

Enzymatic digestibility and pretreatment degradation products of AFEX-treated hardwoods (*Populus nigra*)

Rutgers University has made this article freely available. Please share how this access benefits you.
Your story matters. <https://rucore.libraries.rutgers.edu/rutgers-lib/50923/story/>

This work is an **ACCEPTED MANUSCRIPT (AM)**

This is the author's manuscript for a work that has been accepted for publication. Changes resulting from the publishing process, such as copyediting, final layout, and pagination, may not be reflected in this document. The publisher takes permanent responsibility for the work. Content and layout follow publisher's submission requirements.

Citation for this version and the definitive version are shown below.

Citation to Publisher Balan, Venkatesh, da Costa Sousa, Leonardo, Chundawat, Shishir P. S., Marshall, Derek, Sharma, Lekh N., Chambliss, Charles Kevin & Dale, Bruce E. (2009). Enzymatic digestibility and pretreatment degradation products of AFEX-treated hardwoods (*Populus nigra*). *Biotechnology Progress* 25(2), 365–375. <http://dx.doi.org/10.1002/btpr.160>.

Citation to this Version: Balan, Venkatesh, da Costa Sousa, Leonardo, Chundawat, Shishir P. S., Marshall, Derek, Sharma, Lekh N., Chambliss, Charles Kevin & Dale, Bruce E. (2009). Enzymatic digestibility and pretreatment degradation products of AFEX-treated hardwoods (*Populus nigra*). *Biotechnology Progress* 25(2), 365–375. Retrieved from [doi:10.7282/T3ZP48DW](https://doi.org/10.7282/T3ZP48DW).

Terms of Use: Copyright for scholarly resources published in RUcore is retained by the copyright holder. By virtue of its appearance in this open access medium, you are free to use this resource, with proper attribution, in educational and other non-commercial settings. Other uses, such as reproduction or republication, may require the permission of the copyright holder.

Article begins on next page

1
2
3
4 **Enzymatic digestibility and pretreatment degradation**
5
6
7
8 **products for AFEX treated hardwoods (*Populus nigra*)**
9

10
11
12
13 Venkatesh Balan^{1,*}, Leonardo da Costa Sousa¹, Shishir P. S. Chundawat¹,
14
15
16 Derek Marshall¹, Lekh N. Sharma², C. Kevin Chambliss² and Bruce E. Dale¹
17
18

19
20
21 ¹ *Biomass Conversion Research Laboratory, Department of Chemical Engineering and*
22
23 *Materials Science, Michigan State University, E. Lansing, MI – 48824*
24

25
26 ² *Department of Chemistry and Biochemistry, Baylor University, Waco, TX 76798*
27
28
29
30
31
32

33 **Abstract:**
34

35 There is growing need to find alternatives to crude oil as the primary feed stock for the
36
37 chemicals and fuel industry and ethanol has been demonstrated to be a viable alternative.
38
39 Among the various feed stocks for producing ethanol, poplar (*Populus nigra X populus*
40
41 *maximowiczii*) is considered to have a great potential for biorefineries in the US, due to
42
43 their widespread availability and good productivity in several parts of the country. We
44
45 have optimized AFEX pretreatment conditions (180 °C, 2:1 ammonia to biomass loading,
46
47 233% moisture, 30 min. residence time) and by adding different combinations of
48
49 enzymes (commercial cellulases and xylanases) in order to achieve high glucan and xylan
50
51 conversion (93 and 65%, respectively). We have also identified and quantified several
52
53 important degradation products formed after AFEX using liquid chromatography
54
55
56
57
58
59
60

1
2
3 followed by mass spectrometry (LC-MS/MS). As a part of degradation product analysis
4
5 we have also quantified oligosaccharides in the AFEX water wash extracts by acid
6
7 hydrolysis. It is interesting to note that corn stover (C4 grass) can be pretreated
8
9 effectively using mild AFEX pretreatment conditions, while on the other hand hardwood
10
11 poplar needs much harsher AFEX conditions to obtain equivalent sugar yields upon
12
13 enzymatic hydrolysis. Based on the oligosaccharide analysis comparison between AFEX
14
15 treated stover and poplar, we conclude that pretreatment severity and enzymatic
16
17 hydrolysis efficiency are dictated to a large extent by lignin carbohydrate complexes and
18
19 arabinoxylan cross-linkages for the AFEX process.
20
21
22
23
24
25
26
27
28

29 *Key words:* Corn stover, Poplar, AFEX pretreatment, enzymatic hydrolysis,
30
31 Lignocellulose
32
33
34
35

36 * To whom correspondence should be addressed. Tel: 517-336-4615, Fax: 517-337-
37
38 7904. E-mail: balan@msu.edu
39
40
41
42
43

44 **Introduction:**

45
46 The growing U.S. appetite for petroleum, fueled together with growing demand in
47
48 China, India, and rest of the world, has pushed crude oil prices to a new high. The United
49
50 States consumes over 20 million barrels of petroleum per day, of which over 60% is
51
52 imported. Crude oil prices have risen as high as \$147 per barrel (bbl) in July 2008, a
53
54 remarkable 400% increase in cost over the last decade (www.wtrg.com). Hence, there is a
55
56
57
58
59
60

1
2
3 growing urgency to find suitable alternatives to petroleum-derived fuels. Bioethanol is
4 one such suitable prospect that can provide a potentially low cost, environmentally-
5 friendly way to reduce gasoline consumption while helping reduce net carbon dioxide
6 emissions (1). Thus, a considerable amount of research is currently underway to
7 economically produce ethanol from lignocellulosics, which are far more abundant in
8 nature and cheaper to produce compared to conventional feedstocks (e.g. sugarcane, corn
9 starch) (2, 3).

20
21
22 Cellulose, one of the major components of the plant cell wall, is a linear condensation
23 polymer consisting of D-anhydroglucopyranose joined together by β -1,4-linkage with a
24 degree of polymerization ranging from 100 to 20,000. Adjacent cellulose molecules are
25 coupled by extensive hydrogen bonds and Van der Waals forces resulting in a parallel
26 crystalline alignment, and producing a rigid and stable supramolecular structure with low
27 accessibility to chemicals and enzymes (4). In addition, cellulose is embedded in a
28 hemicellulose and lignin matrix which makes it even more recalcitrant during enzymatic
29 hydrolysis (5-7). Disruption of these bonds by thermochemical pretreatment, using acid
30 or base, increases cell wall porosity and drastically enhances the accessibility of enzymes
31 to the sugar polymers (8).

32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49 A novel alkaline pretreatment method to improve lignocellulose digestibility is the
50 ammonia fiber expansion (AFEX) process (9). Ammonia is added to the biomass under
51 high pressure (200-700 psi) with varying temperatures (60-200 °C) before rapidly
52 releasing the pressure. The AFEX process appears to be economically attractive and on-
53
54
55
56
57
58
59
60

1
2
3 going research has allowed further cost improvements (10). AFEX is thought to
4
5 decrystallize cellulose, while partially hydrolyzing hemicellulose through hydrolysis of
6
7 lignin-hemicellulose ester cross-linkages (11).
8
9

10
11
12 Effectiveness of pretreatment and enzymatic hydrolysis is dictated to a large extent by
13
14 biomass composition, among other factors. During enzymatic hydrolysis, enzyme tends
15
16 to irreversibly bind to lignin, an aromatic polymer, through hydrophobic interactions that
17
18 cause loss in activity. Hence, the amount and composition of lignin critically affect the
19
20 digestion of sugar polymers to soluble sugars. Lignin content in grasses (15-20%) is
21
22 relatively low compared to hard and softwood (20-35%) (12). This could be one of the
23
24 major reasons that grasses, can be more easily digested via hydrolytic enzymes compared
25
26 to hardwood following AFEX pretreatment (13). In addition to lignin, the arabinoxylan
27
28 content of the biomass is also thought to play a crucial role to the effectiveness of the
29
30 pretreatment process. In monocots of the arabinoxylan is connected to lignin via ester
31
32 and ether linkages (5, 14-17). These linkages are between arabinose side chains of
33
34 hemicelluloses and hydroxyl/carboxyl functionalities of lignin (e.g. ferulic acid).
35
36 Ammonia has the tendency to cleave these ester linkages (18-20) in biomass via
37
38 ammonolysis. It is interesting to note that the arabinose (found mostly as arabinoxylan)
39
40 content of grasses (3-6%) is significantly higher compared to hardwood (<1%). Thus the
41
42 higher the arabinoxylan content and possible ester linkages to arabinoxylan, the greater
43
44 its susceptibility to cleavage during the AFEX process (21). Cleavage of these linkages
45
46 promotes the disruption of the cell wall complex structure in such a way that the enzymes
47
48 can access the cellulose and hemicellulose more efficiently. Fundamental understanding
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 of how these linkages are chemically modified during the pretreatment process may help
4
5 eventually design a genetically modified cell wall which can be more easily pretreated
6
7 under mild conditions using fewer chemicals (17).
8
9

10
11
12 Water washing of AFEX treated corn stover was shown to remove some of the AFEX-
13 mediated surface deposits resulting in a 13-15% weight loss (w/w) (22). The cleavage of
14
15 the acetyl groups from hemicellulose results in the formation of acetamide and acetic
16
17 acid. Lactic acid and formic acid are also formed during AFEX, likely due to alkali
18
19 induced degradation of reducing sugars and lignin (23, 24). Moreover, phenolic
20
21 compounds like p-coumaric and ferulic acid are also produced due to the hydrolysis of
22
23 lignin-hemicellulose cross-linkages (25). Phenolic aldehydes were more likely to be
24
25 produced in oxidative alkaline conditions (24). Some components such as 4-
26
27 hydroxybenzaldehyde and 4-hydroxybenzoic acid are thought to be monomeric lignin
28
29 extractives that are easily released after the pretreatment process, unlike the other lignin
30
31 cleavage products (26, 27). Identification and quantification of these degradation
32
33 products using liquid chromatography-tandem mass spectrometry (LC-MS/MS) will
34
35 provide valuable information that will help understand the kinetics and mechanisms by
36
37 which these compounds are formed and enable design of optimized AFEX pretreatment
38
39 conditions.
40
41
42
43
44
45
46
47
48
49

50
51 Biologically inhibitory effects of compounds from wood hydrolysates depend on the
52
53 chemical structure and reactivity: e.g. terpenes > aldehydes > polyhydroxy aromatics, and
54
55 formic acid > acetic acid (28) for yeast fermentations. Low molecular weight (MW)
56
57
58
59
60

1
2
3 compounds and salts are able to penetrate cell membranes, and are thus more toxic to
4 fermentative microbes compared to high molecular weight compounds (29). Lower MW
5
6 compounds influence the expression and activity of sugar and ion transporters in the cell
7
8 membrane. Mechanisms for inhibition of microbial growth and ethanol production due to
9
10 weak acids, furans and phenols have been reviewed recently (30).
11
12
13
14
15
16

17
18 Among the various sources of biomass, agricultural residues like corn stover and
19
20 hardwoods like hybrid poplar (*Populus nigra x Populus maximowiczii*) are of interest.
21
22 Woody biomass has several advantages compared to agricultural residues including
23
24 ability to be stored “on the stump”, higher mass density that reduces transport cost and
25
26 increased sugar content (i.e. glucan (~40-50%) and xylan (~20-30%) content.
27
28 Considerable information is available on steam explosion (31), organosolv pretreatment
29
30 (32), acid pretreatment (33, 34) followed by enzymatic hydrolysis of poplar, but very
31
32 little work has been reported on alkaline pretreatment (in particular AFEX) (35) of
33
34 hardwoods. Corn stover can be pretreated effectively using mild AFEX pretreatment
35
36 conditions, while on the other hand poplar needs much harsher AFEX conditions to
37
38 obtain equivalent sugar yields upon enzymatic hydrolysis. In this paper we discuss how
39
40 AFEX pretreatment severity and enzymatic hydrolysis efficiency are dictated by the plant
41
42 cell wall ultra structure and composition of various components, such as lignin
43
44 carbohydrate complexes (LCC) and arabinoxylan cross-linkages. Details of the AFEX
45
46 pretreatment conditions, degradation products formed during pretreatment, sugar
47
48 conversions for varying enzyme loadings and the mass balance for poplar and corn stover
49
50 are also discussed in this article.
51
52
53
54
55
56
57
58
59
60

Materials and Methods:

Lignocellulosic substrate. Hardwood poplar was provided by National Renewable Energy Laboratory (CO, Denver), and was milled using a 50 mm sieve. The moisture content was measured using a moisture analyzer (Model MF-50, A&D). The samples had approximately 50% (total weight basis) moisture and were stored at -20 °C freezer until further experiments were carried out.

Compositional Analysis. Compositional analyses of the samples were performed according to NREL Laboratory Analytical Procedures (LAPs): “Preparation of samples for compositional analysis” and “Determination of structural carbohydrates and lignin in biomass” (36). Monomeric sugars were quantified using a Bio-Rad Aminex HPX-87H high performance liquid chromatography (HPLC) column.

AFEX Pretreatment. A bench-top reactor consisted of a 22 ml # 316 stainless steel pressure vessel (PARR Instrument Co, IL). The vessel was loaded with feedstock containing the appropriate moisture content. The vessel was clamped shut and the required amount of ammonia was injected using a pre-weighed sample cylinder. The reactor was heated by placing it inside a slotted aluminium block attached to a Vela hot plate (Cole Parmer, Inc.). The slots in the aluminium heating block were precision milled to present a tight fit around the pressure vessel for even heating and good heat transfer. The reactor was maintained at the desired temperature during the course of the pretreatment. The residence time in the reactor depended on the feedstock treated (e.g. 5

1
2
3 minutes for corn stover and 30 minutes for poplar). It took approximately 30-60 minutes
4
5 to complete one AFEX reaction. At the end of the residence time, the pressure was
6
7 explosively released by abruptly opening the half a inch NPT #316 stainless steel ball
8
9 valve installed on the reactor. The biomass was promptly removed from the reactor and
10
11 left in the hood overnight to allow the residual ammonia to evaporate..
12
13
14

15
16
17 **Washing.** Some samples were washed after AFEX treatment to remove soluble lignin
18
19 and other compounds prior to enzymatic hydrolysis. AFEX treated biomass was washed
20
21 using distilled (de-ionized) water with a substrate to water loading of 1:10 (w/w). The
22
23 slurry was mixed for 30 minutes and the wash liquid was removed from the substrate by
24
25 squeezing the slurry through a filtration cloth (Miracloth, Calbiochem, CA) with typical
26
27 pore size of 22-25 μm . The filtrate was centrifuged at 10,000 RPM (24,000 g) using a
28
29 Beckman Coulter Avanti J-26 XP centrifuge, with a JLA-16.500 rotor, to remove fine
30
31 solid particles which were added back to the solid stream. The wash stream was used for
32
33 further oligosaccharide analysis.
34
35
36
37
38
39
40

41 **Enzymatic Hydrolysis.** The NREL standard protocol (LAP-009) was followed for
42
43 enzymatic hydrolysis. All samples were hydrolyzed in a 0.05 M citrate buffer (pH 4.8) at
44
45 1% glucan loading with the necessary commercial cellulase enzyme (Spezyme CP
46
47 generously provided by Genencor, CAS 9012-528) and β -glucosidase (Novo 188,
48
49 Novozyme). All the enzymes were stored at 4 $^{\circ}\text{C}$ for until further use. Certain samples
50
51 were also hydrolyzed using commercial xylanases (Multifect Xylanase, Genencor). The
52
53 hydrolyzed samples were boiled to denature the enzymes and filtered through a 0.2
54
55
56
57
58
59
60

1
2
3 micron nylon membrane filter at predetermined time periods (72 and 168 h). The samples
4
5 were frozen for subsequent high performance liquid chromatography (HPLC) sugar
6
7 analysis. Sampling was performed at two intervals (72 and 168h) to determine glucan and
8
9 xylan conversions. The protein concentrations of the enzymes were determined by the
10
11 BCA protein assay (Pierce, Rockford, IL). The protein concentrations of the respective
12
13 enzymes were as follows; Spezyme CP (123 mg/ml; 59 FPU/ml, where FPU is filter
14
15 paper units), β -glucosidase (130 mg/ml) and Multifect xylanase (42 mg/ml).
16
17
18
19

20
21
22 **Mass Balance:** A mass balance for AFEX pretreatment and enzymatic hydrolysis (1%
23
24 glucan) of corn stover and poplar was performed starting with 100 grams (dry weight) of
25
26 biomass. Experiments were done in duplicates, with standard deviations less than 5%.
27
28 For each process step, the glucan, xylan and arabinan compositions of the solid and liquid
29
30 streams were determined using the NREL LAP protocol. For poplar, either 31.3 or 125
31
32 mg protein/g of glucan of cellulase was used, while, for corn stover only 31.3 mg
33
34 protein/g of glucan of cellulase was used. In both cases 33.3 mg protein/g of glucan of β -
35
36 glucosidase was added to prevent cellobiose inhibition. In addition, xylanase was also
37
38 supplemented for corn stover (3.1 mg protein/g of glucan) and poplar (31.3 or 125 mg
39
40 protein/g of glucan). The total masses of each enzyme added (including cellulase, β -
41
42 glucosidase and xylanase) are also shown in the mass balance. These experiments were
43
44 run in duplicates and showed standard deviations below 5%.
45
46
47
48
49
50

51
52
53 **LC-MS/MS Analysis of Degradation Products after AFEX:** Analytical
54
55 characterization of organic degradation products was carried out at Baylor University.
56
57
58
59
60

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Details of the analytical methodology have been reported elsewhere (37) and are summarized below. Degraded products in the pretreated solids were initially extracted with water at 70 °C using an ASE-200 accelerated solvent extraction apparatus (Dionex Corp., Sunnyvale, CA, USA). The pH of the solution was found to be alkaline and was acidified to pH 2 before processing as given in the protocol (38). A 5-ml aliquot of each aqueous extract (hereafter referred to as the aqueous wash stream) was then extracted two times with methyl *tertiary*-butyl ether (MTBE) using the procedure reported by Chen *et al.* (38). The resulting MTBE phases were combined and solvent was evaporated at 55 °C under a gentle stream of nitrogen. All samples were reconstituted in 5 ml water prior to analysis.

Instrumentation employed for analysis consisted of a Varian (Palo Alto, CA, USA) ProStar Model 210 binary pump system, Model 410 auto sampler, and Model 1200L triple-quadrupole mass analyzer. A binary solvent gradient consisting of two solvents A and B. Solvent A consists of aqueous formic acid (0.025% (v/v) formic acid in water) and solvent B consists of 10% aqueous formic acid and 90% acetonitrile was used to achieve chromatographic separation on a 150 mm × 4.6 mm (S 03 μm, 99) YMC Carotenoid column (Waters, Milford, MA, USA) connected in series to a 1 mm RP C18 OPTI-Guard column (Altech, Deerfield, IL, USA). Additional chromatographic parameters were as follows: injection volume, 50 μL; column temperature, 30 °C; flow rate, 750 μL/min. It is important to note that these chromatographic conditions are similar to those reported in previous work. The only major difference was the substitution of formic acid for phosphoric acid. This change was implemented to improve mobile phase compatibility

1
2
3 with MS detection. Upon exiting the column, the mobile phase was directed to both a
4
5 UV-visible photodiode array (PDA) detector and the mass analyzer, which was operated
6
7 exclusively in negative electro spray ionization (–ESI) mode. The majority of target
8
9 analytes were assessed by monitoring an optimized MS/MS for each compound, with
10
11 parent ion $[M - H]^-$ selected in the first quadrupole. A microL flow-splitter was inserted
12
13 between the PDA detector and the mass spectrometer such that the volume of liquid
14
15 passing through the flow-splitter was diverted 50:50 between the mass analyzer and the
16
17 waste line. Mass spectrometry parameters held constant during all experiments were as
18
19 follows: nebulizing gas, O₂ at 60 psi; drying gas, N₂ at 20 psi; drying gas temperature,
20
21 400 °C; needle voltage, 4500 V; collision gas, Ar at 2.0 mTorr. Exceptions include acetic
22
23 acid, furfural, and 5-hydroxymethylfurfural (5-HMF) which were not amenable to mass
24
25 spectral monitoring under these conditions and were instead detected via UV
26
27 spectroscopy.
28
29
30
31
32
33
34
35
36

37 **Accelerated solvent extractor (ASE) protocol:** The extraction was done using a Dionex
38
39 ASE 200 extractor at 70°C with two cycles of water. The samples used for each
40
41 extraction were between 0.5-1 gm (in 11 ml cell) at 70 °C and 1500 psi with static time
42
43 for 10 mins and purge time for 60 seconds.
44
45
46
47

48 **Quantitation of target analytes by mass spectrum analysis:** This was accomplished
49
50 using a multipoint, internal-standard calibration curve. Calibration standards were
51
52 prepared by successive dilutions of a stock solution consisting of the neat chemicals in
53
54 water. Aliquots of each calibration standard were extracted with MTBE prior to analysis
55
56
57
58
59
60

1
2
3 (as described above). Response factors were determined for each analyte by dividing the
4
5 peak area of the analyte by the peak area of the internal standard, and calibration curves
6
7 were constructed by plotting a linear regression ($r^2 \geq 0.99$) of response factor versus
8
9 analyte concentration. It has been previously demonstrated (38) that this approach to
10
11 quantitation does not require independent knowledge of extraction efficiencies in order to
12
13 assess analyte concentrations. However, the approach does not correct for the potential
14
15 influence of co-extracted matrix components on electro spray ionization.
16
17
18
19

20
21
22 For this reason, data quality was assessed via analysis of a matrix spike for each analyzed
23
24 sample. Calculated spike recoveries (data not shown) revealed negligible matrix
25
26 influence in samples derived from untreated poplar (i.e., spike recoveries were essentially
27
28 quantitative in these samples). More pronounced matrix interference was identified for
29
30 some analytes in both samples derived from pretreated materials, and matrix effects were
31
32 most pronounced in the sample derived from high-lignin poplar. Nevertheless, spike
33
34 recoveries demonstrated that analyte concentrations derived from calibration curves were
35
36 accurate within a factor of 1-3, independent of sample type, and this was deemed
37
38 sufficient to support the goals of the present study.
39
40
41
42
43
44
45

46 **Monomeric and Oligomeric Sugar Analysis:** A high performance liquid
47
48 chromatography (HPLC) system was used for sugar analysis. The HPLC system
49
50 consisted of Waters (Milford, MA) Pump and Waters 410 refractive index detector. An
51
52 Aminex HPX-87P carbohydrate analysis column (BioRad, Hercules, CA) equipped with
53
54 a deashing guard cartridge (BioRad) was used for quantifying sugars in hydrolyzate.
55
56
57
58
59
60

1
2
3 Degassed HPLC grade water was used as the mobile phase at 0.6 ml/min at a column
4
5 temperature of 85 °C. The injection volume was 10 µl with a run time of 20 min. Mixed
6
7 sugar standards were used to quantify cellobiose and other monosaccharides (glucose,
8
9 xylose, galactose, arabinose and mannose) in the samples.
10
11

12
13
14
15 Oligosaccharides in the liquid stream were quantified by acid hydrolysis based on the
16
17 NREL LAP protocol (<http://www.nrel.gov/biomass/pdfs/42623.pdf>). The monomeric
18
19 sugars produced after acid hydrolysis were quantified by high-performance liquid
20
21 chromatography (HPLC) using a Bio-Rad Aminex HPX-87H ion exclusion column
22
23 (60°C; 5 mM H₂SO₄; flow rate, 0.6 ml min⁻¹; injection volume, 10 µl) and differential
24
25 refractive index detector.
26
27
28

29
30
31 **FT-IR ATR analysis:** A Spectrum One FTIR system (Perkin Elmer, Wellesley, MA)
32
33 with a universal ATR (Attenuated Total Reflection) accessory was used to qualitatively
34
35 monitor chemical changes in the AFEX treated and untreated poplar and corn stover
36
37 respectively. The samples were pressed uniformly and tightly against the diamond
38
39 surface using a spring-loaded anvil. Mid-IR spectra were obtained by averaging 4 or 16
40
41 scans from 4,000 to 600 cm⁻¹ at 2 cm⁻¹ resolution. Baseline and ATR corrections for
42
43 penetration depth and frequency variations were carried out using the Spectrum One
44
45 software supplied with the equipment. The region between 1,550 and 1800 cm⁻¹ was
46
47 selectively monitored to check the ester and amide linkage stretching frequency at 1740
48
49 and 1664 cm⁻¹ respectively.
50
51
52
53
54
55
56
57
58
59
60

Results and Discussion:

Poplar AFEX Optimization. In order to optimize the AFEX condition for poplar, both moisture and temperature were varied for fixed ammonia to biomass loadings of 1:1 (w/w). Pretreated samples were tested at 1% glucan loading (15 ml reaction volume) using 31.3 mg of cellulase and 33.3 mg of β -glucosidase per gram of glucan at 50 °C over a period of 168 hours. As we raised the pretreatment temperature from 120 °C to 200 °C we saw a steady increase in glucan conversions. Further increases in glucan conversion were also noticed when we raised the moisture from 50% to 233% (dwb, dry weight basis of feedstock) (Figure 1)

Based on preliminary AFEX optimization studies we found that higher glucan conversions were obtained for AFEX done at 180 °C and 233% (dwb) moisture. To study the effect of varying ammonia to biomass loadings the pretreatment temperature and moisture was held at 180 °C and 233% (dwb) moisture, respectively. Of the three ammonia:biomass loadings (i.e. 1:1, 2:1 and 3:1, w/w), 2:1 loading gave the highest glucan conversion (Figure 1). Compared to acid pretreatment, AFEX is a relatively dry to dry process where both cellulose and hemicellulose are retained in the solids stream. Commercially available cellulases (e.g. Spezyme CP) do not have sufficient hemicellulase activity to adequately digest AFEX treated biomass (39). Therefore, we supplemented it with xylanases (using Multifect Xylanase) at varying concentrations (0-100% of the total milligrams of cellulase protein) (figure not shown). Increasing xylanase supplementation increases both glucan and xylan conversion. There is however little

1
2
3 improvement noticed when xylanase loadings were beyond 100% of the total cellulase
4 protein loadings. For the sake of simplicity we have only shown enzymatic hydrolysis
5 results for xylanase supplementation at 100% of total cellulase protein. About 15%
6 improvement in glucan conversion was noticed for enzymatic hydrolysis done with
7 xylanase supplementation when compared to using cellulase alone. Xylanase
8 supplementation aids the removal of xylan polymer embedded in the cellulose matrix
9 which in turn synergistically helps improve cellulase accessibility to cellulose (39). In
10 addition, 125 mg per gram of glucan of cellulase loading produced about 25% higher
11 glucan conversion when compared to 31.3 mg of cellulase per gram of glucan for AFEX
12 treated Poplar (Figure 1).
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28

29 **AFEX Pretreatment for Poplar vs. Corn Stover.** It is well known that grasses
30 (monocots) and woody (dicots) species have a complex cell wall structure which are
31 quite different from each other. Hardwoods are a good source of cellulosic fiber (higher
32 than stover cellulose content) and their lignin and monolignol composition are very
33 different from corn stover. Our results show that poplar needed much higher temperature,
34 moisture and ammonia loadings during AFEX pretreatment in order to achieve
35 significant glucan hydrolysis yields (using 31.3 mg of cellulase and 33.3 mg of β -
36 glucosidase per gram of glucan) compared to corn stover. Both untreated and AFEX
37 treated corn stover hydrolysis results were similar to the one we reported earlier (22).
38 Even higher temperatures (e.g. 180 °C) could yield close to 50% glucan and 35% xylan
39 conversion, with 1:1 ammonia to biomass loadings. (Figure 2). For corn stover, the best
40 AFEX condition was found to be close to 90 °C, 60% moisture and 1:1 ammonia to
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 biomass loadings. Closer inspection of the biomass composition (Table 1), reveals that
4
5 corn stover has lower glucan, lignin and higher xylan, arabinan and ash content compared
6
7 to poplar. Corn stover was also found to contain higher extractives (8.5%, of which 2.2%
8
9 is sucrose) compared to poplar (~3.4%). Of the various components in biomass, it has
10
11 been demonstrated that the percentage of lignin in biomass will influence the enzymatic
12
13 hydrolysis (40). Since ammonia cleaves the arabinoxylan and acetyl-xylan ester linkages,
14
15 their composition in biomass appear to be a determining factor on ultimate sugar yields
16
17 during pretreatment using ammonia (41-42).
18
19
20
21
22
23

24 Lignin, an aromatic polymer, is one of the major components contributing to biomass
25
26 recalcitrance. Lignin binds irreversibly to enzymes and reduces available enzyme
27
28 activity during hydrolysis (43). Pretreatment improves enzyme accessibility by cleaving
29
30 certain lignin and lignin-hemicellulose cross-linkages (8). In acidic pretreatments and
31
32 organosolv processing, most of the hemicellulose and some of the lignin is hydrolyzed
33
34 and chemically extracted from the insoluble cellulose matrix. AFEX also produces some
35
36 lignin derived products but does not physically separate them into a separate stream.
37
38 Water washing the biomass prior to enzymatic hydrolysis improves glucan conversion by
39
40 up to 7% for corn stover (22) and by up to 6% for poplar for low enzyme loading (results
41
42 not shown). Multiple explanations are possible for this observation including: 1)
43
44 presence of enzyme-inhibiting products produced during AFEX pretreatment, 2) partial
45
46 lignin removal leading to less enzyme adsorption to lignin or 3) opening up of the
47
48 cellulose-hemicellulose-lignin matrix thereby allowing easier penetration of enzymes.
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 In addition to lignin, the arabinoxylan composition of monocots is an important factor
4 affecting digestibility of the biomass. Most of the arabinoxylan in the cell wall is
5
6 connected to lignin through ether and ester linkages (5, 14-17). These linkages are
7
8 typically between glucuronic acid and arabinose side chains of hemicelluloses and
9
10 hydroxyl/carboxyl functionalities of lignin (e.g. ferulic and coumaric acid). Ammonia has
11
12 tendency to cleave these ester linkages (18, 20) via ammonolysis. It is interesting to note
13
14 that the arabinoxylan content of grasses (3-6%) is significantly higher than hardwoods
15
16 (<1%, for poplar). With a higher arabinoxylan content of the cell wall, more ester
17
18 linkages are likely to be cleaved during AFEX pretreatment. Arabinoxylans help form
19
20 bridges between lignin and hemicellulose/cellulose that reduce enzyme accessibility.
21
22 Cleavage of these linkages would help increase the cell wall pore volume, reduce the
23
24 protective barrier of lignin and enhance enzyme accessibility to cellulose and
25
26 hemicellulose. This is one factor that could explain why grasses are more easily
27
28 digestible after AFEX pretreatment compared to hardwood poplar. Further support for
29
30 this hypothesis would come from detailed quantification of oligosaccharides and other
31
32 degradation products using Matrix-Assisted Laser Desorption Ionization-Time Of
33
34 followed by mass spectroscopy (MALDI-TOFMS) produced during AFEX pretreatment
35
36 for both corn stover and poplar.
37
38
39
40
41
42
43
44
45
46
47

48 **Mass Balance.** A complete mass balance for AFEX pretreatment and enzymatic
49
50 hydrolysis was done for both corn stover and poplar. For poplar, enzymatic hydrolysis
51
52 was performed either using 31.3 or 125 mg of cellulase per gram of glucan, while for
53
54 corn stover only 31.3 mg of cellulase per gram of glucan was used. In both cases, 33.3
55
56
57
58
59
60

1
2
3 mg of β -glucosidase per gram of glucan was used to prevent cellobiose inhibition of the
4
5 cellulases. In addition, xylanase was supplemented for both corn stover (3.1 mg/g of
6
7 glucan) and poplar (either 31.3 or 125 mg/g of glucan). Both for corn stover and poplar
8
9 the mass balance was done for both unwashed washed material after AFEX pretreatment
10
11 and prior to enzymatic hydrolysis. All enzymatic hydrolysis experiments were done in
12
13 duplicates, at a 100 ml scale with 1% glucan loading for 168 h at 50 °C (Figure 4 and 5).
14
15 Close to 95% mass closure was achieved both for corn stover and poplar. For poplar,
16
17 about 66 % glucan and 44% xylan conversion (31.3 mg of cellulase, 33.3 mg of β -
18
19 glucosidase and 31.3 mg of xylanase per gram of glucan) was noticed for lower enzyme
20
21 loading in unwashed samples. On the other hand, for higher enzyme loadings (125 mg of
22
23 cellulase, 33.3 mg of β -glucosidase and 125 mg of xylanase per gram of glucan), we get
24
25 93% glucan and 65% xylan conversion respectively. When we washed poplar sample
26
27 between pretreatment and enzymatic hydrolysis we see an improvement (~5%) in glucan
28
29 and xylan conversion at lower enzyme loading. However, no improvement was noticed in
30
31 glucan conversion when we increased the enzyme loading up to 125 mg of cellulase and
32
33 xylanase per gram of glucan (Figure 4).
34
35
36
37
38
39
40
41
42
43

44 For corn stover (using 31.3 mg of cellulase, 33.3 mg of β -glucosidase and 1.4 mg of
45
46 xylanase/gram of glucan), the glucan and xylan conversions for unwashed samples (88%
47
48 and 68%) and washed samples (93% and 66%) are shown in Figure 5. Compared to
49
50 unwashed corn stover, we see an increase in glucan and a slight decrease in xylan
51
52 conversion for washed samples. As reported earlier (44, 45) washing helps reduce
53
54
55
56
57
58
59
60

1
2
3 removal of several degradation products like acetic acid, phenolic acids, HMF, furfural
4
5 which are potentially inhibitory during enzymatic hydrolysis.
6
7

8
9
10 **Oligosaccharide analysis.** The wash streams generated from untreated/AFEX treated
11 poplar and corn stover were analyzed for monomeric and oligomeric sugars using the
12 Aminex 87P column giving rather interesting results. The untreated poplar wash stream
13 had a little monomeric/polysaccharide, while untreated corn stover had approximately
14 26.0, 0 and 0.12 mg of glucose, xylose and arabinose per gram of biomass respectively
15 (Table 2). We also saw an increase in glucose, xylose and arabinose concentration of
16 27.8, 1.22 and 0.39 mg respectively per gram of biomass after acid hydrolysis of the
17 untreated corn stover wash stream. This confirms the presence of very low concentrations
18 of short-chain saccharides (mostly DP 2 based on Aminex 87P chromatograms, data not
19 shown) in untreated corn stover, unlike poplar.
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35

36 The AFEX pretreated poplar wash stream also showed very low concentrations of
37 monomeric sugars prior to acid hydrolysis. After acid hydrolysis, we observe a small
38 increase in glucose concentration and about 54 fold increases in xylose concentration and
39 0.8 fold increases in arabinose concentration (Table 2). Compared to untreated corn
40 stover, AFEX pretreated samples have decreased monomeric glucose concentration by up
41 to 6.6 while there is not much change in xylose and arabinose concentration. The
42 reduction in soluble sugar concentration during AFEX pretreatment could be due to alkali
43 induced degradation of these monomeric sugars (46). Upon performing mild acid
44 hydrolysis of AFEX treated corn stover wash stream, we observe about 2.9, 300 and 79
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 fold increase in glucose, xylose and arabinose concentrations, respectively. Our present
4
5 results are in accordance with previous work (47, 48) in which aqueous ammonia treated
6
7 corn plant fractions produced more arabinoxylans in the wash stream. Hemicellulose
8
9 contains multiple side chains, rich in arabinose as the linker molecule bound to other
10
11 hemicellulose side chains or phenolic acids bound to lignin, that cause steric inhibition of
12
13 enzymes (49). AFEX cleaves these lignin-hemicellulose ester linkages, thereby releasing
14
15 water extractable arabinoxylan rich oligomers along with other phenolics. The wash
16
17 stream is now being further analyzed to determine the effect of biomass source and
18
19 AFEX pretreatment severity on the nature of oligosaccharides obtained.
20
21
22
23
24
25
26

27 **FT-IR Analysis.** Further evidence confirming cleavage of ester linkages in both corn
28
29 stover and poplar comes from FT-IR analysis. Both untreated and AFEX treated samples
30
31 were analyzed for various stretching frequencies in the region 1550 to 1800 cm^{-1} . Some
32
33 of the important peaks identified from literature that can serve to illustrate the effect of
34
35 AFEX on biomass composition are the ester carbonyl peak at 1740 cm^{-1} , the aldehyde
36
37 peak at 1640 cm^{-1} and amide linkages at 1664 cm^{-1} (18, 19) The ester-carbonyl bonds are
38
39 typically present in the hemicellulose and hemicellulose–lignin complexes. A decrease in
40
41 these peaks at 1740 and 1640 cm^{-1} are directly related to deesterification of hemicellulose
42
43 during AFEX (18-20). From figure 3 we can see that the stretching ester carbonyl
44
45 frequency at 1740 cm^{-1} totally disappears upon AFEX pretreatment. A 1640 cm^{-1} peak
46
47 appears in both AFEX pretreated samples of corn stover and poplar. This observation
48
49 further confirms the fact that hemicellulose based ester linkages are ammonolyzed,
50
51
52
53
54
55
56
57
58
59
60 resulting in the formation of their respective amides during AFEX.

1
2
3
4
5
6 **Degradation product analysis by LC-MS.** ASE/water extractions of AFEX treated
7
8 samples gave insight into the mechanism of AFEX, which was further explored by
9
10 qualitative and quantitative analysis of the extract. The protocol employed for water
11
12 washing the biomass is reported elsewhere (22). Approximately 6-8% and 13-15% by
13
14 mass (based on initial dry weight) of the untreated and AFEX treated corn stover,
15
16 respectively was lost in the washing step. These results are also consistent with
17
18 previously reported water wash data (weight loss of 12%, based on dry weight of
19
20 biomass) for super/sub-critical ammonia treated birch wood (50, 51). However, the wash
21
22 extractive for ammonia treated birch wood was found to be largely acetamide (5-7%
23
24 based on dry wt) formed due to the ammonolysis of the heavily acetylated hemicellulose
25
26 under severe reaction conditions. Detailed compositional analysis of the wash extractives
27
28 for AFEX treated poplar showed an interesting range of compounds as explained below.
29
30
31
32
33
34
35

36 Comparison of ASE extracts derived from untreated and AFEX-treated poplar
37
38 demonstrated that the concentrations of certain compounds increased several folds
39
40 following AFEX pretreatment. Analytical concentrations ($\mu\text{g/g}$ dry weight of extracted
41
42 material) of components monitored by LC-MS/MS are shown in Table 3. These data
43
44 demonstrate that AFEX treatment resulted in the production of a variety of aliphatic
45
46 organic acids. As compared to samples derived from untreated poplar, concentrations of
47
48 lactic, malonic, methylmalonic, succinic and levulinic acids in wash streams derived from
49
50 treated materials were increased between 10 and 190 fold. The highest concentrations
51
52 were observed for lactic acid, which is known to be formed via alkali induced peeling and
53
54
55
56
57
58
59
60

1
2
3 terminal degradation of polymeric sugars (23). Succinic acid was also present at
4 relatively high concentrations. Furfural and 5-HMF were not detected in AFEX treated
5 samples. However, a noticeable increase in the concentration of 2-furoic acid was
6 observed in both AFEX treated samples as compared to the untreated materials. It is
7 possible that furoic acid could have been formed directly via hydrolytic/oxidative
8 cleavage of lignin. The cleavage of the acetyl groups in hemicellulose and lignin also
9 resulted in increased concentrations of acetic acid in the AFEX wash streams (data not
10 shown). Sugar-derived aldehydes were not monitored as part of this work, as they are
11 expected to undergo condensation reactions under high temperature and alkaline
12 conditions (52). Some of these compounds were found to inhibit microbes during ethanol
13 fermentation (53, 54).
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31

32 AFEX also released measurable amounts of various aromatic acids and aldehydes,
33 presumably due to base-catalyzed cleavage of lignin polymers. A significant increase in
34 the phenolic content of the wash streams was observed following AFEX treatment, as
35 indicated by increased concentrations of salicylic acid, 4-hydroxybenzaldehyde, syringic
36 acid, vanillin, vanillic acid, homovanillic acid, 4-hydroxyacetophenone, benzoic acid and
37 syringaldehyde. Phenolic acids such as 4-hydroxycoumaric acid and ferulic acid are
38 expected to be produced upon hydrolysis of hemicellulose-lignin ester cross-links (25).
39 However, relatively low concentrations of these compounds were noticed for AFEX-
40 treated poplar. It is possible that these components were first formed and then further
41 degraded via hydrolytic/oxidative cleavage at the more severe poplar pretreatment
42 conditons, resulting in lower observed concentrations. Phenolic aldehydes are more
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 likely to be produced in oxidative alkaline conditions (24). Nevertheless, some
4
5 components such as 4-hydroxybenzaldehyde and 4-hydroxybenzoic acid are thought to
6
7 be monomeric lignin extractives that are easily released after the AFEX process, unlike
8
9 other lignin cleavage products (26, 27).
10
11
12
13

14 **Conclusion:**

15
16
17
18
19 Varying AFEX pretreatment conditions and enzyme combinations were tested on poplar
20
21 and corn stover. Based on different AFEX conditions tested, it was found that the optimal
22
23 AFEX conditions for poplar (2:1 ammonia to biomass loading, 233% moisture on dwb
24
25 and 180 °C) and for corn stover (1:1, ammonia to biomass loading, 60% moisture and 90
26
27 °C) respectively. Adding xylanase enzymes along with commercial cellulase preps
28
29 improved both glucan and xylan conversion both for poplar and corn stover. Complete
30
31 mass balance for both pretreatment and hydrolysis has been shown for both poplar and
32
33 corn stover. Based on the present results, corn stover required much less severe AFEX
34
35 conditions (i.e. less ammonia and lower treatment temperatures) compared to poplar.
36
37 These differences have been correlated to both lignin and arabinoxylan content of the
38
39 biomass. The ester linkages connecting arabinoxylan to lignin phenolics are broken
40
41 during AFEX based on evidence from FT-IR and wash stream oligosaccharides analysis.
42
43 In addition to oligosaccharides, several aliphatic and aromatic organic acids were also
44
45 generated from both high and low lignin poplar. These were quantified using a recently-
46
47 developed LC-MS/MS methodology. Required pretreatment severity and enzyme
48
49 consumption both might be significantly reduced by making alterations to several cell
50
51
52
53
54
55
56
57
58
59
60

1
2
3 wall components (e.g. lignin and arabinoxylan content). Comparison of other herbaceous
4
5 and woody species will help us better understand the relationships between biomass
6
7 composition, cell wall ultra-structure to effectiveness of AFEX pretreatment and
8
9 enzymatic hydrolysis.
10
11

12 13 14 15 **Acknowledgements:**

16
17
18
19
20 We would like to acknowledge Professor Charles Wyman (UC Riverside) and other
21
22 CAFI-II team collaborators for useful criticism and helpful insights. The project was
23
24 funded by the U.S. Department of Energy (contract DE-FG36-04GO14017). The
25
26 participation of Lekh Sharma and Dr. Kevin Chambliss was supported by the National
27
28 Research Initiative of the USDA Cooperative State Research, Education and Extension
29
30 Service, grant number 2005-35504-16335. We also thank James Heidenreich and Dona
31
32 Hardy for their support during the initial stages of this project, Genencor International
33
34 (Rochester, NY) for supplying commercial enzymes and Rajesh Gupta (Auburn
35
36 University) for conducting compositional analysis on AFEX treated poplar.
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

References:

1. Ohara, H. Biorefinery, *Appl. Microbiol. Biotechnol.* **2003**, *62*, 474–477.
2. Ragauskas, A. J.; Williams, C. K.; Davison, B. H.; Britovsek, G.; Cairney, J.; Eckert, C. A.; Frederick Jr, W. J.; Hallett, J. P.; Leak, D. J.; Liotta, C. L.; Mielenz, J. R.; Murphy, R.; Templer, R.; Tschaplinski, T. The Path Forward for Biofuels and Biomaterials. *Science* **2006**, *311*, 484-489.
3. Gray, K. A.; Zhao, L.; Emptage, M. Bioethanol. *Current Opinion in Chemical Biology* **2006**, *10*, 141–146.
4. Mantanis, G. I.; Young, R. A.; Rowell, R. M. Swelling of compressed cellulose fiber webs in organic liquids. *Cellulose* **1995**, *2*, 1-22.
5. Bidlack, J.; Malone, M.; Benson R. Molecular Structure and Component Integration of Secondary Cell Walls in Plants. *Proc. Okla. Acad. Sci.* **1992**, *72*, 51-56.
6. Somerville, C.; Bauer, S.; Brininstool, G.; Facette, M.; Hamann, T.; Milne, J.; Osborne, E.; Paredes, A.; Persson, S.; Raab, T.; Vorwerk, S.; Youngs, H. Toward a Systems Approach to Understanding Plant Cell Walls. *Science* **2004**, *306*, 2206 – 2211.
7. Grabber, J. H. How Do Lignin Composition, Structure, and Cross-Linking Affect Degradability? A Review of Cell Wall Model Studies. *Crop Sci.* **2005**, *45*, 820–831.
8. Mosier, N.; Wyman, C.; Dale, B. E.; Elander, R.; Lee, Y. Y.; Holtzapple, M.; Ladisch, M. Features of promising technologies for pretreatment of lignocellulosic biomass. *Bioresour. Technol.* **2005**, *96*, 673-686

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
9. Dale, B. E, Method for increasing the reactivity and digestibility of cellulose with ammonia. **1986**, US patent No. 4600590.
10. Eggeman, T.; Elander, R. T. Process and economic analysis of pretreatment technologies. *Bioresource Technol.* **2005**, *96*, 2019–2025.
11. Teymouri, F. Laureano-Perez, L.; Alizadeh, H.; Dale, B. E. Optimization of the ammonia fiber explosion (AFEX) treatment parameters for enzymatic hydrolysis of corn stover. *Bioresource Technol.* **2005**, *96*, 2014–2018.
12. Chang, M. C. Harnessing energy from plant biomass. *Curr Opin Chem Biol.* **2007**, *11*, 677-684.
13. Chen, F.; Dixon, R.A. Lignin modification improves fermentable sugar yields for biofuel production. *Nature Biotechnology* 2007, *25*, 759 – 761.
14. Jeffries, T. W. Biodegradation of lignin and hemicelluloses. *Biochemistry of Microbial Degradation*, C. Ratledge (ed.), Kluwer Academic Publisher, Netherlands, **1997**, p233–277.
15. Hatfield, R. D.; Ralph, J.; Grabber, J. H. Cell wall cross-linking by ferulates and diferulates in grasses. *J. Sci. Food Agric.* **1999**, *79*, 403-407.
16. Cosgrove, G. J. Growth of the plant cell wall. *Nature reviews* **2005**, *6*, 850 - 861.
17. Ralph J. What Makes a Good Monolignol Substitute? In *The science and Lore of the Plant cell wall: Biosynthesis, structure and function*, Hayashi T (eds.) Brown walker press, Boca Raton. **2006**, p 285-293.
18. Buettner, M. R.; Lechtenberg, V.L.; Hendrix, K. S.; Hertel, J. M. Composition and Digestion of ammoniated tall fescue Hay. *J. Animal Science* **1982**, *54*, 173-178.

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
19. Pawlak, Z.; Pawlak, A.S. A Review of Infrared Spectra from Wood and Wood Components Following Treatment with Liquid Ammonia and Solvated Electrons in Liquid Ammonia. *Applied Spectroscopy Reviews* **1997**, *32*, 349–383.
 20. Rosca, L.; Puhringer, R.; Schmidt, H.; Tanczos, I. New aspects in studying and application of ammonia treatment of softwood. Kudela J, Kurjatko S (edt), *Wood Structure and Properties* Arbora publishers. **2002**, p 127-129.
 21. Grabber JH, Hatfield RD, Lu F, Ralph J. Coniferyl ferulate incorporation into lignin enhances the alkaline delignification and enzymatic degradation of cell walls. *Biomacromolecules*. **2008**, *9*, 2510-2516.
 22. Chundawat, P.S.; Venkatesh, B.; Dale, B. E. Effect of Particle Size Based Separation of Milled Corn Stover on AFEX pretreatment and Enzymatic Digestibility. *Biotechnol. Bioeng.* **2007**, *96*, 219-231.
 23. Sjöström, E. Carbohydrate degradation products from alkaline pretreatment of biomass. *Biomass Bioenergy* **1991**, *1*, 61–64.
 24. Klinke, H.B.; Ahring, B.K.; Schmidt, A.S.; Thomsen, A. B. Characterization of degradation products from alkaline wet oxidation of wheat straw. *Bioresour. Technol.* **2002**, *82*, 15–26.
 25. Lawther, J. M.; Sun, R. The fractional characterisation of polysaccharides and lignin components in alkaline treated and atmospheric refined wheat straw. *Ind. Crops Prod.* **1996**, *5*, 87–95.
 26. Baeza, J.; Freer, J. Chemical characterization of wood and its components. In: Hon DNS, Shirashi N (eds) *Wood and cellulosic chemistry*. Dekker, New York, **1997**, p 275–374.

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
27. Jönsson, L. J.; Palmqvist, E.; Nilvebrant, N. O.; Hahn-Hägerdal, B. Detoxification of wood hydrolysates with laccase and peroxidase from the white-rot fungus *Trametes versicolor*. *Appl. Microbiol. Biotechnol.* **1998**, *49*, 691–697
28. Leonard, R. H.; Hajny, G. J. Fermentation of wood sugars to ethyl alcohol. *Ind. Eng. Chem.* **1945**, *37*, 390–395.
29. Sierra-Alvarez, R.; Lettinga, G. The methanogenic toxicity of wastewater lignins and lignin related compounds. *J. Chem. Tech. Biotechnol.* **1991**, *50*, 443–455.
30. Palmqvist, E.; Hahn-Hägerdal, B. Fermentation of lignocellulosic hydrolysates. I. Inhibition and detoxification. *Bioresour. Technol.* **2000**, *74*, 17–24.
31. Cantarella, M.; Cantarella, L.; Gallifuoco, A.; Spera, A.; Alfani, F.; Effect of inhibitors released during steam-explosion treatment of poplar wood on subsequent enzymatic hydrolysis and SSF. *Biotechnol Prog.* **2004**, *20*, 200-206.
32. Pan, X.; Gilkes, N.; Kadla, J.; Pye, K.; Saka, S.; Gregg, D.; Ehara, K.; Xie, D.; Lam, D.; Saddler, J. Bioconversion of hybrid poplar to ethanol and co-products using an organosolv fractionation process: optimization of process yields. *Biotechnol. Bioeng.* **2006**, *94*, 851-61.
33. Esteghlalian, A.; Hashimoto, A. G.; Fenske, J. J.; Penner, M. H. Modeling and optimization of the dilute-sulfuric-acid pretreatment of corn stover, poplar and switchgrass. *Bioresource Technology.* **1997**, *59*, 129-136.
34. Davison, B.H.; Drescher, S. R.; Tuskan, G. A.; Davis, M. F.; Nghiem, N. P. Variation of S/G ratio and lignin content in a *Populus* family influences the release of xylose by dilute acid hydrolysis. *Appl. Biochem. Biotechnol.* **2006**, *129-132*, 427-435.

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
35. Chang, V. S.; Nagwani, M.; Kim, C. H.; Holtzapple, M. T. Oxidative lime pretreatment of high-lignin biomass: poplar wood and newspaper. *Appl. Biochem. Biotechnol.* **2001**, *94*, 1-28.
36. LAP protocol available at National Renewable Energy Lab website (http://www.nrel.gov/biomass/analytical_procedures.html).
37. Sharma, L. N.; Becker, C.; Chambliss, C. K. Analytical characterization of fermentation inhibitors in biomass pretreatment samples using liquid chromatography, UV-visible spectroscopy, and tandem mass spectrometry," *Methods in Molecular Biology* **2008** (in press).
38. Chen, S-F.; Mowery, R. A. Castleberry VA, van Walsum GP, and Chambliss CK High-performance liquid chromatography method for simultaneous determination of aliphatic acid, aromatic acid and neutral degradation products in biomass pretreatment hydrolysates. *J. Chromatogr. A*, **2006**, *1104*, 54-61.
39. Dien BS, Ximenes EA, O'Bryan PJ, Moniruzzaman M, Li XL, Balan V, Dale B, Cotta MA Enzyme characterization for hydrolysis of AFEX and liquid hot-water pretreated distillers' grains and their conversion to ethanol. *Bioresour. Technol.* **2007** (ahead of print, doi:10.1016/j.biortech.2007.09.030).
40. Vanholme, R.; Morreel, K.; Ralph, J. And Boerjan, W. Lignin engineering. *Current opin. plant Biol.* 2008, *11*, 278-285.
41. Lizbeth L-P, Teymouri, F.; Alizadeh, H.; Bruce E Dale, B. E. Understanding factors that limit enzymatic hydrolysis of biomass: characterization of pretreated corn stover. *Appl Biochem Biotechnol.* **2005**, 121-124, 1081-1099.

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
42. Laureano-Perez L, Dale BE, Zhu L, O'Dwyer JP, Holtzapple M. Statistical correlation of spectroscopic analysis and enzymatic hydrolysis of poplar samples. *Biotechnol Prog.* **2006**, 22, 835-841.
43. Sutcliffe, R. and Saddler, J.N. The role of lignin in the adsorption of cellulases during enzymatic treatment of lignocellulosic material. *Biotechnol. Bioeng. Symp.* 1986, **17**, 749–762.
44. Maria Cantarella, M.; Cantarella, L.; Gallifuoco, A.; Spera, A.; and Alfani, F. Effect of Inhibitors Released during Steam-Explosion Treatment of Poplar Wood on Subsequent Enzymatic Hydrolysis and SSF. *Biotechnol. Prog.* **2004**, 20, 200-206.
45. Sakai S, Tsuchida Y, Nakamoto H, Okino S, Ichihashi O, Kawaguchi H, Watanabe T, Inui M, Yukawa H. Effect of lignocellulose-derived inhibitors on growth of and ethanol production by growth-arrested *Corynebacterium glutamicum* R. *Appl Environ Microbiol.* **2007**, 73, 2349-53.
46. Agyei-Aye K, Chian MX, Lauterbach JH, Moldoveanu SC. The role of the anion in the reaction of reducing sugars with ammonium salts. **2002**, *Carbohydrate Research*, 337, 2273-2277.
47. Kurakake. M.; Kisaka, W.; Ouchi, K.; Komaki, T. Pretreatment with ammonia water for enzymatic hydrolysis of corn husk, bagasse and switchgrass. *Appl. Biochem. Biotech.* **1999**, 90, 251-259.
48. Sewalt, V. J. H.; Fontenot, J. P.; Allen, V.G.; Glasser, W.G. Fiber composition and in Vitro digestibility of corn stover fractions in response to ammonia treatment. *J. Agric. Food Chem.* **1996**, 44, 3136-3142.

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
49. Saulnier, L.; Thibault, J. F. Ferulic acid and diferulic acids as components of sugar-beet pectins and maize bran heteroxylans. *J. Sci. Food Agric.* **1999**, *79*, 396–402.
50. Chou, Y. C. T. Supercritical ammonia pretreatment of lignocellulosic materials. *Biotechnol. Bioeng. Symp.* **1986**, *17*, 19-32.
51. Weimer, P.J.; Chou, Y. C. T. Weston. W.M., Chase, D.B. Effect of supercritical ammonia on the physical and chemical structure of ground wood. *Biotechnol. Bioeng. Symp.* **1986**, *17*, 5-18
52. March, J. Advanced organic chemistry reactions mechanisms and structure, 3rd ed. Wiley, New York, **1985**, p1346.
53. Caidian Luo, David L. Brink and Harvey W. Blanch, Identification of potential fermentation inhibitors in conversion of hybrid poplar hydrolyzate to ethanol. *Biomass and Bioenergy* **2002**, *22*, 125-138.
54. Oliva, J. M.; Saez, F.; Ballesteros, I.; Gonzalez, A.; Negro, M. J.; Manzanares, P.; Ballesteros, M. Effect of lignocellulosic degradation compounds from steam explosion pretreatment on ethanol fermentation by thermotolerant yeast *Kluyveromyces marxianus*. *Appl. Biochem. Biotechnol.* **2003**, *105 -108*, 141-53.

Figure Captions:

Figure 1a: Enzymatic hydrolysis for AFEX treated poplar under varying conditions. Here, I) Glucan conversions for poplar as a function of AFEX conditions. All experiments were performed at 1:1 ammonia to biomass ratio and 30 minutes of residence time. In A) the effect of temperature was studied, fixing the moisture content at 233% (dwb). In B) the effect of moisture content on glucose conversions was studied using 180°C as a fixed temperature. All experiments used enzyme loadings of 31.3 mg of cellulase protein and 33.3 mg of β -glucosidase protein per gram of glucan. II.) Glucan conversions for poplar as a function of varying AFEX conditions and enzyme loadings. Here, experiments were done using two different enzyme loadings; (A) low enzyme loading (31.3 mg of cellulase protein, 33.3 mg of β -glucosidase protein per gram of glucan) and (B) high enzyme loading (125 mg of cellulase protein and 33.3 mg of β -glucosidase protein per gram of glucan). In some of experiments either 31.3 mg (low) or 125 mg (high) per gram of glucan of xylanase enzyme was also supplemented. The AFEX pretreatment conditions are at a fixed temperature (180 °C) and moisture (233%, dwb) and varying ammonia to biomass loadings (1:1, 2:1 and 3:1, w/w). All the hydrolysis experiments were done in duplicates.

Figure 2: Comparison of glucan (in dark bars) and xylan (in white bars) conversions for untreated, AFEX treated and washed AFEX treated samples prior to hydrolysis for both corn stover and poplar, respectively. Pretreatment temperatures during the AFEX process

1
2
3 are shown in the brackets. AFEX pretreatment was done using 1:1 ammonia to biomass
4 loading for both the corn stover and poplar, with 60% moisture (corn stover) and 233%
5 moisture (poplar) (% dry weight basis of substrate). Enzymatic hydrolysis was done
6 using 31 mg/g of glucan of cellulase and 33 mg/g of glucan of β -glucosidase for both
7 corn stover and poplar at 50 °C, for 168 h. All the hydrolysis experiments were done in
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

duplicates.

Figure 3: FT-IR ATR spectra for untreated/AFEX treated corn stover and poplar, respectively. Here, (i) Untreated corn stover (CS-UT), (ii) AFEX treated corn stover (CS-AFEX), (iii) Untreated poplar (Poplar-UT) and (iv) AFEX treated poplar (Poplar-AFEX). Stretching frequencies at 1664 cm^{-1} and 1740 cm^{-1} correspond to amide and ester linkages, respectively, as denoted by corresponding dotted lines.

Figure 4: Mass balance for AFEX treated poplar during pretreatment and enzymatic hydrolysis after 168h is shown. AFEX was performed at 180 °C and 700 psi for 30 minutes, using 2:1 ammonia to biomass ratio and 233% of moisture (dry biomass basis). Here, (A) unwashed poplar and (B) washed poplar after pretreatment and before enzymatic hydrolysis are represented. Enzymatic hydrolysis was done using 31.3 or 125 of cellulase protein and 33 mg of β -glucosidase protein per gram of glucan were used (results for the latter case of higher enzyme loading are shown in brackets for poplar). Multifect xylanase was supplemented using 31 or 125 mg protein/g of glucan.

1
2
3 **Figure 5:** Mass balance for AFEX treated corn stover during pretreatment and enzymatic
4 hydrolysis after 168h is shown. AFEX was performed at 90°C for 5 minutes, using 1:1
5 ammonia to biomass ratio and 60% of moisture (dry biomass basis). Here, (A) unwashed
6 corn stover and (B) washed corn stover after pretreatment and before enzymatic
7 hydrolysis are represented. Enzymatic hydrolysis was done using 31.3 mg of cellulase
8 protein and 33.3 mg of β -glucosidase protein per gram of glucan. Multifect xylanase was
9 supplemented using 3.1mg /g of glucan in all experiments.
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Figure 1

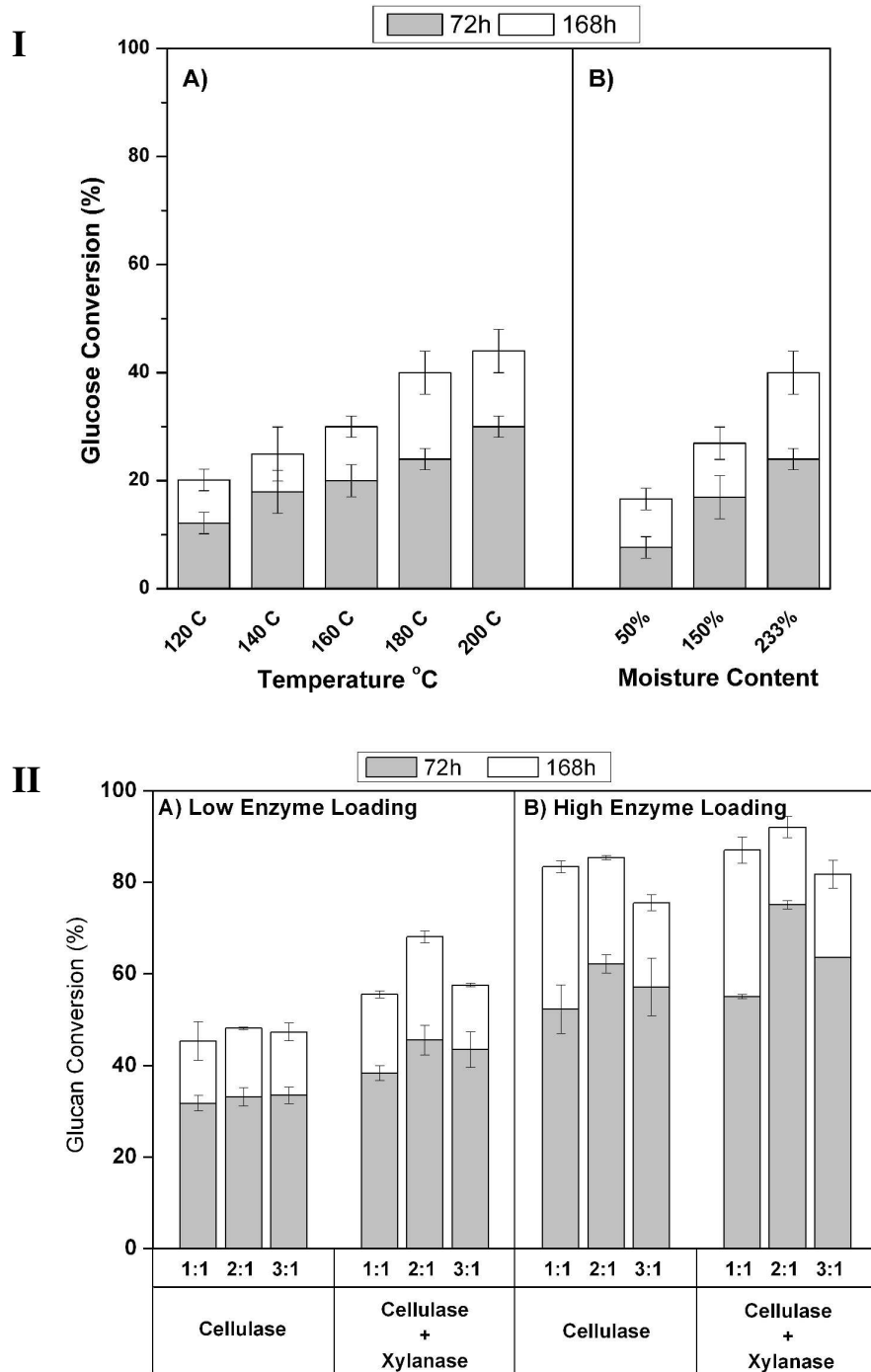
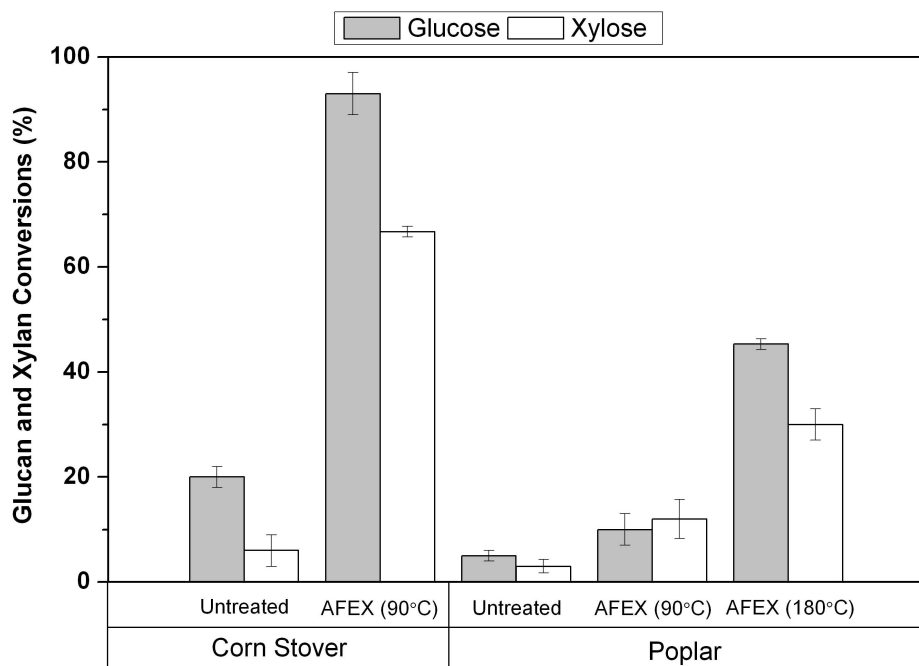


Figure 2



ew

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Figure 3

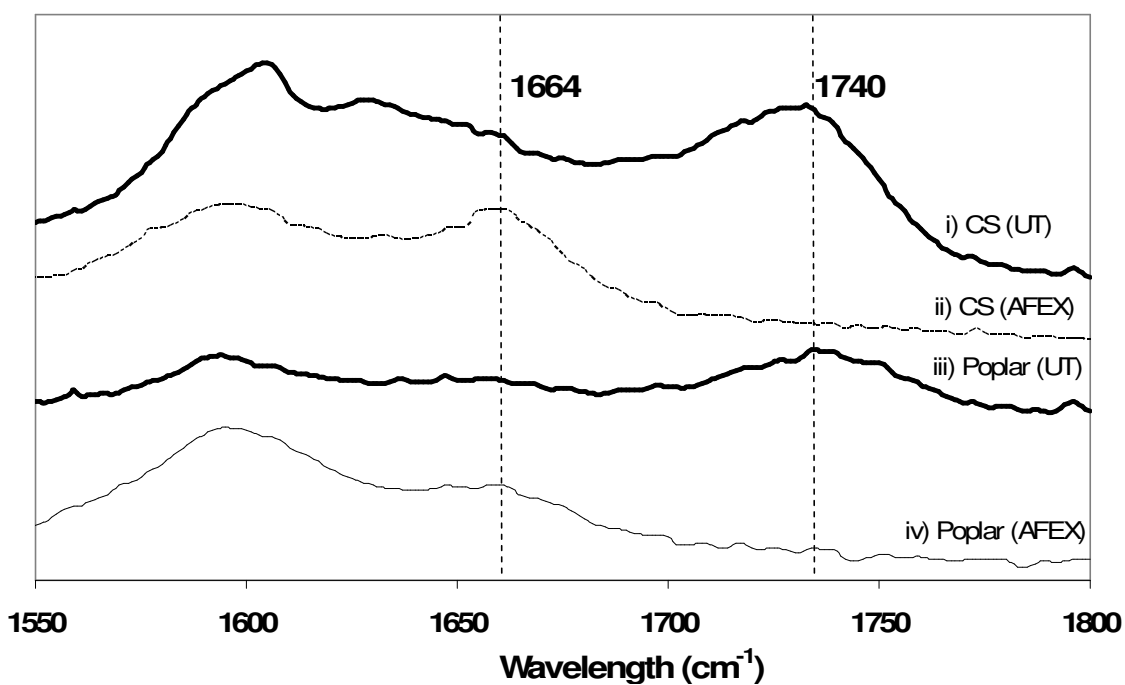


Figure 4

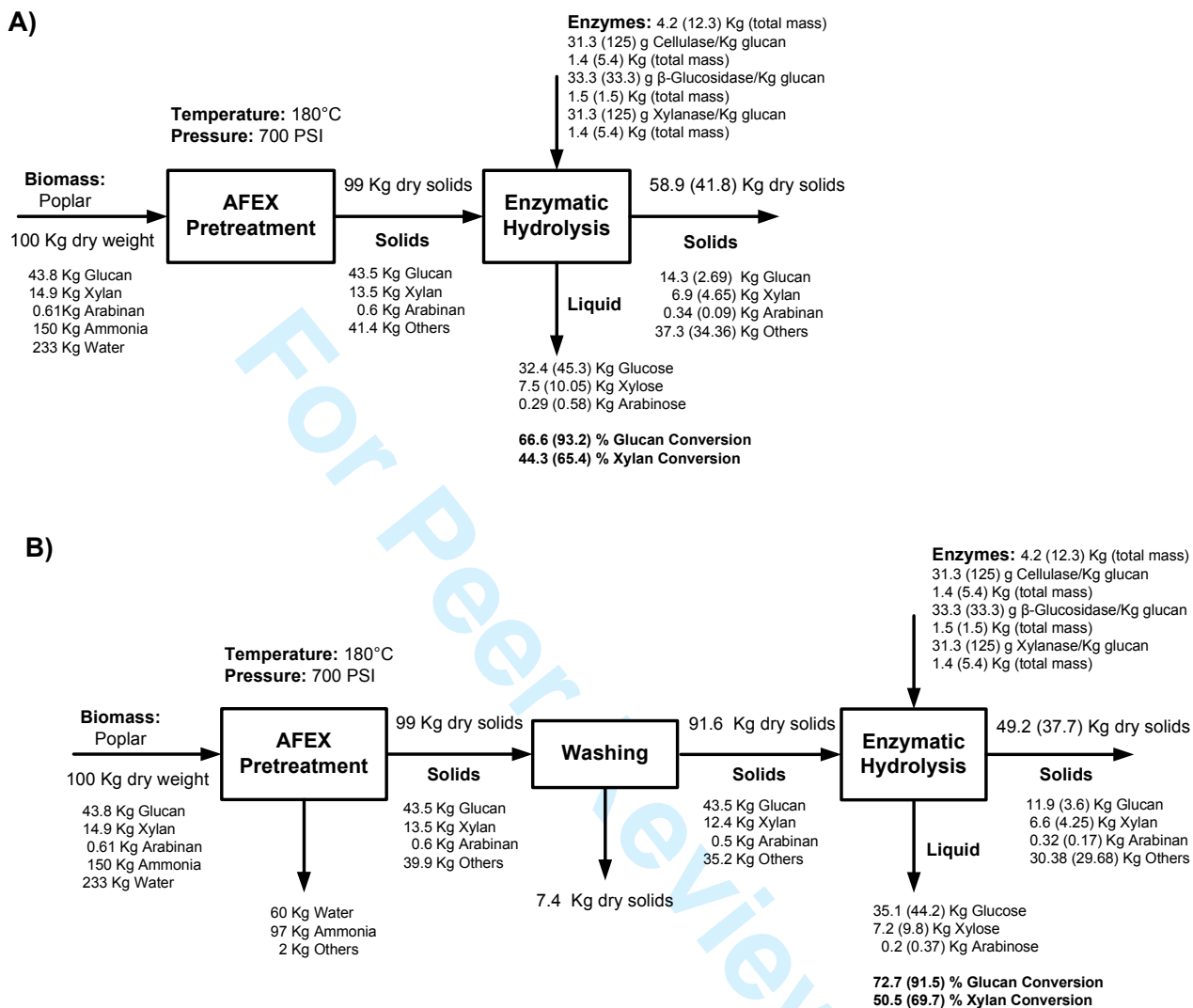


Figure 5

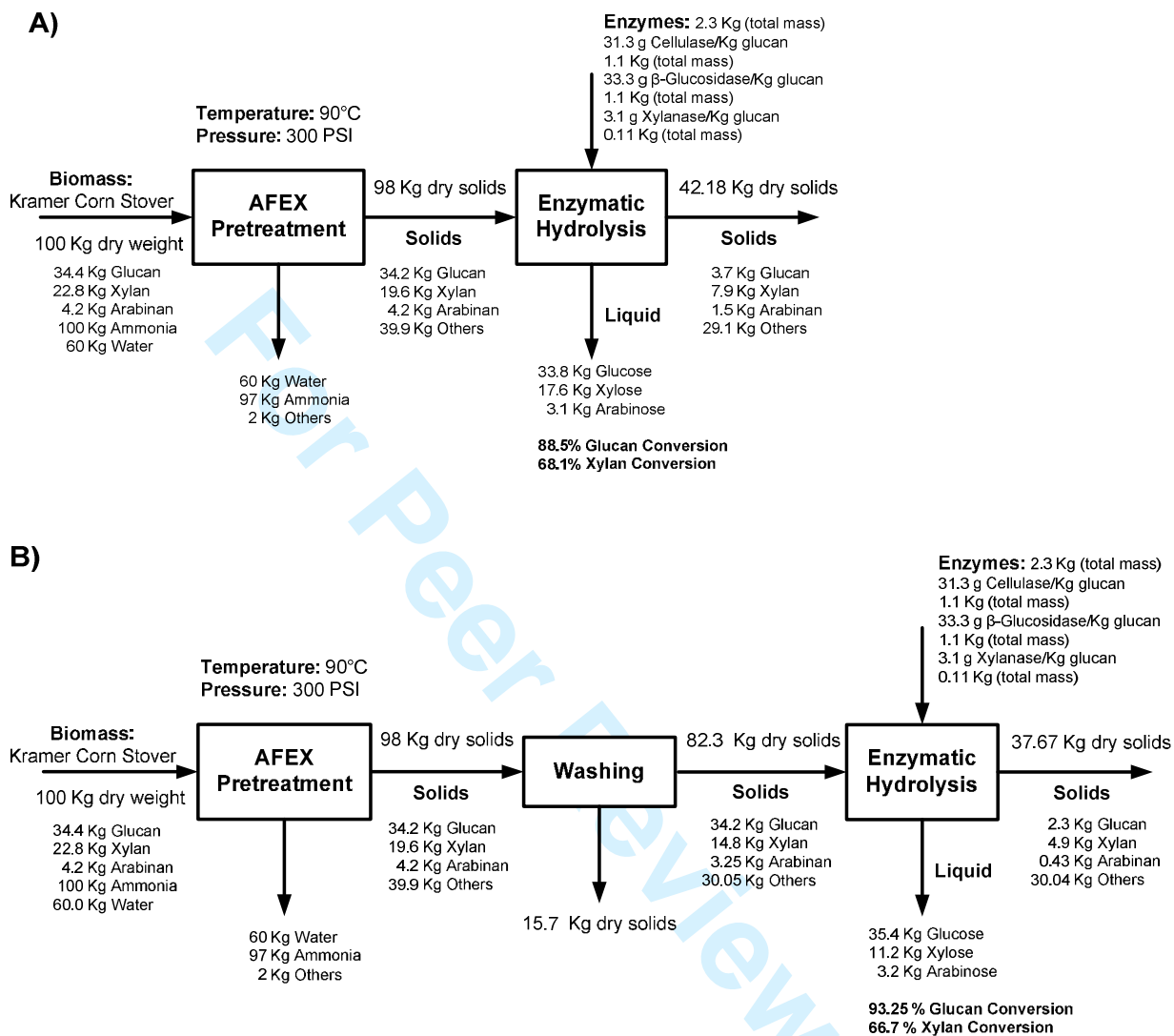


Table 1: Compositional analysis (% wt/wt, dry basis) for Kramer Corn Stover and High-lignin Poplar

Contents	Corn Stover	Poplar
Glucan	34.4	43.8
Xylan	22.8	14.9
Arabinan	4.2	0.61
Mannan	0.6	3.9
Galactan	1.4	1.0
Lignin	11.0	29.1
Protein	2.3	*Nd
Acetyl	5.6	3.6
Ash	6.1	1.1
**Extractives	8.5	3.6

* Nd- Not determined; ** Water extractives

Table 2: Water extractable carbohydrates produced during AFEX pretreatment of corn stover and poplar (mg per gm of original dry biomass). All experiments were done in triplicates.

	Glucose Average (mg/gm of dry biomass)	Xylose	Arabinose	Glucose Stdev (mg/gm of dry biomass)	Xylose	Arabinose
Poplar						
Untreated	0.17	0.14	0.12	0.09	0.17	0.10
Untreated (after acid hydrolysis)	2.63	0.24	0.16	0.39	0.13	0.10
AFEX treated	0.19	0.21	0.06	0.11	0.09	0.05
AFEX treated (after acid hydrolysis)	2.00	11.36	0.49	1.06	1.82	0.12
Corn stover						
Untreated	26.00	0	0.12	2.50	0	0.07
Untreated (after acid hydrolysis)	27.86	1.22	0.39	3.10	0.10	0.10
AFEX treated	6.57	0.15	0.12	2.66	0.21	0
AFEX treated (after acid hydrolysis)	17.91	48.98	9.53	1.40	12.30	2.88

Table 3: Small organics present in ASE water extracts of untreated (UT) and AFEX treated poplar (μg per gm of dry weight of substrate).

Analytes	Molecular formula	Poplar (UT) $\mu\text{g/g}$	Poplar (AFEX) $\mu\text{g/g}$	Fold Increase
Aliphatic Acids				
Malonic acid	$\text{CH}_2(\text{COOH})_2$	23.2	11.4	--
Lactic acid	$\text{CH}_3\text{CH}(\text{OH})\text{COOH}$	27.6	1411.9	51
Maleic acid	$\text{HOOC}-\text{CH}=\text{CH}-\text{COOH}$ (Cis)	0.6	4.4	7
<i>Cis</i> -aconitic acid	$\text{HOOC}-(\text{CH}_2-\text{COOH})\text{C}=\text{CH}-\text{COOH}$	0.9	1.3	1
Methylmalonic acid	$\text{HOOC}-\text{CH}(\text{CH}_3)-\text{COOH}$	0.4	74.0	74
Succinic acid	$\text{HOOC}-\text{CH}_2-\text{CH}_2-\text{COOH}$	2.0	196.0	97
Fumaric acid	$\text{HOOC}-\text{CH}=\text{CH}-\text{COOH}$ (Trans)	0.5	8.2	8
<i>Trans</i> -aconitic acid	$\text{HOOC}-(\text{CH}_2-\text{COOH})\text{C}=\text{CH}-\text{COOH}$	BDL	BDL	--
Levulinic acid	$\text{CH}_3\text{COCH}_2\text{CH}_2\text{COOH}$	BDL	53.7	54
Glutaric acid	$\text{HOOC}(\text{CH}_2)_3\text{COOH}$	30.4	21.4	1
Itaconic acid	$\text{CH}_2=\text{C}(\text{COOH})\text{CH}_2\text{COOH}$	BDL	1.0	1
2-Hydroxy-2-methylbutyric acid	$\text{C}_2\text{H}_5\text{C}(\text{OCH}_3)(\text{OH})\text{COOH}$	0.3	0.1	--
Adipic acid	$\text{HOOC}(\text{CH}_2)_4\text{COOH}$	0.5	7.2	15
Furans				
2-Furoic acid	$\text{C}_5\text{H}_4\text{O}_3$	0.3	8.6	25
Aromatic acids				
Gallic acid	$\text{C}_6\text{H}_2(\text{OH})_3\text{COOH}$	BDL	0.2	0.2
3,4-Dihydroxybenzoic acid	$\text{C}_6\text{H}_3(\text{OH})_2\text{COOH}$	2.5	4.5	2
3,4-Dihydroxybenzaldehyde	$\text{C}_6\text{H}_3(\text{OH})_2\text{CHO}$	0.5	15.1	15
Salicylic acid	$\text{OHC}_6\text{H}_4\text{COOH}$	56.0	115.7	2
4-Hydroxybenzaldehyde	$\text{C}_6\text{H}_4(\text{OH})\text{CHO}$	0.7	60.2	90
Vanillic acid	$\text{OHC}_6\text{H}_3\text{OCH}_3\text{COOH}$	0.8	31.3	39
Homovanillic acid	$\text{OHC}_6\text{H}_3\text{OCH}_3\text{CH}_2\text{COOH}$	BDL	25.6	26
4-Hydroxyacetophenone	$\text{OHC}_6\text{H}_4\text{COCH}_3$	0.1	10.6	11
Caffeic acid	$\text{C}_6\text{H}_3(\text{OH})_3\text{CH}=\text{CHCOOH}$	0.2	0.4	2
Syringic acid	$\text{OHC}_6\text{H}_3(\text{OCH}_3)_2\text{COOH}$	2.4	71.9	30
Vanillin	$\text{OHC}_6\text{H}_3\text{OCH}_3\text{CHO}$	4.9	429.2	88
4-Hydroxybenzoic acid	$\text{C}_6\text{H}_4(\text{OH})\text{COOH}$	2.0	11.2	11
Benzoic acid	$\text{C}_6\text{H}_5\text{COOH}$	4.1	304.7	305
Syringaldehyde	$\text{OHC}_6\text{H}_3(\text{OCH}_3)\text{CHO}$	6.0	949.3	159
4-Hydroxycoumaric acid	$\text{OHC}_6\text{H}_4\text{CH}=\text{CH}-\text{COOH}$	1.8	4.6	3
Ferulic acid	$\text{OHC}_6\text{H}_3(\text{OCH}_3)\text{CH}=\text{CHCOOH}$	4.7	5.4	1.1
Sinapic acid	$\text{OHC}_6\text{H}_2(\text{OCH}_3)_2\text{CH}=\text{CHCOOH}$	0.2	0.9	5
<i>Para</i> -toluic acid	$\text{C}_6\text{H}_4(\text{CH}_3)\text{COOH}$	9.3	9.9	1.1

BDL = below detection limit