 2011a). While a redirection of carbon to acetate occurs, the carbon partitioned into the lactate sink is diminished, which may be expected to cause an over-reduction of the electron carriers.  $\Delta ccmR$  maintains a steady redox poise

in vial experiments, while upon milking, the NAD(P)H levels are lowered as compared to the WT with improvements in hydrogenase activity. Therefore, we propose that to compensate for loss of carbon through lactate excretion, the cells enhance the carbon-neutral hydrogenase as a reductant vent in order to maintain redox homeostasis.

#### 4.3) Insertion of gap $1^+$ overexpression removes the glycolytic chokepoint and improves $H_2$ .

The loss of the *ccmR* expression resulted in an increase in the expression of hydrogenase genes as well as the hydrogenase activity – the *in vivo* hydrogenase activity as measured by headspace measurements increased by a modest 1.5- to 2-fold (Table 4-2, Fig. 4-3 B) and that under milking increased 2.3-fold (Fig. 4-5 A). Enhancing the glycolytic rate (Kumaraswamy et al., 2013) and removing alternate competing reductant sinks (McNeely et al., 2010) are proven methods to improve  $H_2$  production further. The glycolytic bottleneck was alleviated by introducing gap1 overexpression (Kumaraswamy et al., 2013) in the  $\Delta ccmR$  mutant strain, resulting in a 4-fold higher headspace  $H_2$  production. This yield was further enhanced by the deletion of lactate dehydrogenase, resulting in 8-fold higher H<sub>2</sub> production. In both mutants containing gap1 overexpression, the glycolytic rate and the glycogen accumulation improved (Fig. 4-7). This increase in glycolysis shifted the fermentative metabolism back towards lactate synthesis, while additional blocking of lactate biosynthesis in the  $\Delta ldh$  strain causes excretion of pyruvate (Fig. 4-6 C). Interestingly, the  $\Delta ccmR$  strain itself has a 16-fold higher gap1 gene expression (Table 4-2) with no improvement in the glycolytic rate (Fig. 4-7). However, when the gap1 expression was artificially improved by introducing the overexpression construct, together with an increase in the total accumulated glycogen, it also resulted in an increase in the glycolytic rates and  $H_2$  production. Taken together, loss of CcmR appears to regulate glycolysis tightly under fermentative conditions using hydrogenase as a vent for excess reducing equivalents. The introduction of  $gap1^+$  appears to override the regulatory controls of glycolysis and gluconeogenesis, indicating that much higher GAPDH levels overcome the bottleneck in glycolysis.

#### *4.4)* Theoretical energy conversion efficiency (ECE):

The energy conversion efficiency (ECE) of fermentative  $H_2$  production is the ratio of the combustion enthalpies of  $H_2$  produced to that of the glucose consumed. Table 4-4 gives the glucose to H<sub>2</sub> molar conversion ratios and calculated ECE values for the WT,  $\Delta ccmR$ ,  $(\Delta ccmR + gap 1^+)$  and  $(\Delta ccmR + gap1^+ + \Delta ldh)$  mutant under headspace measurements and under milking conditions. As exponential phase cultures were used in the study instead, the molar conversion ratios were significantly lower than the previously reported values for stationary cultures (McNeely et al., 2010). Autofermentation of the exponentially grown cells gave an ECE value of 0.34% for the WT strain (headspace measurement), which was improved to 1.8% upon milking. The  $\Delta ccmR$  mutant was found to have a higher ECE value than the WT under both non-milking (0.64%) and under milking (3.8% conditions) even though the glycolytic rate was marginally lower in the  $\Delta ccmR$ strain. Thus, the deletion of *ccmR* clearly results in improving the availability of reductant to hydrogenase. On the other hand, the double and the triple mutant had ECE values similar to the WT under non-milking conditions (0.53% and 1.3% respectively); however, upon milking, the ECE values increased to higher values of 4.2% and 8.9% respectively. These results are in complete agreement with the literature that the headspace accumulation does indeed limit  $H_2$  production (McNeely et al., 2011), (Ananyev et al., 2012). The improvement in the ECE value between  $\triangle ccmR$ and WT clearly show that under non-milking conditions, introduction of  $\Delta ccmR$  deletion improves the availability of reductant to  $H_2$ , while introduction gap 1 overexpression increases the amount of reductant generated via glycolysis.

	WT	∆ccmR	$gap1^+$	ldh <sup>-</sup>	$\Delta ccmR+$	$\Delta ccmR+$
					$gap1^+$	gap1 <sup>+</sup> ldh-
Molar conversion (H2 mols/glucose mols) (headspace 120 h)	0.03	0.06	0.03	0.23	0.05	0.12
ECE (headspace 120 h)	0.34	0.64	0.34	2.4	0.53	1.3
Molar conversion (H2 mols/glucose mols)(milking 20h)	0.18	0.39	N.D	N.D	0.42	0.87
ECE (milking 20h)	1.8	3.9	N.D	N.D	4.2	8.9

Table 4-4. Calculated energy conversion efficiency for  $H_2$ . Calculated ECE values based on the  $H_2$  produced to glycogen consumed under headspace and under milking conditions.

In conclusion, our results show that loss of ccmR gene expression remodels the fermentative carbon metabolism and uses hydrogenase as a terminal oxidase to rebalance excess reducing equivalents as volatile H<sub>2</sub>. The CcmR deletion strain may serve as a good platform to improve H<sub>2</sub> production through further metabolic engineering.
## Chapter 5: Conclusions

Understanding how microalgae partition photosynthetically fixed carbon into the terminal carbon sinks is fundamental for improving biofuel production. The objective of this thesis was to gain insights into the molecular basis of carbon partitioning in a model microalga *Chlamydomonas reinhardtii*. I have employed a series of starchless mutants in conjugation with nutrient starvation as a model system for modulating the normal carbon distribution in this microalga. In chapters 2 and 3, several experiments have been presented leading to a greater understanding of carbon partitioning and its coupling to photosynthesis. Chapter 4 is an example where prior knowledge of central carbon metabolism has been used to effectively enhance hydrogen production.

In chapter 2, we investigated the role of starch biosynthesis in affecting photosynthetic growth and carbon partitioning in *sta6*, a starchless mutant of *Chlamydomonas reinhardtii* lacking ADP-glucose pyrophosphorylase (AGPase). This mutant is unable to convert glucose-1-P to ADP-glucose, the precursor to starch biosynthesis. At the start of the project, we expected to observe that loss of starch biosynthesis would alter carbon partitioning among the other carbon products under photoautotrophic growth conditions. However, during the course of experimentation, we found that the mutant did not change its fraction of lipids and proteins but did produce 20% lower biomass. Probing the underlying mechanism revealed an intact photosynthetic electron transport chain in the mutant, but possessing a greatly attenuated rate of NADPH reoxidation and a 3 fold higher ATP pool size. This phenotype manifests only above a particular light threshold, indicating the importance of starch biosynthesis when excess energy is available and that, starch biosynthesis in *Chlamydomonas* plays a critical role in light energy storage and rebalancing energy.

As *sta6* did not exhibit an altered carbon partitioning under nutrient replete growing conditions, in chapter 3, we imposed nutrient starvation as a tool to shift microbial cellular metabolism to favor selected bioproducts. Prior work has shown that under nitrogen (N) starvation, *Chlamydomonas reinhardtii* redirects biosynthesis from proteins to starch and lipids, and in

mutants lacking starch biosynthesis genes, preferentially to lipids. Our results revealed three important metabolite nodes, 'the carbohydrate node', 'the TCA node' and 'the fatty acid node' that serve as indicators of possible flux changes during N starvation. Both transient <sup>13</sup>C-metabolite fluxes produced from fixed  $H^{13}CO_3^{-}$  and steady-state metabolite pool sizes reveal reduced flux through the TCA cycle and increased flux through gluconeogenesis as system wide responses to N starvation in strains with or without starch biosynthesis. Interestingly, our results showed that even in the absence of starch biosynthesis, gluconeogenesis is the inherent response to N starvation. This is achieved by redirecting carbon into the oxidative pentose phosphate (OPP) pathway via G6P. Even though the use of OPP leads to a significant loss in  $CO_2$ , this pathway generates the additional reductant needed for lipid biosynthesis. Thus, the loss of starch biosynthesis forces a remodeling of metabolism that is necessary to increase the reductant available for lipid biosynthesis. Simultaneous measurements of adenylate nucleotides (ATP, ADP, AMP) revealed a tight correlation between elevated cellular energy charge (CEC) and the flux into gluconeogenesis. The three nodes and the CEC dependent changes of flux are important mutagenesis targets that could be explored further. From these studies of two starchless mutants, I conclude that together with the inherent (transcriptional/translational) response to nutrient stress, the energy requirements for the synthesis of the terminal carbon sink plays an important role in governing the metabolic fluxes. Thus, even in the absence of starch biosynthesis, gluconeogenesis is maintained and coupled to the oxidative pentose phosphate pathway to generate the excess reductant necessary for lipid biosynthesis. Therefore, in the future, modulating the ATP:NADPH ratio so as to improve the NADPH availability need to be devised to enhance the fluxes towards lipids.

In chapter 4, we enhanced biohydrogen production in a euryhaline cyanobacterium, *Synechococcus* sp. PCC 7002. We initiated our study with a carbon concentration mechanism mutant, CcmR. CcmR is a LysR-type transcriptional regulator that is known to represses the expression of genes encoding high-affinity components of the carbon concentration mechanism

(CCM) in cyanobacteria. Even though, CcmR was presumed to be unrelated to fermentative metabolism, prior results from our collaborator demonstrated that the inactivation of CcmR alters the expression of multiple fermentative genes, especially hydrogenase expression in *Synechococcus 7002*. Therefore, to test if the CCM affects dark anaerobic metabolism, we measured the kinetics of autofermentation in the wild type (WT) and the ccmR deletion strains of *Synechococcus* sp. PCC 7002, measuring both excreted metabolites by mass spectrometry and the intracellular concentration of NAD(P)H continuously by fluorescence. Upon fermentation, loss of CcmR resulted in a 1.5-fold increase in the *in vivo* hydrogenase activity, while continuous removal of produced H<sub>2</sub> (by milking) boosted H<sub>2</sub> production to 2.6-fold compared to the Wt. Further, by sequentially increasing glycolytic rate and removing alternate competing pathways, we were able to metabolically engineer the CcmR strain and further boost H<sub>2</sub> production levels by 6-8-fold above that from the wild type strain.

Taken together, the results from my thesis, furthers our understanding of carbon partitioning while simultaneously providing proof of concept of how targeted metabolic engineering can help in improving productivity. The carbohydrate node (DHAP-FBP junction) described in chapter 3 is definitely worth exploring as metabolic engineering target to modulate carbon partitioning between terminal sinks. The elucidation of the regulatory mechanisms would help make algal biofuels both commercially viable as well as a sustainable alternative fuel. References:

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